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Colon-targeted protein delivery by using solid lipid nanoparticles

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ABSTRACT

Recently, therapeutic proteins have been used to combat life-threatening diseases. To date, oral routes have been developed to deliver proteins. Since the proteolytic degradation in the upper gastrointestinal tract frequently occurs, we need a formulation and strategy to protect and deliver protein to the colon. In this research, solid lipid nanoparticles (SLNs) were prepared in eight formulae for colon-targeted delivery with various glyceryl monostearate, Tween 80, soy lecithin, and polyethylene glycol 6000 concentrations. Bovine serum albumin (BSA) was used for the protein model in the system. All formulae were characterized by their morphology, particle size, polydispersity index (PDI), zeta potential, and entrapment efficiency. The distribution of particle sizes varied between 94.24 and 186.8 nm and the zeta potential value ranged from -22.8 to -34.3 mV. Formula 4 (F4) showed the best entrapment efficiency of 78.69% with a particle size of 102.1 nm, PDI of 0.349, and zeta potential of -26.1 mV. F4 was then coated with Eudragit S100 and measured for its drug release profile *in vitro*. F4 coated with Eudragit S100 held the drug released in the gastric and released all the BSA in the colon condition. These results indicated that BSA–SLN F4 could be a promising delivery system to obtain optimal colon-targeted parameters.

INTRODUCTION

Due to recent advances in pharmaceutical biotechnology, therapeutic proteins have been used to combat life-threatening diseases (Rehman *et al.*, 2016). Since 2011, the US Food and Drug Agency has developed and approved 62 recombinant proteins to treat several clinical indications, such as cancers, inflammation, exposure to infectious agents, and genetic disorders (Lagassé *et al.*, 2017). These therapeutic proteins are delivered mainly via the parenteral route. The parenteral route has some disadvantages, such as invasiveness, hypersensitivity reaction for some people, and higher cost than other routes. Furthermore, the short plasma half-life and high elimination rate of protein make a repeated injection needed to obtain therapy effect, so nonparenteral routes, such as the oral route, have been developed to deliver protein (Martins *et al.*, 2007).

The oral route to deliver protein has some barriers, especially the bioavailability of the protein. Proteins is are degraded

by proteolytic enzymes and pH of stomach and intestine. The limitation of protein to cross cell membranes and its probability of undergoing enzymatic and/or proteolytic degradation in the upper gastrointestinal (GI) tract can be solved by delivering proteins directly to the colon as the release and absorption target area (Akash et al., 2015; Martins et al., 2007). Less proteolytic activity in colon mucosa decreased CYP3A4 activity (Iswandana et al., 2018a). Reduced P-gp expression makes the colon suitable as a systemic absorption site for protein (Date et al., 2016; Rajpurohit et al., 2010). However, the colon's location at the distal end of the GI tract causes the protein to pass the wide ranges of the GI tract's pH, which can make the protein unstable and inactive (Lu et al., 2016). Besides that, various conditions and environments during delivery via the GI tract affect the formulation including the pH, enzyme, electrolyte, transit time, and pressure (Iswandana et al., 2017). Using a nanoparticle as a protein carrier can overcome the problem because a nanoparticle is stable in the GI's environment and can protect the encapsulated drug from extreme pH and enzyme degradation (Lu et al., 2016).

A colon-targeted drug delivery system can be based on pH changes, enzyme activity by colon bacteria, time system, and nanoparticles. The nanoparticles system approach makes the protein dissolve and become trapped in the nanoparticles matrix.

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In the past few decades, lipid-based drug delivery systems have gained attention due to their biocompatibility and lipophilicity. One of the lipid nanoparticles used for colon-targeted drug delivery systems is solid lipid nanoparticles (SLNs). An SLN delivery system allows for protein release, increasing protein stability, carrying large quantities of protein, and protecting the protein from the first pass effect. Furthermore, SLNs are considered nanoparticles with better cell tolerability with a size of less than 1,000 nm (Irianti *et al.*, 2020).

In this research, Bovine serum albumin (BSA) was used as a protein model incorporated in the SLN by the doubleemulsion method. SLNs were further coated with Eudragit S100, and the *in vitro* study was carried out to observe the amount of BSA that can reach the colon.

MATERIALS AND METHODS

Materials

BSA (Sigma-Aldrich, Saint Louis, MO), glyceryl monostearate (Chemical Point, Oberhaching, Germany), Tween 80 (Brataco, Jakarta, Indonesia), soy lecithin (Unitechem Co., Ltd., Suqian, China), inulin (BENEO-Orafti SA, Oreye, Belgium), polyethylene glycol (PEG) 6000 (NOF Corporation, Tokyo, Japan), Eudragit S100 (Evonik, Bekasi, Indonesia), Lowry reagent, trichloroacetic acid (Merck, Darmstadt, Germany), and Aquadest were acquired.

Colyophilization BSA with inulin

An inulin solution (5 mg/ml) was added to the BSA solution (2 mg/ml) and stirred homogeneously. The mixed solution was placed in a vial and frozen in a freezer at -80° C and then freeze-dried at 0.2 mBar.

Preparation of BSA-SLN

Table 1 shows the formula of BSA–SLN. The method of SLN preparation used in this study was double emulsion. Glyceryl monostearate was melted at 55°C. Then, soy lecithin, 20 ml of BSA solution (2 mg/ml), and PEG 6000 were added. The solution was stirred using a T25 digital ULTRA-TURRAX homogenizer (IKA, Kuala Lumpur, Malaysia) for 10 minutes at 10,000 rpm. The emulsion formed was added to the surfactant solution (Tween 80) and sonicated for 25 minutes.

BSA-SLN cCoating with Eudragit S100

An Eudragit S100 solution with a concentration of 5% w/v was prepared using deionized water. The optimum BSA–SLN

formula solution was then added to the Eudragit S100 solution with the same volume. The addition of SLN to the polymer solution was carried out drop by drop with continuous agitation at room temperature (20°C) for 10 minutes to produce the desired coating (Zhang *et al.*, 2014). The SLN coating results were immediately frozen at -85° C and lyophilized using a freeze-dryer at -65° C condenser temperature and 0.211 mBar pressure for 24 hours.

Characterizations of BSA-SLN

Particle Size Distribution

The measures of the mean intensity (Dp) and polydispersity index (PDI) were taken by photon correlation spectroscopy using Zetasizer Nano ZS (Malvern Panalytical, Malvern, UK). Measurements were made on the SLN dispersion in 1:1,000 Aquadest at 25°C with a scattering angle of 90°.

Zeta Potential

Zeta potential from the SLN was measured by electrophoretic light scattering using the Zetasizer Nano ZS (Malvern Panalytical, Malvern, UK). Zeta potential measurements were used to determine the physical stability of the SLN and the electric charge on the SLN's surface. Measurements were made on the SLN dispersion in a 1 :1,000 Aquadest at 25°C with a scattering angle of 90°.

Particle Morphology

The surface morphology of the SLN-Eudragit S100 was seen using a transmission electron microscope (FEI Tecnai G2 20 S-TWIN, Hillsboro, OR). SLN-Eudragit S100 powder was dispersed in distilled water. As many as one drop of the dispersion was dropped on the carbon-coated copper network to form a thin liquid layer (Elmowafy *et al.*, 2017). The layers on the grid were allowed to dry at room temperature. Next, the grid was attached to the instrument, and photos were taken at various magnifications with an EagleTM CCD Camera (Shah *et al.*, 2016).

Entrapment Efficiency of SLN

The entrapment efficiency of BSA in the SLN is determined indirectly by calculating the BSA that is not trapped. The final SLN preparation results were centrifuged at a speed of 12,000 rpm. Protein is calculated from the free BSA in the supernatant from the centrifugation result. 1 ml of supernatant was taken, and then the BSA that is not trapped was precipitated using a 10% TCA solution. The precipitate was then dissolved in 1 ml

Table 1. Formulae of SLN containing BSA.

| Formulae | BSA-inulin (%) | GMS (%) | Tween 80 (%) | Soy lecithin (%) | PEG 6000 (%) |
|----------|----------------|---------|--------------|------------------|--------------|
| F1 | 5.0 | 1.0 | 1.0 | 0.25 | 0.25 |
| F2 | 5.0 | 2.0 | 1.0 | 0.25 | 0.25 |
| F3 | 5.0 | 1.0 | 1.5 | 0.25 | 0.25 |
| F4 | 5.0 | 2.0 | 1.5 | 0.25 | 0.25 |
| F5 | 5.0 | 1.0 | 1.0 | 0.5 | 0.25 |
| F6 | 5.0 | 2.0 | 1.0 | 0.5 | 0.25 |
| F7 | 5.0 | 1.0 | 1.5 | 0.5 | 0.25 |
| F8 | 5.0 | 2.0 | 1.5 | 0.5 | 0.25 |

deionized water. The BSA in the solution was tested using the Lowry assay. The absorbance was measured at a wavelength of 750 nm using a microplate reader (VersaMax, Chattanooga, TN). The BSA concentration was calculated by comparing it against a standard curve of known standard BSA concentrations. The percentage of entrapment efficiency was calculated using the following formula:

Entrapment efficiency (%)

$$= \frac{\text{Total amount of drug loaded} - \text{free drug in supernatant}}{\text{Total amount of drug loaded}} \times 100$$

In Vitro Drug Release Studies

The *in vitro* release studies were carried out by a modified dissolution method using a dialysis membrane. Before use, dialysis bags with a molecular weight cut-off of 100 kDa were immersed in distilled water for 12 hours. Furthermore, 100 mg of optimal SLN formula was suspended in distilled water. One milliliter of the suspension was filled in a dialysis bag. Then, the dialysis bag was put in a beaker glass containing 250 ml of release media operated at 37°C using a magnetic stirrer at a speed of 100 rpm. The *in vitro* release study was carried out in hydrochloric acid media of 0.1 N pH 1.2, phosphate buffer pH 7.4, and phosphate buffer pH 6.8 that simulate the stomach, small intestine, and colon, respectively (Iswandana *et al.*, 2018b). The time of drug release in the hydrochloric acid media pH 7.4 for 3 hours, and in the phosphate buffer media pH 6.8 for 3 hours.

A total of 1 ml of the sample was taken at 15, 30, 45, 60, 90, 120, and 180 minutes of each medium and then replaced immediately with new media at the same volume to maintain sink conditions. The BSA content in the dissolution medium was analyzed using the Lowry protein assay. The absorbance was measured at a wavelength of 750 nm using a microplate reader (VersaMax, Chattanooga, TN). The BSA concentration was calculated using a standard calibration curve.

RESULTS AND DISCUSSION

Colyophilization BSA with inulin

Colyophilization is a process of protecting the BSA from conformational changes and denaturation during storage and passing through the GI tract. Colyophilization is carried out by coating the BSA protein with inulin. The ratio of the concentration of BSA and inulin is 1 :2.5. From the experimental results, we obtained 32 mg of BSA-inulin containing 10 mg of BSA. The addition of inulin to BSA by colyophilization can increase protein stability and decrease denaturation caused by conformational change. Inulin as a protein protectant can prevent the unfolding of proteins (Furlán *et al.*, 2010).

Preparation of BSA-SLN

The preparation of BSA–SLN uses the double-emulsion method. From the experimental results, it was found that the appearance of the SLN is a white-yellow, odorless, suspension-like solution, as shown in Figure 1.



Figure 1. BSA--SLN preparation from formulations 1-8.

| Formula | Particle size (nm) | iPDI | Zeta potential (mV) | Entrapment efficiency (%) |
|---------|--------------------|-------|---------------------|---------------------------|
| F1 | 112.5 | 0.248 | -28.5 | 59.65 |
| F2 | 114.1 | 0.423 | -22.8 | 76.29 |
| F3 | 100.4 | 0.312 | -23.5 | 64.45 |
| F4 | 102.1 | 0.349 | -26.1 | 78.69 |
| F5 | 186.8 | 0.381 | -34.3 | 71.17 |
| F6 | 100.4 | 0.394 | -30.3 | 78.21 |
| F7 | 188 | 0.578 | -27.9 | 42.21 |
| F8 | 94.24 | 0.232 | -30.5 | 32.77 |

Table 2. Particle size, PDI, zeta potential, and entrapment efficiency of BSA-SLN.

Preliminary characterizations of BSA-SLN

Particle Size Distribution

The particle size distribution is measured by a particle size analyzer and seen at the Z-Ave value. The Z-Ave value is the average value of the particle size in the sample. The distribution of particle sizes ranges from 94.24 nm to 186.8 nm as can be seen in Table 2. This result proves that, from the F1 to F8 formulae, all have the ideal size for the SLN, which is in the range of 50-1,000 nm. The increase of Tween 80 as a surfactant leads to a decrease in particle size and PDI due to reduced interfacial tension by a higher concentration of surfactant. The addition of soy lecithin as a cosurfactant can also decrease the mean particle size by stabilizing emulsion (Niculae et al., 2013). Formula 4 (F4) shows the best result of particle size measurement with 1.5% Tween 80 and 0.25% soy lecithin. Previous research showed that a smaller PDI (less than 0.7) indicates that the SLN has homogenous particle distribution (Trinovita et al., 2019). All the formulae tested in this research proved to achieve the ideal PDI for the SLN and therefore show that the SLN is distributed homogeneously.

Zeta Potential

Zeta potential is the electrical potential between the medium and interface of the fluid attached to the particle. Suppose the absolute value of the zeta potential is too high. In that case, the system deflocculates due to increased repulsion and cake dispersion. Suppose the zeta potential decreases below a particular value. In that case, attractive forces exceed the repulsive forces so that the particles will unite. In general, the zeta potential value of an emulsion is $\pm/-30$ mV. Based on Table 2, it is found that the zeta potential value of the formula varies from -22.8 mV to -34.3 mV. According to previous research, a zeta potential greater than ± 30 mV or -30 mV stipulates the excellent stability of the dispersion. This strongly indicates that all formulae of the SLN could be more stable (Chandran *et al.*, 2018).

Entrapment Efficiency of SLN

The entrapment efficiency value is the ability of the matrix to trap the protein. The measurement of entrapment efficiency is carried out indirectly by calculating the amount of BSA that is not trapped. The value of entrapment efficiency can be seen in Table 2. From the observations, the most considerable SLN's entrapment efficiency is F4, which is 78.69%. Greater entrapment efficiency can be obtained by increasing the concentration of surfactant to



Figure 2. TEM of the optimized formula (F4 SLN).

improve BSA's stability and emulsifying capacity in the SLN system (Trinovita *et al.*, 2019). Double-emulsification methods were found to increase drug solubility and entrapment efficiency (Ngwuluka *et al.*, 2017). This research proved that the F4 SLN provides high entrapment efficiency, which can load a higher protein concentration.

Final characterizations of BSA-SLN

After conducting preliminary characterizations of BSA–SLN, the optimum formula was obtained based on the best initial descriptions, namely, F4. Furthermore, F4 preparation was evaluated for further tests, such as particle morphology and *in vitro* drug release studies.

Particle morphology

The particle morphology of the SLN was obtained from transmission electron microscopy (TEM). The results of the TEM of F4 can be seen in Figure 2. The SLN particles consist of a hydrophobic core with a surfactant monolayer layer based on the test results. The inside of the core contains proteins that are dispersed in a lipid matrix. The SLN's morphological characteristic, almost spherical with a smooth surface, was observed using a TEM photograph. Further, no precipitation of BSA was observed, indicating the SLN was stable (Kesharwani *et al.*, 2016). **Table 3.** Cumulative drug release of SLN in HCl, PBS 7.4, and PBS 6.8 (n = 3).

| Medium | Times (minute) | % Average cumulative drug release | |
|-------------------------|----------------|--------------------------------------|--|
| HCl pH 1.2 | 15 | $0.00\% \pm 0.00\%$ | |
| | 30 | $2.08\% \pm 1.84\%$ | |
| | 45 | $7.74\% \pm 6.94\%$ | |
| | 60 | $12.58\% \pm 6.74\%$ | |
| | 90 | $21.81\% \pm 9.85\%$ | |
| | 120 | $25.97\% \pm 4.91\%$ | |
| Phosphate buffer pH 7.4 | 135 | $39.54\% \pm 7.36\%$ | |
| | 150 | $47.37\% \pm 9.53\%$ | |
| | 165 | $51.81\% \pm 8.30\%$ | |
| | 180 | $56.85\% \pm 9.77\%$ | |
| | 210 | $59.65\% \pm 10.91\%$ | |
| | 240 | $63.73\% \pm 10.72\%$ | |
| | 300 | $69.26\% \pm 9.55\%$ | |
| Phosphate buffer pH 6.8 | 315 | $77.93\% \pm 8.92\%$ | |
| | 330 | $84.36\% \pm 5.16\%$ | |
| | 345 | $88.64\% \pm 6.22\%$ | |
| | 360 | $92.00\% \pm 5.57\%$ | |
| | 390 | $97.49\% \pm 3.79\%$ | |
| | 420 | $100.94\% \pm 4.35\%$ | |
| | 480 | $102.93\% \pm 3.19\%$ | |



Figure 3. Cumulative drug release profile of F4 SLN. Data are expressed as mean \pm SD (n = 3). HCl media pH 1.2 (0–120 minutes); phosphate buffer pH 7.2 (120–300 minutes); and phosphate buffer pH 6.8 (300–480 minutes).

In Vitro Drug Release Studies

Based on the research results shown in Table 3 and Figure 3, the SLN F4 formulation could deliver a maximum drug to the colon. $102.93\% \pm 3.19\%$ BSA was released from the system after 8 hours of release study in three different media. Therefore, the F4 SLN had a potential delivery system in enhancing protein absorption in the colon. However, drug release had begun to occur in the simulated gastric and small intestine fluids. The release of BSA in the SLN system of F4 showed initial burst release followed by a slow controlled release of proteins. The drug release study indicated that, using F4 SLN, 25.97% \pm

4.91% BSA was released in the HCl solution pH 1.2 in the first 2 hours. These results can be caused by the imperfect coating of BSA—SLN with Eudragit S100. The addition of excipients such as sustained-release polymers or polysaccharides can be used to prevent premature release.

CONCLUSION

Altogether, it can be concluded that F4 showed the best formulation to obtain the optimal parameter in producing the SLN. Furthermore, the SLN coated with Eudragit S100 performed best to hold the drug release in the stomach and provide better BSA release in the colon condition ($102.93\% \pm 3.19\%$).

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CONFLICT OF INTEREST

The authors declare no conflicts of interest, financial or otherwise.

AUTHOR CONTRIBUTIONS:

All authors made substantial contributions to conception and design, acquisition of data, or analysis and interpretation of data; took part in drafting the article or revising it critically for important intellectual content; agreed to submit to the current journal; gave final approval of the version to be published; and agree to be accountable for all aspects of the work. All the authors are eligible to be an author as per the international committee of medical journal editors (ICMJE) requirements/guidelines.

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