



# The potential of leaf extract of *Pangium edule* Reinw as HIV-1 protease inhibitor: A computational biology approach

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## ABSTRACT

The leaf of Pangli (*Pangium edule*) is used as food in North Sulawesi. According to a study conducted *in vitro*, Pangli leaf extract suppressed the replication of the human immunodeficiency virus (HIV) inside CD4<sup>+</sup> helper T cells. The current study aimed to determine the compounds extracted from Pangli leaves and investigate the potential of the targeted compounds against human immunodeficiency virus type 1 (HIV-1) protease inhibitors (PIs) using an *in silico* approach. Dry powder of the Pangli leaves was extracted using n-hexane and then analyzed with gas chromatography–mass spectrometer (GC–MS) to obtain information about the compounds contained in Pangli leaves. Each compound's potential ability as HIV-1 PIs were evaluated by using AutoDock Vina and compared with nelfinavir and amprenavir, known as potent HIV PIs. Our present study revealed that at least 53 compounds were detected in the n-hexane extract of Pangli leaf using GC–MS analysis. The docking study revealed that (5.beta.) pregnane-3,20.beta.-diol, 14.alpha., 18.alpha.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diy)]-diacetate had the most binding affinity against the HIV-1 PIs. This finding provides a strong indication that this compound might have potential as an HIV-1 PIs. Our *in silico* study concluded that compound (5.beta.) pregnane-3,20.beta.-diol, 14.alpha., 18.alpha.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diy)]-diacetate in Pangli leaf has the potential to be further developed as an inhibitor of HIV-1 protease. Hence, it presumably serves as a very potent anti-HIV lead compound.

## INTRODUCTION

Acquired immunodeficiency syndrome is an asexually transmitted disease caused by the human immunodeficiency virus (HIV). This virus attacks the human immune cells, which causes a decrease in the immune system. This type of virus has two main classifications: human immunodeficiency virus type 1 (HIV-1)

and human immunodeficiency virus type 2 (HIV-2). Nevertheless, specifically, HIV-1 is mostly found to spread throughout the world and has a higher infection rate than HIV-2 (Eurtivong *et al.*, 2019; Sillapachaiyaporn and Chuchawankul, 2019; Vijayan *et al.*, 2017). The latest report by the World Health Organization (2020) revealed that the estimated annual number of people infected and living with HIV in 2018 was between 32.700.000 and 44.000.000, whereas in Indonesia alone in 2018, 550.000–750.000 people are living with HIV infections (World Health Organization, 2020).

This virus, classified as a retrovirus, is capable of infecting CD4<sup>+</sup> (cluster differential four) lymphocytes. CD4<sup>+</sup> lymphocytes play an important role to fight infections in the human body, but

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their value can be reduced due to HIV attacks (Moir *et al.*, 2011) and the mechanism of HIV includes the weakening of the immune system by infecting and destroying CD4<sup>+</sup> cells, which ultimately disrupts the functioning of human immunity. HIV attaches to the surface of CD4<sup>+</sup> and other cells to get entries. However, the virus will not enter other cells if only attached to the CD4<sup>+</sup> protein (Marchetti *et al.*, 2013; Naif, 2013). Therefore, another way that a virus needs to gain access into cells is the presence of an HIV-1 essential coreceptor. Two primary chemokine receptors that play a significant role in HIV-1 infection are CCR5 and CXCR4 (or fusion) (Maartens *et al.*, 2014; Simon *et al.*, 2006).

Treatment of HIV infection using a different combination of antiretroviral (ART) drugs involves at least three combinations of drugs (Bhatti *et al.*, 2016). There are five classifications of ART drugs circulating in the community: non-nucleoside reverse transcriptase inhibitors, nucleoside reverse transcriptase inhibitors, protease inhibitors (PIs), HIV integrase strand transfer inhibitors, and inhibitors (CCR5 coreceptor antagonist and fusion inhibitors) (Arts and Hazuda, 2012; Namthaband Mamidala, 2014). The critical step in the PI action occurs at the final stage of the viral maturation before leaving the host cell (Hughes *et al.*, 2011). In the life cycle, PIs are responsible for producing all the viral enzymes and structural proteins needed to produce mature and virulent virions (Ghosh *et al.*, 2016).

Medicinal plants are often used as sources of active compounds that have the potential to be developed as antimicrobial, antifungal, antibacterial, and antiviral (Calland *et al.*, 2012; Estevam *et al.*, 2015; Hu *et al.*, 2013; Ningsih *et al.*, 2019; Nuraskin *et al.*, 2019, 2020; Pratiwi *et al.*, 2015; Rahmad *et al.*, 2019; Sardi *et al.*, 2013; Tallei *et al.*, 2019a; Tallei *et al.*, 2019b). One of the medicinal plants in which the leaves are used as food for the people of North Sulawesi is Pangli (*Pangium edule*) (Calland *et al.*, 2012; Hu *et al.*, 2013; Rahmad *et al.*, 2019). The plant is a source of many active compounds with antibacterial (Kasim and David, 2013) and antifungal (Membalik *et al.*, 2018) activities, so it is often used as traditional medicines. According to Mapanawang and Elim (2019), Pangli leaf extract had a high inhibition activity to cease the replication of HIV. A previous study revealed that the *Cassia fistula*'s leaf extract had been used as a potential source for a laxative drug (Sakulpanich and Gritsanapan, 2009). Therefore, this research aimed to investigate *in silico* using molecular docking approach the potential of the compounds in the Pangli leaf to inhibit HIV-1 protease's activity.

## MATERIALS AND METHODS

### Sample collection and preparation

Pangli leaves were collected from Tempang village, North Sulawesi, Indonesia. The leaves were washed with water, air-dried under the shade at room temperature, and made into powder using a mixer grinder.

### Extraction of leaves

0.5 g of powdered leaves was extracted in 1 ml n-hexane solvent for 24 hours. The solution was then centrifuged, and the supernatant was used for subsequent analysis.

### Gas chromatography–mass spectrometer (GC–MS) analysis

The identification of the chemical composition of the n-hexane extract of Pangli leaves was performed using gas

chromatography (Thermo Scientific TRACE™ 1310), linked with a mass spectrometer (Thermo Scientific ISQ LT Single Quadrupole). A 60m HP-5MS Ultra Inert capillary column was used as the stationary phase. This column had a length of 30 m, 0.25 mm inner diameter, and 0.25 µm film thickness with a maximum temperature of 325/350°C. Chemical ionization with methane gas (40%, 2.0 ml/minutes) was employed in the negative ion mode at a voltage of 70 eV. Ultra-high purity helium was used as a carrier gas at a constant flow rate of 1 ml/minutes. 1 ml of sample solution was manually inserted in the splitless mode, with a split ratio of 1:50. Injection and mass transfer line temperature were set at 250°C. The total running time of the GC–MS was 60 minutes.

### Receptor preparation

The docking analysis used in this study followed the methodologies utilized in Tallei *et al.* (2020). The three-dimensional (3D) structural information of the macromolecule HIV-1 protease (PDB ID: 3NU3) determined by X-ray crystallography was downloaded from the protein data bank (<http://www.rcsb.org/pdb>). The downloaded HIV-1 protease file was opened using BIOVIA Discovery Studio Visualizer 2020. The attached water molecules and ligand to the receptor were removed, and the receptor was stored in the .pdb format. Using AutoDock Tools (Morris *et al.*, 2009), HIV-1 protease was modified with the addition of polar hydrogen and then allowed to dock ligands flexibly into the protease's rigid binding pocket, while all of the ligands' torsional bonds were set free. The file was saved in .pdbqt format.

### Determination of the active site of the receptor

The potential of the ligand-binding site (pocket) on the HIV-1 receptor 3D structure was identified using the CASTp web server (<http://sts.bioe.uic.edu/castp/>). CASTp uses the latest algorithms and computational chemical geometric analysis for the receptor's analytical validation cavities and pockets (Tian *et al.*, 2018).

### Ligand preparation

The structures of identified compounds from Pangli leaves were retrieved from PubChem online server (<http://pubchem.ncbi.nlm.nih.gov>). The targeted compounds were downloaded and saved in .sdf format. Each ligand molecule was added to the OpenBabel software (O'Boyle *et al.*, 2011) and converted to .pdb format. Using AutoDock Tools (Morris *et al.*, 2009), torque adjustment was made by detecting root and adjusted as desired. The file was saved in .pdbqt format.

### Receptor–ligand docking

The AutoDock Tools (Morris *et al.*, 2009) was used to calculate the ligand binding to HIV protease. A grid spacing of 1 Å and the grid points in X-, Y-, and Z-axis were set at 30 × 28 × 26 Å, respectively. The determination of this grid box has been dependent on the type of docking used. The three-dimensional map is only made as wide as the area to be docked (targeted docking). The grid center coordinates were placed at X: 17.547, Y: 22.310, Z: 970. AutoDock Vina (Trott and Olson, 2009) was used to perform molecular docking. Vina configuration was typed in notepad and saved as "conf.txt." The Vina program was run through a command prompt directed to the folder to be docked using the formula *vina -config conf.txt -log log.txt*. The results of the docking calculations were obtained in the notepad file format.

## Analysis and visualization

The docking conformation of the ligands and receptors was determined by selecting the pose which have the highest affinity (most negative Gibbs free energy). The results from each affinity value of the conformation were obtained in the .pdbqt format. The visualization of 3D and two-dimensional (2D) structures from the docking calculations was carried out using BIOVIA Discovery Studio Visualizer 2020. Using this program, the .pdbqt containing the highest affinity score was opened. The interaction of the receptor's amino acid with the ligand in 3D and 2D was saved as image files.

## Lipinski's rule of five (Ro5) determination

Lipinski's Ro5 components consists of H-bond donor, H-bond acceptor, molecular weight, and an octanol-water partition coefficient (log P). The pharmacokinetic properties of the compounds were predicted using SCFBIO prediction (<http://www.scfbio-iitd.res.in/software/drugdesign/lipinski.jsp>).

## RESULTS AND DISCUSSIONS

The n-hexane solvent extracted 53 predicted nonpolar compounds that were detected using GC-MS (Fig. 1). GC-MS is beneficial because it has a reasonably complete function. GC can separate semivolatile and volatile compounds with an excellent resolution; however, it cannot identify them. MS can provide detailed structural information for most compounds so that they can be identified precisely but cannot immediately separate them (Hussain and Maqbool, 2014). However, GC-MS only predicts structure because not all molecules undergo derivatization only once or receive derivatization (Barding *et al.*, 2013). At least, there are 53 possible different components with different similarity index. Some bioactive compounds possess antiviral activities (Mapanawang and Elim, 2019).

The compounds extracted from Pangli are presented in Table 1. Lipinski's Ro5 of each compound and the docking analysis results of all compounds against HIV-1 protease (PDB ID: 3NU3) are shown in Table 2. All compounds did not violate the

Ro5 criteria. Amprenavir and nelfinavir, which are antiretroviral drugs used in treating the HIV, were used as positive controls.

The active site of the protease HIV-1 receptor was identified using the CASTp web server. The active site was depicted in area 441,854 by volume of 306,180. A total of 35 amino acid residues on the active site included Arg-A:8, Leu-A:23, Asp-A:25, Gly-A:27, Ala-A:28, Asp-A:29, Asp-A:30, Val-A:32, Lys-A:45, Ile-A:47, Gly-A:48, Gly-A:49, Ile-A:50, Leu-A:76, Thr-A:80, Pro-A:81, Val-A:82, Ile-A:84, Arg-B:108, Leu-B:123, Asp-B:125, Gly-B:127, Ala-B:128, Asp-B:129, Asp-B:130, Val-B:132, Ile-B:147, Gly-B:148, Gly-B:149, Ile-B:150, Gly-B:152, Phe-B:153, Pro-B:181, Val-B:182, and Ile-B:184. The amino acid residues identified by the CASTp web server generated the results as the involvement of the amino acid residues in the native ligand-receptor interactions. This result shows that the ligand's binding site is right on the active site of the receptor. The accuracy of the ligands to bind to the receptor is seen from the amino acid residues that interact in the binding of the ligands and receptors (Rakib *et al.*, 2020). Figure 2 shows the interacting amino acid residues of HIV-1 protease with the native ligand {3-[(4-amino-benzenesulfonyl)-isobutyl-amino]-1-benzyl-2-hydroxy-propyl}-carbamic acid tetrahydro-furan-3-Yl ester (native ligand) in 2D (A) and 3D (B) configurations.

Figures 3 and 4 show the docking analysis results displaying the interacting amino acid residues of HIV-1 protease with two ligand molecules served as positive controls and the selected top six ligands that had the best binding affinity among the compounds tested. The bonds between the positive control ligand amprenavir and tested ligands phthalic acid, butyl undecyl ester, bis(3,5,5-trimethylhexyl) phthalate, and octadecane, 3-ethyl-5-(2-ethylbutyl) with the receptors showed that all of the binding amino acid residues exactly act on the active site of the receptor. However, control ligands amprenavir and other tested ligands (5.beta.) pregnane-3,20.beta.-diol, 14.alpha.,18.alpha.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diy)]-diacetate, ethyl cholate and psi.,psi.-carotene, 1,1',2,2'-tetrahydro-1,1'-dimethoxy had one amino acid residue that is not attached to the active site.

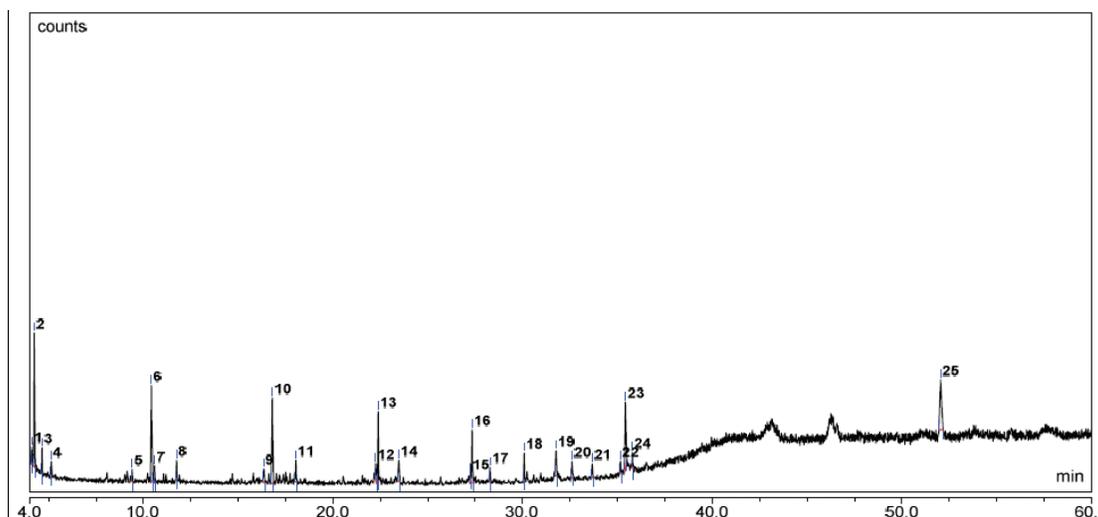


Figure 1. GC-MS chromatogram of n-hexane extract of the leaf of *P. edule*.

**Table 1.** Compounds identified using GC–MS analysis from the n-hexane Pangli leaf extract.

Peak	RT	Compound name#Hit1	Compound name#Hit2	Compound name #Hit3	Ret. area (%)
1	4.1	Hydroxylamine, O-(2-methylpropyl)-	Octadecane, 6-methyl-	Nonane, 4-ethyl-5-methyl-	1.44
2	4.24	Hexane, 2,4-dimethyl-	Octane	Heptane, 2,4-dimethyl-	9.68
3	4.64	Tridecane, 3-methylene-	3-Trifluoroacetoxylododecane	1-Octyn-3-ol	1.89
4	5.13	Decane	Hexane, 2,3,4-trimethyl	Octadecane, 6-methyl-	1.13
5	9.4	3,6-Octadecadiynoic acid, methyl ester	2-Nitrohept-2-en-1-ol	3-(Prop-2-enyloxy)dodecane	1.14
6	10.43	Decane, 2,3,5,8-tetramethyl-	Decane	Undecane	9.76
7	10.59	Octane, 4-ethyl-	Octadecane, 6-methyl-	1-Decene, 4-methyl-	1.55
8	11.74	Tridecane	Octadecane, 6-methyl-	Undecane	1.87
9	16.34	Hydroxylamine, O-decyl-	1-Iodo-2-methylnonane	2,4,6,8-Tetramethyl-1-undecene	1.48
10	16.79	Dodecane, 2,6,11-trimethyl-	1-Iodo-2-methylundecane	Pentadecane	9.26
11	18.04	Octadecane, 6-methyl-	Undecane, 4,8-dimethyl-	Decane, 2,3,5,8-tetramethyl	2.26
12	22.24	Octadecane, 6-methyl-	1-Octadecanesulphonyl chloride	1-Iodo-2-methylnonane	2.06
13	22.38	1-Iodo-2-methylnonane	1-Iodo-2-methylundecane	Decane, 2,3,5,8-tetramethyl	7.54
14	23.47	Octadecane, 6-methyl-	Dodecane, 2-methyl-	Heptadecane, 2,6,10,14-tetramethyl-	2.69
15	27.25	Octadecane, 3-ethyl-5-(2-ethylbutyl)-	Decane, 2,3,5,8-tetramethyl	Octadecane, 6-methyl-	0.64
16	27.33	Eicosane, 2-methyl-	1-Iodo-2-methylundecane	Decane, 2,3,5,8-tetramethyl	6.27
17	28.28	Octadecane, 6-methyl-	Octadecane, 3-ethyl-5-(2-ethylbutyl)-	Eicosane, 7-hexyl-	1.71
18	30.08	E-7-Tetradecenol	3-Decen-1-ol, (E)-	3-Decen-1-ol, (Z)-	3.33
19	31.77	Octadecane, 2,6-dimethyl-	Heptadecane, 2,6-dimethyl-	Octadecane, 6-methyl-	3.08
20	32.6	Octadecane, 3-ethyl-5-(2-ethylbutyl)-	Phthalic acid, butyl undecyl ester	Bis-(3,5,5-trimethylhexyl) phthalate	1.9
21	33.66	2-Decen-1-ol	Trans-2-Undecen-1-ol	Octadecane, 1-(ethenyloxy)-	1.62
22	35.16	Cyclopropanedodecanoic acid, 2-octyl-, methyl ester	Stearic acid, 3-(octadecyloxy)propyl ester	[1,1'-Bicyclopropyl]-2-octanoic acid, 2'-hexyl-, methyl ester	1.31
23	35.42	3,7,11,15-Tetramethyl-2-hexadecen-1-ol	1-Hexadecen-3-ol, 3,5,11,15-tetramethyl-	Isophytol	10.53
24	35.79	Stearic acid, 3-(octadecyloxy)propyl ester	Octadecane, 3-ethyl-5-(2-ethylbutyl)-	Octadecanoic acid, 2-(octadecyloxy)ethyl ester	1.82
25	52.04	(5.beta.) Pregnane-3,20.beta.-diol, 14.alpha.,18.alpha.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diyl)]-diacetate	Ethyl cholate	.psi...psi.-Carotene, 1,1',2,2'-tetrahydro-1,1'-dimethoxy-	14.06

HIV-1 protease was chosen as a target receptor because its use as a potential target was validated in experiments that showed that mutations in this protein resulted in defective, noninfectious virus (Patick and Potts, 1998). Its use as a promising target for antiviral drug therapy was shown by previous research results (Ghosh *et al.*, 2016; Patick and Potts, 1998; Razzaghi-Asl *et al.*, 2015). Like other retroviruses, HIV-1 protease is related to the cellular aspartyl protease family, including pepsin and renin (Adamson, 2012). Examples of this group of drugs include saquinavir, ritonavir, indinavir, nelfinavir, amprenavir, lopinavir, atazanavir, and fosamprenavir (Badley, 2005). As a corollary, nelfinavir and amprenavir were selected as positive controls in this study.

The ligand molecules with the least binding energy are considered as the compounds of the highest binding affinity (Chella-Perumal *et al.*, 2014). The compound that has a higher binding affinity more than the comparative ligand (amprenavir and nelfinavir) is (5.beta.) pregnane-3,20.beta.-diol, 14.alpha.,18.alpha.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diyl)]-diacetate. This compound shows a better score of  $-17.2$  and a peak level in the chromatogram graph indicates the maximum number of the compound (14.06%). It also has H-bonds with the receptor. The three parameters that are usually taken into consideration

when calculating the results are bond affinity, H-bond energy, and residual interactions (Ladokun *et al.*, 2018).

Other compounds phthalic acid, butyl undecyl ester, bis(3,5,5-trimethylhexyl) phthalate, .psi...psi.-carotene, 1,1',2,2'-tetrahydro-1,1'-dimethoxy-, ethyl cholate, Octadecane, 3-ethyl-5-(2-ethylbutyl)- showed the binding affinity value such as  $-11$ ,  $-11$ ,  $-10.9$ ,  $-10.5$ , and  $-10.2$  kcal/mol, respectively. These values are higher compared to nelfinavir but not with amprenavir. According to Zhang *et al.* (2008), HIV-1 proteases that belong to the retroviruses class have a protein dimer structure wherein the protein has two single amino acid subunits that form the symmetry of protein dimers. Each ligand will interact with these two amino acid chains. Therefore, each amino acid residue involved in the interaction is characterized by codes A and B.

The results of molecular docking were visualized with the BIOVIA Discovery Studio 2020 program in 3D and 2D. Interactions occur in the form of hydrophobic interactions, hydrogen bonds, and electrostatic interactions. The more hydrogen bonds formed with amino acid residues, the stronger the bond will be, and the energy score will be lower and more stable (Shah and Misra, 2011). Hydrogen bonds are interactions between hydrogen (H) atoms which are covalently bonded to atoms such as flour (F), nitrogen (N), and

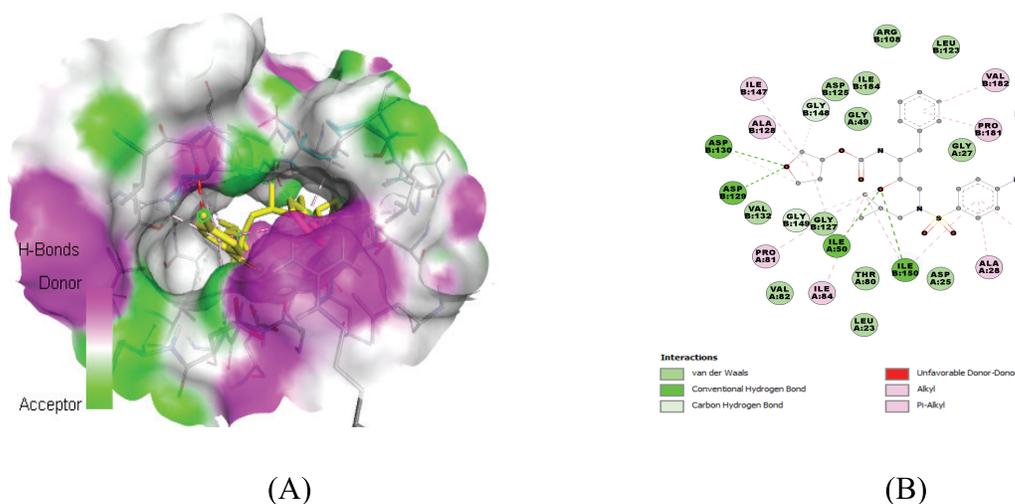
**Table 2.** Lipinski's Ro5 of HIV-1 protease potential inhibitors and their binding affinity.

Ligand properties	PubchemID	Mol. weight <500	HBD < 5	HBA<10	Log <i>p</i> < 5	Violation	Binding energy (kcal/mol)
Amprenavir	65016	–	–	–	–	–	–12.8
Nelfinavir	64143	–	–	–	–	–	–10.1
(5.beta.) Pregnane-3,20.beta.-diol, 14.alpha.,18.alpha.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diyl)]-diacetate	537242	489	0	6	5.96	1	–17.2
Phthalic acid, butyl undecyl ester	6423450	376	0	4	5.58	1	–11
Bis(3,5,5-trimethylhexyl) phthalate	34277	418	0	4	6.69	1	–11
.psi.,.psi.-Carotene, 1,1',2,2'-tetrahydro-1,1'-dimethoxy-	5366411	312	5	6	–0.05	0	–10.9
Ethyl Cholate	6452096	436	3	5	3.93	0	–10.5
Octadecane, 3-ethyl-5-(2-ethylbutyl)-	292285	282	0	0	7.76	1	–10.2
Eicosane, 7-hexyl-	292289	366	0	0	8.22	1	–9.8
Stearic acid, 3-(octadecyloxy)propyl ester	551406	312	5	6	–0.05	0	–9.7
Octadecanoic acid, 2-(octadecyloxy)ethyl ester	545632	312	5	6	–0.05	1	–9.4
1-Octadecanesulphonyl chloride	66281	352.5	0	2	6.08	1	–9.3
Eicosane, 2-methyl-	519146	296	0	0	6.69	1	–9.1
Heptadecane, 2,6,10,14-tetramethyl-	29036	296	0	0	7.86	1	–8.9
Cyclopropanedodecanoic acid, 2-octyl-, methyl ester	534597	366	0	2	6.77	1	–8.3
Decane, 2,3,5,8-tetramethyl-	545611	198	0	0	5.13	1	–8.3
Dodecane, 2,6,11-trimethyl-	35768	212	0	0	5.66	1	–8.3
Heptadecane, 2,6-dimethyl-	545603	268	0	0	7.37	1	–8.1
1-Hexadecen-3-ol, 3,5,11,15-tetramethyl-	551282	296	1	1	6.36	1	–7.9
3-Trifluoroacetyloxydodecane	534402	282	0	2	5.01	1	–7.9
Octadecane, 2,6-dimethyl-	545598	282	0	0	7.76	1	–7.9
Tridecane, 3-methylene-	519705	196	0	0	5.48	1	–7.9
Octadecane, 6-methyl-	93065	268	0	0	7.51	1	–7.8
2,4,6,8-Tetramethyl-1-undecene	536235	210	0	0	5.44	1	–7.8
Octadecane, 1-(ethenyl)-	13585	296	0	1	5.48	1	–7.8
3-(Prop-2-enoyloxy)dodecane	543279	240	0	2	4.63	0	–7.7
3,6-Octadecadiynoic acid, methyl ester	536196	290	0	2	4.87	0	–7.7
Pentadecane	12391	212	0	0	6.09	1	–7.7
[1,1'-Bicyclopropyl]-2-octanoic acid, 2'-hexyl-, methyl ester	552098	322	0	2	6.13	1	–7.5
Dodecane, 2-methyl-	15270	184	0	0	5.17	1	–7.5
Undecane, 4,8-dimethyl-	28454	184	0	0	5.02	1	–7.4
Tridecane	12388	184	0	0	5.32	1	–7.3
Nonane, 4-ethyl-5-methyl-	519249	170	0	0	4.64	0	–7.2
Isophytol	10453	296	1	1	6.36	1	–7.1
Undecane	14257	156	0	0	4.54	0	–7
Trans-2-Undecen-1-ol	5365004	170	1	1	3.28	0	–6.9
1-Decene, 4-methyl-	518719	154	0	0	4.17	0	–6.8
2-Propyl-1-heptanol	24847	158	1	1	2.97	0	–6.8
2-Nitrohept-2-en-1-ol	5362713	159	1	3	1.33	0	–6.6
3-Decen-1-ol, (E)-	5352493	156	1	1	2.89	0	–6.6
3-Decen-1-ol, (Z)-	5352846	156	1	1	2.89	0	–6.6
Decane	15600	198	0	0	5.13	1	–6.5
Heptane, 2,4-dimethyl-	16656	128	0	0	3.47	0	–6.5
Octane, 4-ethyl-	85925	142	0	0	4	0	–6.4
1-Octyn-3-ol	13166	126	1	1	1.56	0	–6.3
E-7-Tetradecenol	5362726	212	1	1	4.46	0	–6.3

(Continued)

Table 2. (Continued)

Hydroxylamine, O-decyl-	34704	173	2	2	3.02	0	-6.3
3,7,11,15-Tetramethyl-2-hexadecan-1-ol	5366244	296	1	1	6.36	1	-6.2
Hexane, 2,3,4-trimethyl	13533	128	0	0	3.32	0	-6.2
1-Iodo-2-methylnonane	537317	269	0	0	4.18	0	-6.1
Hexane, 2,4-dimethyl-	11511	114	0	0	3.09	0	-6.1
Octane	356	114	0	0	3.37	0	-5.9
2-Decen-1-ol	5364942	156	1	1	2.89	0	-5.8
Hydroxylamine, O-(2-methylpropyl)-	536422	89	2	2	0.53	0	-5.2
1-Iodo-2-methylundecane	545590	297	0	0	5.2	1	-4.9



**Figure 2.** (A) The 3D configuration of the interacting amino acid residues of HIV-1 protease receptor with the {3-[(4-amino-benzenesulfonyl)-isobutyl-amino]-1-benzyl-2-hydroxy-propyl}-carbamic acid tetrahydro-furan-3-Yl ester (native ligand) and (B) the 2D configuration which show the types of contacts formed between receptor and ligand.

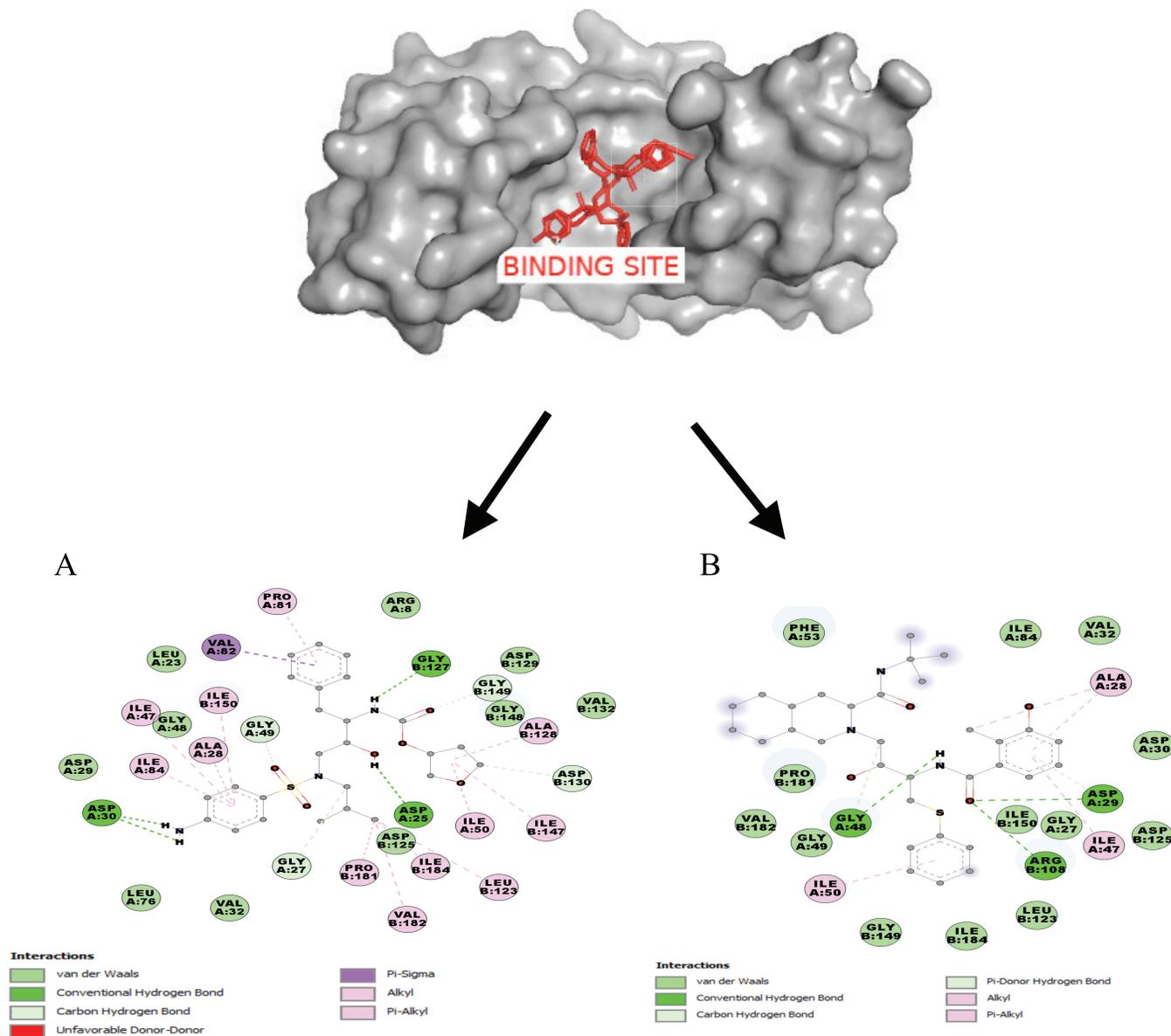
oxygen (O) (Glowacki *et al.*, 2013). Figure 4 shows that (5.beta.) pregnane-3, 20.beta.-diol, 14.alpha.,18.alpha.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diy)]-diacetate has the most hydrogen bonds with amino acids that occur in residues Asp-B:129 and Asp-B:130, so it has the lowest score energy and can also be said to be the most stable ligand among others. Bis(3,5,5-trimethylhexyl) phthalate, psi.,psi.-carotene, 1,1',2,2'-tetrahydro-1,1'-dimethoxy, and phthalic acid, butyl undecyl ester have only one hydrogen bond that occurs in the Ile-B-150, Gly-B:148, and Gly-B:149 residue, respectively.

Hydrophobic interactions avoid the liquid environment and tend to cluster together in the globular structure of proteins (Lins and Brasseur, 1995). Therefore, the hydrophobic interaction formation can reduce the interaction of amino acid residues with water. Hydrophobic interactions can be in pi-Sigma and alkyl/pi-Alkyl bonds (Zubair *et al.*, 2020). Six tested ligands have interaction with these bonds. Each of these amino acid residues is a residue of nonpolar amino acids. Nonpolar (hydrophobic) amino acid residues tend to form clusters in the interior of the protein so that they can prevent the protein from coming into contact with water (Arwansyah *et al.*, 2014).

Interactions in each ligand and amino acid residues involve several important residues in the interaction of receptors and ligands. As seen in Figure 4, only (5.beta.) pregnane-3, 20.beta.-diol, 14.alpha.,18.alpha.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diy)]-diacetate and psi.,psi.-carotene, 1,1',2,2'-tetrahydro-

1,1'-dimethoxy have a salt bridge at residue Asp-A:25 and Asp-B:130. It is because one of the most common forms of salt appears from aspartic acid. The residues Ile-A:50, Gly-B:149, Ile-A:84, Ile-B:184, and Asp-B:130 always appear in every ligand-receptor interaction, both electrostatic interaction and hydrogen bonding. Residue Ile-A:50 and Asp-B:130 are positioned inside the unprotected receptor; therefore, the water molecules make hydrogen bonds with it. It can increase receptor and ligand binding (Perez *et al.*, 2007). According to Brik and Wong (2003), the HIV-1 protease enzyme's active site is located between dimers containing aspartate catalytic residues that contain other supporting subunits. The residue is located at Asp-A: 25, Thr-A: 26, and Gly-A: 27, which also proves that the ligand (5.beta.) pregnane-3, 20.beta.-diol, 14.alpha.,18.alpha.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diy)]-diacetate precisely works on the enzyme's active site.

Al-Fifi *et al.* (2015) studied chitosan as a ligand added to phenol and hydroxymethylcarbonyl compounds, which were then docked to HIV-1 protease. The ligand has hydrogen bonds with receptors located on the amino acid residues Thr80, Ala28, Gly27, and Asp25. It shows that the amino acid ligand works in 2 areas of the receptor-binding site. Compared with the study of Manikrao *et al.* (2012), darunavir has hydrogen bonds with receptors located in the amino acid residues Asp-A: 25, Asp-A: 30, Asp-B: 25, Asp-B: 29, and Asp-B: 30. This ligand is one of the synthetic drugs approved by the Food and Drug Association



**Figure 3.** Docking position of amprenavir (A) and nelfinavir (B) on HIV-1 protease.

to inhibit HIV-1 protease. This drug only works on one area of the amino acid binding site, Asp-A: 25, whereas in this study, the ligand (5.β.) pregnane-3, 20.β.-diol, 14.α., 18.α. -[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diyl)]diacetate works in two areas of amino acid binding site. The more the bonds that form at the binding site area between the ligand and the receptor, the molecule will be more stable, and the energy will be more negative so that the potential activity of the ligand will be stronger (Manna *et al.*, 2017; Nitami and Febriansah, 2019).

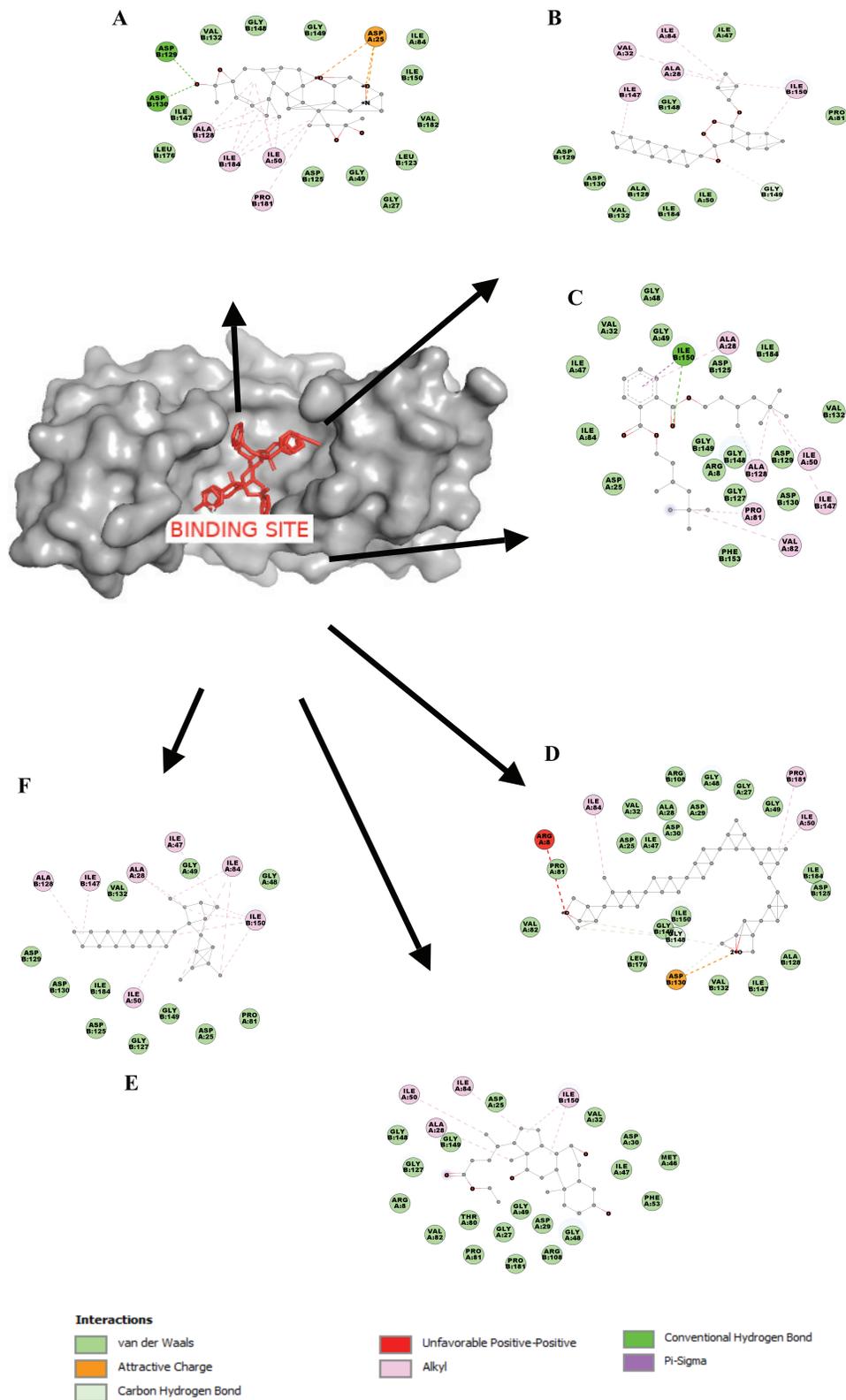
Mapanawang and Elim (2019) found that there were 15 active compounds present in this Pangi plant. Four of these compounds have the highest percentage levels and are thought to have an activity to inhibit the performance of HIV (Table 3).

The results of the binding affinity of the four compounds are not better than amprenavir, which has a binding affinity of -12.8 kcal/mol. However, squalene has a slightly higher affinity binding value than nelfinavir which has a binding affinity of -10.1

kcal/mol. The results of this *in silico* analysis also are shown in Mapanawang and Elim's (2019) research, in which some compounds in the Pangi leaves are thought to be able to inhibit the performance of HIV-1, especially in the HIV-1 protease enzyme. They proved this prediction through *in vitro* studies.

The compound (5.β.) pregnane-3,20.β.-diol, 14.α.,18.α.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diyl)]diacetate was reported in *Datura stramonium* (Altameme *et al.*, 2015) and *Melissa officinalis* (Fierascu *et al.*, 2017). Plant *D. stramonium* was also reported to have an antiviral potential (Ivanov and Nikalov, 1988). Ethyl cholate and bis(3,5,5-trimethylhexyl) phthalate were reported to have potential antiviral activity against SARS-CoV-2 in an *in silico* study (Tallei *et al.*, 2020).

To determine that a compound can be accepted as a lead compound, Lipinski's Ro5 can be used to evaluate the drug-likeness of a chemical compound. We have the following criteria: (1) less than five hydrogen-bond donors, (2) less than ten hydrogen-



**Figure 4.** Docking position of (5β.) pregnane-3,20β.-diol, 14α.,18α.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diyl)]-diacetate (A), phthalic acid, butyl undecyl ester (B), bis(3,5,5-trimethylhexyl) phthalate (C), psi.,psi.-carotene, 1,1',2,2'-tetrahydro-1,1'-dimethoxy (D), ethyl cholate (E) dan octadecane, 3-ethyl-5-(2-ethylbutyl) (F) on HIV-1 protease protein.

**Table 3.** The binding affinity of the compounds found in by Mapanawang and Elim (2019) and the percentage of their contents.

Pubchem ID	Ligand properties	Binding affinity with 3NU3	Percentage
5281	Octadecanoic acid	-8.5	24.6%
6380722	Squalene	-11.0	21.22%
985	Hexadecanoic acid	-7.2	15.08%
5280435	Phytol	-8.0	10.33%

bond acceptors, (3) molecular mass less than 500 Dalton, and (4) log P not greater than five (Shaji, 2018). Therefore, each docking compound was checked whether it met Lipinski's Ro5. This rule is used to determine whether certain chemical compounds have chemical and physical properties to be used as active drugs that can be taken orally in humans and evaluate the drug-likeness (Lipinski, 2004). Table 2 shows that 53 ligands have very low and low bioavailability because they only have at most one violation. Overall, the compound showed good compatibility as they satisfied Lipinski's Ro5. Therefore, these compounds might exhibit orally active drug-likeness. It means how a compound easily binds to receptor and how good its permeability across the cell membrane (Singh *et al.*, 2013).

## CONCLUSION

The existence of bioactive compounds shows the effective use of plant components by conventional experts for different circumstances. To sum up, the phytochemicals obtained from the leaf of *P. edule* exerted potential activity against HIV protease, of which (5.β.) pregnane-3,20.β.-diol, 14.α.,18.α.-[4-methyl-3-oxo-(1-oxa-4-azabutane-1,4-diyl)]-diacetate., phthalic acid, butyl undecyl ester, bis(3,5,5-trimethylhexyl) phthalate, .psi.,.psi.-Carotene, 1,1',2,2'-tetrahydro-1,1'-dimethoxy-, ethyl cholate, octadecane, 3-ethyl-5-(2-ethylbutyl)- have possessed significant activity against HIV protease. The proposed *in silico* studies hence suggest that the *P. edule* extract could serve as an HIV-1 PI. Further experimental analysis might be needed to justify the activity of the phytochemicals at the molecular level. In addition, many more such bioactive ingredients from medicinal plants existing in the rich Indonesian biodiversity need to be further explored. This study could provide the basis for alternative therapeutics against the invading biological pathogens, including HIV.

## AUTHOR CONTRIBUTIONS

All authors made substantial contributions to conception and design, acquisition of data, or analysis and interpretation of data; took part in drafting the article or revising it critically for important intellectual content; agreed to submit to the current journal; gave final approval of the version to be published; and agree to be accountable for all aspects of the work.

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## CONFLICTS OF INTEREST

The authors report no conflicts of interest in this work.

## ETHICAL APPROVAL

This study does not involve the use of animals or human subjects.

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