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# Stability indicating UPLC-MS/MS method for quantification and identification of cefepime and its degradants in API

# Jabeen<sup>1\*</sup> (D), Bangalore Venkatappa Suma<sup>2</sup> (D)

<sup>1</sup>Research Scholar, Department of Pharmaceutical Chemistry, Faculty of Pharmacy, M.S. Ramaiah University of Applied Sciences, Bengaluru, India. <sup>2</sup>Department of Pharmaceutical Chemistry, Faculty of Pharmacy, M.S. Ramaiah University of Applied Sciences, Bengaluru, India.

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## ABSTRACT

An analytical method was developed and validated for determining stability studies of the drug cefepime (CFP) by Ultra Performance Liquid Chromatography - Tandem Mass Spectrometer (UPLC-MS/MS). In this research, Ultra Performance Liquid Chromatography (UPLC) C18 ethylene hybrid column was used. About 0.2% formic acid in the water and acetonitrile (ACN), in 20:80 v/v ratios, was used as the mobile phase with a 0.15 ml/minute flow rate. In multiple reaction monitoring modes, positive electrospray ionization (ESI) was used. The chromatogram of standard Cefepime (CFP) was found to have a retention time of  $0.82 \pm 0.02$  minute. The degradation of CFP was tested under different stress conditions and the method was validated to meet the International Conference on Harmonization of Technical Requirement for Registration of Pharmaceuticals for Human Use guidelines. The retention time of the forced degraded solutions of CFP was found to be 0.76, 0.83, 0.81, and 0.82 minutes for forced acidic, oxidation, thermal, and neutral conditions, respectively. The MS/MS spectra for CFP at 179.15 and 241.33  $\pm$  0.06 in different stress conditions (except basic) with retention times at 0.90  $\pm$  0.02 and 0.84  $\pm$  0.10 minutes indicate Cefepime degradant 1 (CD1) and Cefepime degradant 2 (CD2), respectively. The MS/MS spectra for CFP basic degradatint (CD1) in the basic medium were obtained at an m/z ratio of 179.15. The research findings conclude that CFP was unstable with partial degradations in all conditions, and complete degradation in basic medium.

# INTRODUCTION

Pharmaceuticals tend to deteriorate to degrade with time. The degradation process is a decrease in distinct characteristics over time. A drug's shelf life is estimated by analyzing two independent properties after it has been stressed to degrade. First, determine how much active ingredient is lost. Second, identify the degradation products that result from the aging process. Degradants need to be predicted and assessed at concentrations greater than 0.1% of the parent molecule in this test, which can be challenging. Trace analysis is required for this part of the assay, which demands the establishment of a stability-indicating assay method (SIAM) for the drug and its dosage forms. Physical, chemical, microbiological,

and environmental degradation determine the quality and purity of a drug product. Over time, an active pharmaceutical ingredient (API) or excipients may transit from a metastable state to a more thermodynamically stable state. A drug molecule can undergo chemical degradation due to several reactions such as oxidation, isomerization, hydrolysis, racemization, elimination, and UV degradation. In addition, there may be interactions with excipients and other pharmaceuticals (Connors et al., 1986). When water interacts with parental formulations, it leads to hydrolytic breakdown (Waterman et al., 2002). The most frequent cause of pharmaceutical degradation is oxidative degradation, which can take one of three paths: i) peroxide-accelerated, which involves exposing the drug to dilute peroxide solutions; ii) photochemically stimulated; and iii) radically simulated, which involves reversal loss of electrons (Bajaj et al., 2012; Cosa, 2004; Deidda et al., 2018; Quadri et al., 2014; Sahu et al., 2018).

Cefepime (CFP) was developed in 1994 as a fourthgeneration cephalosporin. Microorganisms such as *Pseudomonas* 

<sup>\*</sup>Corresponding Author

Jabeen, MS Ramaiah University of Applied Sciences, Bangalore, India. E-mail: syed.jabeen2 @ gmail.com

Figure 1. Structure of Cefepime.

aeruginosa, Enterobacteriaceae, Staphylococcus aureus, and multidrug-resistant Streptococcus pneumoniae are susceptible to CFP. The chemical name of CFP (Fig. 1) is (6R,7R)-7-{[(2Z)-2-(2-amino-1,3-thiazol-4-yl)-2-(methoxyimino)acetyl] amino}-3-[(1-methylpyrrolidinium-1-yl) methyl]-8-oxo-5-thia-1azabicyclo[4.2.0]oct-2-ene-2-carboxylate. It is a semi-synthetic drug having an aminothiazolyl group and a Z-methoxyimine moiety at C-7, expanding its spectrum and enhancing its β-lactamase stability. At the C-3 position, the presence of the quaternary N-methylpyrrolidine group aids in Gram-negative bacteria penetration. It works by preventing cell wall formation by inhibiting protein synthesis. It is used to treat urinary tract infections, septicemia, staphylococcal infections, bronchitis, intra-abdominal infections, skin and its structures infections, pneumonia infections, and in febrile neutropenic patients (Glish and Burinsky, 2008; Lemke et al., 2002). CFP stability was estimated using several methods using chromatography (Abd El Aziz Shama et al., 2021; Al Kamaly, 2022; Bjergum et al., 2021; De Borba et al., 2008, Dos Anjos et al., 2022; El-Beltagy et al., 2019; El-Dars et al., 2019; Fage et al., 2021; Isla et al., 2005; Jagadeesh Kumar et al., 2010; Jiang et al., 2010; Jiménez Palacios et al., 2005; Kalyani et al., 2018; Kommana et al., 2014; Liu et al., 2018; Mameli et al., 2019; Moorthy et al., 2020; Nemutlu et al., 2009, Ocaña González et al., 2004; Ohmori et al., 2011; Patil et al., 2018; Rehm and Rentsch, 2020; Rodrigues et al., 2016; Seraisso et al., 2022; Shrestha et al., 2014; Siddiqui et al., 2010; Sun et al., 2022; Sundara Raj et al., 2013; Sunitha et al., 2013; Van Vooren and Verstraete, 2021; Zander et al., 2015) but these methods had some disadvantages as complex mobile phase composition, increased matrix effect, decreased sensitivity, and inability to identify degradants using trace analysis in addition to fragmentation (Niessen, 2010; Niessen and Ricardo, 2017), less environmentally friendly, and time-consuming. However, there have been no previous analytical procedures of trace analysis reported on CFP stability studies and its degradants using ultraperformance liquid chromatography-mass spectrometry (UPLC-MS)/MS (Jabeen and Suma, 2022). The aim was to develop an eco-friendly, selective, sensitive, and accurate method for the analysis of the SIAM for CFP by using UPLC-ESI-QQQ-MS.

# MATERIALS AND METHODS

## **Chemicals and reagents**

CFP standard (99.8%) was acquired as samples from Kshetra Analyticals. The liquid chromatography grade water,

formic acid, dimethyl sulfoxide (DMSO), acetonitrile (ACN), methanol, and all other reagents were procured from SD fine chemicals limited (Hyderabad, India).

#### Instrumentation and chromatographic conditions

The chromatographic separations were performed on a UPLC system (Acquity C18 BEH,  $50 \times 2.1$  mm, 1.7 m) with MS/ MS (UPLC-MS/MS—All Waters, USA). Waters Quattro premier mass spectrophotometer was utilized in tandem MS/MS using a triple quadrupole mass analyzer with an electrospray ionization (+ESI) probe. The system data were collected using the masslynx version 4.1 software. The isocratic elution with a mobile phase composed of 20:80 v/v—0.2% formic acid in water and ACN was used. A flow rate of 0.15 ml/minute, with an injection volume of 5 µl, and a run time of 3 minutes were maintained. The column temperature was maintained at 30°C.

#### Preparations of standard stock solution

#### Standard stock solutions

For CFP, a 1,000  $\mu$ g/ml standard stock solution was prepared using DMSO solvent (Stock Solution A). All of the dilutions of standard stock solution A were performed with 50% methanol (diluent solvent).

#### Standard working solution

To obtain stock solution B of 200  $\mu$ g/ml, 2 ml of CFP stock solution A was diluted to 10 ml using a diluent (50% methanol) solvent. To make stock solution C, 1 ml of stock solution B was diluted to 10 ml with a diluent solvent to yield a concentration of 20  $\mu$ g/ml. The entire standard solution was properly sonicated and filtered using 0.45  $\mu$ m membrane filters before the analysis. Aluminum foil was used to protect stock solution C from light and it was kept in the refrigerator. For stability testing, the normal stock solution C was employed.

## Preparation of the mobile phase

A mixture of 0.2% formic acid in water and ACN 20:80 v/v was used as the mobile phase.

## Validation procedure

According to the International Conference on Harmonization of Technical Requirement for Registration of Pharmaceuticals for Human Use (ICH) guidelines, the proposed UPLC-MS/MS methodology for CFP was validated (ICH, 2003).

#### Forced degradation studies of CFP

#### Standard or control

About 0.5 ml of 20  $\mu$ g/ml CFP solution was diluted with 4.5 ml of diluent to make a standard working solution (2,000 ng/ ml), which was then injected into the UPLC-MS/MS.

## Stress conditions

Forced degradation studies were studied per ICH Q1 guidelines using different stress conditions such as acidic, basic, neutral, peroxide, and thermal degradation. The solutions were analyzed regularly at 6, 12, and 24 hours for a week using UPLC-MS/MS. Hydrolysis of CFP in the presence of water was checked because it is reconstituted using water for injection for intravenous or intramuscular purposes. Degradation of the drug



under stress conditions is confirmed by the reduction in area in the chromatogram and drug quantity.

## **RESULTS AND DISCUSSION**

#### Method development and validation

Using ESI in positive ion mode MS/MS detector optimum settings was investigated to acquire transition for each molecule. Full scan spectra were recorded to find the most prevalent m/z value to optimize the cone voltage with the precursor ions as  $[M+H]^+$  ions. The most frequent ions were found to have the most sensitive transition using collision energies. MS/MS transition for quantification and confirmation for CFP was observed at 481.3  $\rightarrow$  166.85. Typical optimized operating conditions were at capillary voltage: 3.00 and 2.94 kV, cone voltage: 25 and 28 V, desolvation temperature: 400°C and 388°C, source temperature: 120°C, desolvation gas flow: 850 and 848 l/hour, cone gas flow: 100 and 97 l/hour, ion energy: 10.5, multiplier: 550–547 V, syringe pump flow: 100 µl/minutes, and collision gas flow: 0.15 and 0.15 ml/minutes. The chromatogram of standard CFP (for n = 6) was found to have a retention time of  $0.82 \pm 0.02$  minutes. The MS/ MS spectrum of  $[M+H]^+$  of CFP is at m/z of 481.35 with a peak area of 57,736.916 (Fig. 2). The results for CFP validation are summarized in Table 1.

## Forced degradation studies

CFP has degraded in all the stress conditions. A typical chromatogram of CFP and its separated degradants is shown in Figure 3. In acidic, neutral, oxidation, and thermal stress conditions, the presence of m/z 481.42  $\pm$  0.07 [M+H]<sup>+</sup> peak indicates partial degradation of CFP into CFP degradant 1 (CD1) and CFP degradant 2 (CD2). The [M+H]<sup>+</sup> peak at m/z 179.15 and 241.33  $\pm$  0.03 indicate the presence of CD1 and CD2 respectively. In the basic medium, the absence of m/z 481.42  $\pm$  0.07 [M+H]<sup>+</sup> peak indicates complete degradation of CFP, and the presence of [M+H]<sup>+</sup> peak at m/z 179.15 indicates that CFP has completely degraded into CD1. The MS/MS spectra of CD1 and CD2 are given in Figures 4 and 5 respectively. A summary of the SIAM UPLC-MS/MS method for CFP has been given below in Table 2.



Figure 2. Cefepime (A) MS/MS spectra. (B) Standard chromatogram.

Parameter	Values obtained for cefepime Standard value (ICH guideline)			
	0.90 (for area)	% RSD: less than 2		
System suitability (% RSD)	0.90 (for concentration)			
	Tailing factor: 1.03	Tailing factor: less than 2		
Specificity	Drug specific with no interference	Specific to drug		
Linearity	$r^2 = 0.99$ , linear curve.	$r^2 \ge 0.99$ , similar response ratio		
Range (ng/ml)	200–3000	Concentration where drug can be reliably detected		
Accuracy (% recovery)	99.5%-100.8%	% Recovery for each level should be between 98.0% and 102.0%.		
	Repeatability = 0.91			
Precision (% RSD)	Intraday = 1.34	% RSD: Less than 2		
	Interday $= 1.72$			
LOD	66 ng/ml	>2 times the baseline		
LOQ	200 ng/ml	S/N ratio: 10:1		
	Flow rate:0.5			
Robustness (% RSD)	Temperature:0.29	% RSD: Less than 2		
	Rt: 0.89			

Table 1. Summary of validation parameters for the proposed method of cefepime.

## Cefepime



Figure 3. Typical chromatogram of cefepime and its degradants.



Figure 4. MS/MS spectra of degradants CD1.



Figure 5. MS/MS spectra of degradants CD2.

# DISCUSSION

In this research, the retention time of the parent compound of CFP is 0.82 minute with MS/MS spectra of standard CFP obtained at an m/z ratio of 481.42. The retention time of degradant CD1 in different condition was 0.91  $\pm$ 0.01 minute with MS/MS spectra of CD1 at m/z at 179.15. The retention time of degradant CD2 in different condition was 0.84  $\pm$  0.1 minute with

MS/MS spectra of CD2 at m/z at 241.33. The presence of these m/z spectra at 481.42  $\pm$  0.09 in all stress conditions, except basic indicates that the CFP is unstable and has partially deteriorated. The peak in the chromatogram and MS/MS spectra of CFP in forced basic medium has been observed at m/z of 179.15 and Retention time at 0.91  $\pm$  0.01 minute of degradant CD1 indicating CFP is more vulnerable to a basic condition resulting in complete degradation of the drug in basic medium. During the

	CEF and its degradants under different stress conditions						
Forced degradation — condition —	Cefepime		CD1		CD2		Percentage of cefepime
	Peak area	Rt (minute)	Peak area	Rt (minute)	Peak area	Rt (minute)	
Acidic	5,561.47	0.76	459.58	0.92	29,231.25	0.81	84.26
Basic	-	-	16,626.66	0.90	-	-	Completely degraded
Neutral	15,222.305	0.82	240.93	0.91	21,633.82	0.84	56.91
Oxidation	941.85	0.83	4,214.67	0.91	30,307.886	0.99	97.34
Thermal	15,332.58	0.81	353.06	0.90	16,572.46	0.83	56.6

Table 2. Forced degradation studies results.

CD1: Cefepime degradant 1, CD2: Cefepime degradant 2.



Figure 6. Fragmentation pathway of cefepime.



Table 3. Degradation products and their proposed structures.

CD1: Cefepime degradant 1, CD2: Cefepime degradant 2.

entire analysis of CFP, there was no interference observed in the solution. The fragmentation of standard CFP into daughter ion was observed at transition pair of  $481.35 \rightarrow 166.85$ . During the forced degradation studies, CFP and its stability solutions were analyzed by UPLC coupled to electrospray triple quadrupole MS (LC-ESI-QQQ-MS) to identify the possible degraded products under stress conditions. The general fragmentation pattern of CFP in UPLC-MS/MS represents the molecular weight and possible structures of fragments for CFP during long-term stability studies as shown in Figure 6. The molecular weight of degraded products with possible fragments was predicted based on the protonated molecular ions [M+H]<sup>+</sup> and the fragmentation pathway of CFP was established from mass precision and fragmentation mass spectrum. All degradation products and possible fragmentations were characterized and the proposed structures are presented in Table 3.

## CONCLUSION

A novel stability-indicating UPLC-MS/MS method for the determination of stability in different stress conditions for CFP has been performed. Rt of the forced degradation studies solution was reported to be 0.82 minute for CFP, CD1 at 0.91  $\pm$ 0.01 minute, and CD2 at 0.84  $\pm$  0.1 minute; this method has less Rt when compared to available assay methods found in the literature. Forced degradation studies were successfully developed as per ICH guidelines.

# FUTURE PLAN OF WORK

The developed method can be applied to determine degradants of CFP in blood, cerebrospinal fluid, and urine samples by using this automated UPLC-MS/MS technique.

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# **AUTHOR CONTRIBUTIONS**

All authors made substantial contributions to the conception and design, acquisition of data, or analysis and interpretation of data; took part in drafting the article or revising it critically for important intellectual content; agreed to submit to the current journal; gave final approval of the version to be published; and agree to be accountable for all aspects of the work. All the authors are eligible to be an author as per the international committee of medical journal editors (ICMJE) requirements/ guidelines.

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## **CONFLICTS OF INTEREST**

The authors report no conflicts of interest.

#### ETHICAL APPROVAL

This study does not involve experiments on animals or human subjects.

#### DATA AVAILABILITY

All data generated and analyzed are included in this research article.

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