

# Design and in vitro characterization of a topical nanoemulsion-enriched hydrogel of econazole nitrate

Mowafaq M. Ghareeb

Department of Pharmaceutics, College of Pharmacy, University of Baghdad, Baghdad, Iraq.

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## ARTICLE INFO

Received on: 21/08/2018  
Accepted on: 06/12/2018  
Available online: 31/01/2019

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### Key words:

Econazole nitrate,  
Nanoemulsion-enriched  
hydrogel.

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## ABSTRACT

**Background/Objective:** The aim of the study was to prepare a nanoemulsion (NE)-enriched hydrogel of econazole nitrate (EC) for a topical broad-spectrum antimycotic application. EC is classified as a Biopharmaceutical Classification System Class IV (low permeability, low solubility) drug. Nanotechnology is one of the most important approaches to improve drugs solubility and permeability.

**Methods:** Depending on solubility study of EC in NE components which include oils (peppermint oil, eucalyptus oil, and olive oil), surfactants (Tween 40, Brij 35, and Ceto stearyl alcohol), and co-surfactants (Propylene glycol, PEG 400, and Glycerol); peppermint oil was selected as the oil phase. Tween 40 and propylene glycol were selected as the surfactant and co-surfactant, respectively. Deionized water was used as an aqueous phase. A phase diagram was prepared with a 5:1 weight ratio of tween 40 to propylene glycol. One percent of EC was loaded into the selected NE system and combined with hydrogel consisting of carbopol 934 or HPMC as gelling polymer at three different percentages. Evaluation and in vitro release of the product were done.

**Results:** NE Formula (F10) which is composed of 10% w/w of each the oil and surfactant mixture was selected as optimum NE which shows a particle size of 80.57 nm and acceptable viscosity and pH (5.65). Formula (G6) which is composed of NE (F10) combined with hydrogel 3.5% HPMC shows satisfactory physical properties with the complete and prolonged release over 8 hours.

**Conclusion:** The results suggest that the prepared NE-enriched hydrogel can be considered as a promising delivery system for class IV antifungal drug.

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## INTRODUCTION

Econazole nitrate (EC) is a broad-spectrum antimycotic imidazole derivative agent with fungistatic action and is widely used against *Candida albicans* (Ogata *et al.*, 1983). According to the Biopharmaceutical Classification System, econazole nitrate is classified as a Class IV (low permeability, low solubility) drug (Amidon *et al.*, 1995). It is very slightly soluble in water (1.48 µg/ml) but soluble in methanol (Suñer-Carbó *et al.*, 2017). It has a partition coefficient (log *P*) of about 5.5 and a pKa of 6.6 (Remington, 2006; Stahl and Wermuth, 2002). One of the commercial topical dosage

forms of EC is a cream containing 1% of EC. This cream provides neither a complete penetration of EC nor a long residence time of the formulation on the infected area (Ge *et al.*, 2014). Most of the topical dermatological formulations such as creams and ointments have the disadvantages of a less spreading coefficient, sticky nature, and the need for rubbing during application. These limitations can be eliminated through a hydrogel formulation. Despite various advantages of hydrogels, they have a major limitation in the delivery of hydrophobic drugs. Thus, nanoemulsion (NE) combined with hydrogel has proved a boon in the delivery of hydrophobic drugs topically and provides the advantages of hydrogel formulation (Hoare and Kohane, 2008).

NE is a transparent isotropic, heterogeneous system of two immiscible liquids consisting of a fine dispersion of drugs in nanodroplets. It is stabilized by an interfacial layer of surfactant and co-surfactant (Sood *et al.*, 2014). NEs are kinetically and

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\*Corresponding Author  
Mowafaq M. Ghareeb, Department of Pharmaceutics, College of Pharmacy, University of Baghdad, Baghdad, Iraq.  
E-mail: [mopharmacy@yahoo.com](mailto:mopharmacy@yahoo.com); [mopharmacy@gmail.com](mailto:mopharmacy@gmail.com)

thermodynamically stable systems lacking any coalescence or apparent flocculation upon storage and use within the period of the expiration date. NEs with nanodroplet size in the range of 20–400 nm, uniform droplet size distribution commonly have different biopharmaceutical properties comparing to the traditional emulsions type which have particle size higher than 500 nm (Ahmed *et al.*, 2012; Shakeel *et al.*, 2012; Sugumar *et al.*, 2014). NEs show many advantages comparing to conventional emulsions, such as more stability, higher interfacial area, fast absorption by internalization into the enterocyte, enhancement of drug aqueous solubility, and consequently, expected improvement in bioavailability (Gorain *et al.*, 2014). US Food and Drug Administration has approved a number of NEs in the past few years such as Norvir<sup>®</sup>, Neoral<sup>®</sup>, and Fortovase<sup>®</sup> (Zhao *et al.*, 2013). The aim of this study was to control the delivery of econazole nitrate topically with fast and easy permeation through skin epidermis by incorporating the drug in an NE combined with hydrogel to provide more residence time of drug at the infected site with optimum skin penetration.

## MATERIALS AND METHODS

### Materials

Econazole nitrate was generously supplied by Furat pharmaceutical industries, Iraq. Peppermint oil was purchased from Bar-sur-Loup Grasse (a.m) France, eucalyptus oil and olive oil were purchased from Ati lab., Netherland. Tween 40 was purchased from Grin land chemical comp., UK. Brij 35 was purchased from Himedia Lab. Pvt Ltd, India. Cetostearyl alcohol was purchased from Fluca Chemie AG, Switzerland. All other chemicals were of analytical grade.

### Methods

#### Differential scanning calorimetry

The sample of the drug was analyzed by differential scanning calorimetry (DSC) using a Mettler Toledo SR system. Three mg of each sample was placed into a pierced aluminum container. The study was performed under the static air atmosphere in the temperature range of 20°C–320°C at a heating rate of 10°C/minute. The peak temperature was determined.

### HPLC analysis of econazole nitrate

To determine the drug content, the Reverse phase (RP)-high-pressure liquid chromatographic (HPLC) separation was performed on a Shimadzu liquid chromatographic system equipped with an LC pump, DGU-20A5 UV detector, and injector system fitted with 100 µl loop volume. The HPLC analysis was carried out at a flow rate of 2.0 ml/minute using a mobile phase composed of methanol-aqueous 0.05 M ammonium dihydrogen phosphate (85:15, v/v), and the UV-detector was set at 230 nm. The mobile phase was prepared daily, filtered through a 0.45 µm membrane filter (Millipore), and sonicated before use. A Thermo C18 column (100 mm × 4.6 mm id, 3.5 µ) was used for the separation.

The retention time of drug is 4.78 minutes, as shown in the chromatogram in Figure 1 and drug content was calculated from the calibration curve ( $y = 0.7354x + 0.00865$ ) (Cavrini *et al.*, 1982).

### Determination of econazole nitrate solubility in NE components

Powder of EC was added in excess to different NE components which include oils (peppermint oil, eucalyptus oil, and olive oil), surfactants (Tween 40, Brij 35, and cetostearyl alcohol), and co-surfactants (propylene glycol, PEG 400, and glycerol). After shaking at 25°C for 24 hours, samples were centrifuged, and then the supernatant was separated and subjected to centrifugation at 3,000 rpm for 5 minutes to remove the undissolved drug. Samples of these solutions were then collected, and the drug concentration was determined using the HPLC method described here.

### Construction of pseudo-ternary phase diagram

To find out the concentration of components that produce NEs, pseudo-ternary phase diagrams were constructed using water titration method at an ambient temperature (25°C). Peppermint oil was selected as the oil phase. Tween 40 and propylene glycol were selected as surfactant and co-surfactant, respectively. Deionized water was used as an aqueous phase. A phase diagram was prepared with a 5:1 weight ratio of tween 40 to propylene glycol.

Fourteen formulas of NE were prepared at different weight ratios, as shown in Table 1. The mixtures of oil, surfactant, and co-surfactant at certain weight ratios were diluted with

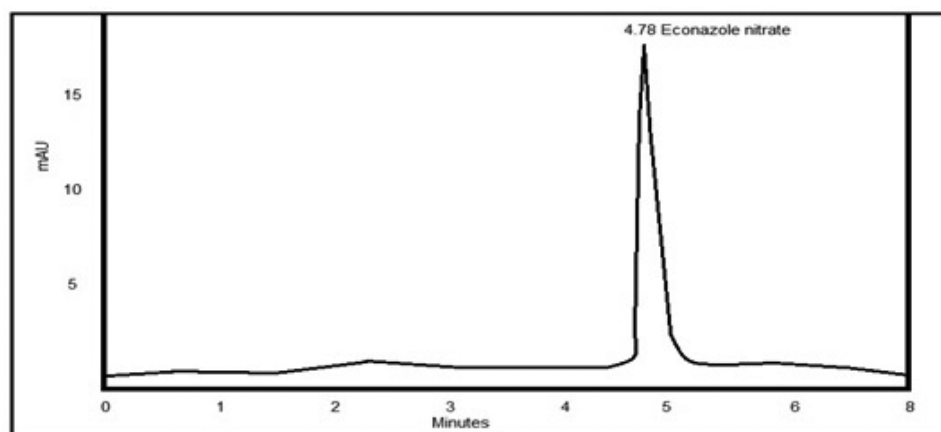


Figure 1. HPLC chromatogram of econazole nitrate.

water dropwise under moderate magnetic stirring. After being equilibrated, the mixtures were assessed visually to determine it is one phase NEs (Jadhav *et al.*, 2015).

#### Drug loading in the NE system

The loading procedure of 1% w/w EC into the selected NE system includes heating of all the system components, except water up to  $40^{\circ}\text{C} \pm 5^{\circ}\text{C}$  using a water bath. Then, the drug is added and the system is subjected to sonication for about 30 minutes at  $40^{\circ}\text{C} \pm 5^{\circ}\text{C}$  using a Copley Scientific sonicator 2200E, UK to ensure complete dissolving of the powder.

After ensuring a complete dissolving of all the molecules of the drug visually, the NE system is cooled down to about  $30^{\circ}\text{C} \pm 5^{\circ}\text{C}$  and then completed by adding the required weight of water up to 100 g.

#### Preparation of nanoemulsion-enriched hydrogel of Econazole nitrate

After loading of drug into the best-chosen nanoemulsion (EC NE), the hydrogel part preparation involves either Carbopol 934 or HPMC as a gelling agent in addition to methyl and propylparaben as a preservative, while propylene glycol as a co-solvent, as shown in Table 2.

The Carbopol 934 gel bases were prepared by dispersing Carbopol 934 in deionized water with a continuous stirring at a

moderate speed using a mechanical shaker (Formulations G1, G2, and G3). The pH of all the formulations was adjusted to 6–6.5 using tri-ethanolamine (Ranga *et al.*, 2012).

In formulations G4, G5, and G6, the gel was prepared by dispersing HPMC in hot distilled water ( $80^{\circ}\text{C}$ ) and the dispersion was cooled and left overnight.

Methyl and propylparaben were dissolved in propylene glycol and mixed with the aqueous phase. Both the NE and aqueous phases were heated to  $70^{\circ}\text{C}$ – $80^{\circ}\text{C}$  separately; then the NE was added to the aqueous phase with continuous stirring until the system temperature drops down to room temperature. The obtained NE was mixed with the hydrogel in 1:1 ratio with gentle stirring to obtain the NE-enriched hydrogel.

#### Determination of particle size of nanoemulsions

The particle size and polydispersity were determined for the prepared NE three times by the use of Angstrom nano laser particle size analyzer (Guttoff *et al.*, 2015).

#### Determination of viscosity of nanoemulsions

The viscosities of the prepared NEs were determined using ProRheo Viscometer ProRheo, Bell technology Ltd, New Zealand spindle 01 at 50 rpm and room temperature ( $25^{\circ}\text{C}$ ) (Guttoff *et al.*, 2015).

**Table 1.** Composition of econazole nitrate nanoemulsion.

Formula code	The ratio of surfactant to co-surfactant = 5:1			Total nanoemulsion weight (g)
	Nanoemulsion composition as weight ratio			
	surfactant/co-surfactant mixture	oil	water	
F1	1	10	89	100
F2	2	10	88	100
F3	3	10	87	100
F4	4	10	86	100
F5	5	10	85	100
F6	6	10	84	100
F7	7	10	83	100
F8	8	10	82	100
F9	9	10	81	100
F10	10	10	80	100
F11	20	10	70	100
F12	30	10	60	100
F13	40	10	50	100
F14	50	10	40	100

**Table 2.** Formulas of econazole nitrate nanoemulsion-enriched hydrogel.

Material(g)	G1	G2	G3	G4	G5	G6
EC NE	50	50	50	50	50	50
Carbopol 934	0.5	1	1.5			
HPMC				1.5	2.5	3.5
Propylene glycol	5	5	5	5	5	5
Methylparaben	0.03	0.03	0.03	0.03	0.03	0.03
Propylparaben	0.01	0.01	0.01	0.01	0.01	0.01
Water up to	100	100	100	100	100	100

### Determination of pH of nanoemulsions

The pH values of the prepared NE samples were measured by a pH meter, Italy at room temperature, pH buffer solutions ( $4.01 \pm 0.02$  and  $9.21 \pm 0.02$ ) were used to calibrate the instrument, and the measurements were carried out in triplicates (Guttoff *et al.*, 2015).

### Physical appearance of nanoemulsion-enriched hydrogel

The prepared hydrogel formulations were inspected visually for their color, homogeneity, consistency, grittiness, and phase separation.

### pH of nanoemulsion-enriched hydrogel

One gram of the hydrogel formulation was dispersed in 10 ml of distilled water, and a digital pen pH meter was used to determine the pH value.

### Viscosity of nanoemulsion-enriched hydrogel

The viscosities of all the prepared hydrogel formulas were measured using ProRheo Viscometer ProRheo, Bell technology Ltd, New Zealand spindle 02 and speed 50 rpm at 25°C, measurements were carried out in triplicate, and the average value was calculated (Baibhav *et al.*, 2012).

### Spreadability of nanoemulsion-enriched hydrogel

The apparatus suggested by Kumar and Verma (2010) was used to determine the spreadability. It consists of a wooden block, which is attached to a pulley at one end. Spreadability was measured on the basis of “Slip” and “Drag” characteristics of the hydrogel. A ground glass slide was fixed on the wooden block. An excess of hydrogel (about 1 g) under study was placed on this ground slide.

The hydrogel preparation was then sandwiched between this slide and second glass slide having the same dimension as that of the fixed ground slide. The second glass slide is provided with the hook.

A weight of 500 mg was placed on the top of the two slides for 5 minutes. The diameter of the produced circle was measured and used as comparative values for spreadability. It is calculated by using the following formula.

$$\text{Spreadability} = W \cdot D/T$$

Where,  $W$  = wt. on the upper slide;  $D$  = diameter of spreading gel; and  $T$  = time until spreading stopped (Kumar and Verma, 2010).

### Extrudability of nanoemulsion-enriched hydrogel

It is a common test to measure the force required to extrude the material from the tube. The extrudability test was carried out using 30 g of hydrogel filled in an aluminum tube. The plunger was adjusted to hold the tube properly. The force of 1 kg/cm<sup>2</sup> was applied for 30 seconds. The quantity of gel extruded was weighed. The procedure was repeated at three equidistance places of the tube (Vikrant and Sonali, 2013).

### Determination of drug content of nanoemulsion-enriched hydrogel

An accurately weighted portion from prepared NE-enriched hydrogel of econazole nitrate was dissolved in methanol, and this solution was filtered and the drug content was determined by RP-HPLC method.

### In vitro drug release from nanoemulsion-enriched hydrogel

A sample of 1 g of the preparation was spread on a filter paper. The hydrogel-loaded membrane was firmly stretched over the edge of a glass tube of 2 cm diameter (opened at both ends). The membrane was tied up with a rubber. The tube was then immersed in the dissolution vessel, which contained 500 ml of the release medium, PBS pH 7.4, and maintained at  $32^\circ\text{C} \pm 0.5^\circ\text{C}$ . The medium was stirred at 50 rpm and 5-ml aliquots were withdrawn from the solution at specified time intervals. Equal volumes of fresh release medium were used to replace the withdrawn samples. The samples were assayed using the HPLC method mentioned previously. Each data point represents the average of three determinations. *In vitro* release studies were recorded for a period of 8 hours (Abdul Rasool *et al.*, 2011).

## RESULTS AND DISCUSSION

### Differential scanning calorimetry

The results in Figure 2 obtained from DSC analysis showed an endothermic peak of EN at 165.32°C and exothermic peak at 199.32°C and these results similar to the published papers which indicate the purity of raw material (El *et al.*, 2017).

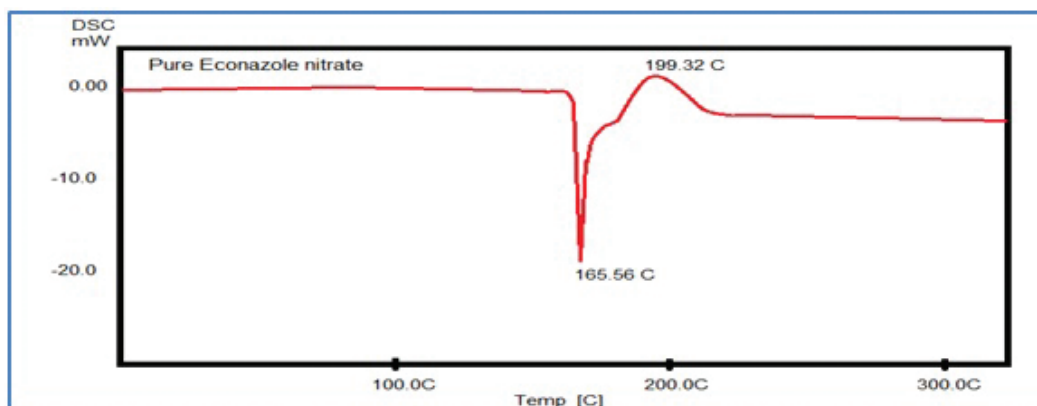


Figure 2. DSC thermogram of econazole nitrate.

### Econazole nitrate solubility in NE components

The solubility EC in oil listed in Table 3 was in the following order; Peppermint oil > Eucalyptus oil > Olive oil, while the solubility in surfactant was in the following descending order; Tween 40 > Brij 35 > Cetostearyl alcohol. On the other hand, the solubility of EC in co-surfactant was in the following descending order; Propylene glycol > PEG 400 > Ethanol. So, suggested a system of NE is peppermint oil: Tween 40/propylene glycol: water with the 5:1 weight ratio of tween 40 to propylene glycol.

### Pseudo-ternary phase diagram

The constructed pseudo-ternary phase diagram (Fig. 3) shows that formulas F10–F14 produce one phase NE and physical characteristics listed in Table 4. The droplet sizes for the formulations are represented in Table 4. The result shows that the droplet diameter and polydispersity index decrease with increasing surfactant mixture ratio in the system. These

**Table 3.** Solubility of econazole nitrate in nanoemulsion components.

Nano-emulsion components	EC solubility mg/ml
Oils	
Peppermint oil	82
Eucalyptus oil	57
Olive oil	6.85
Surfactant	
Tween 40	3.87
Brij 35	2.3
Cetostearyl alcohol	1.2
Co-surfactant	
Propylene glycol	8.45
PEG 400	3.8
Ethanol	1.79

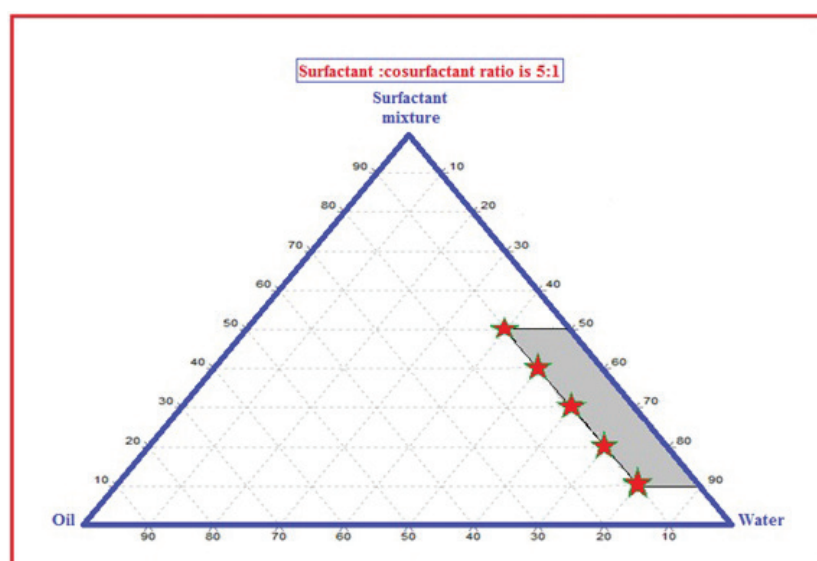
results are similar to that obtained by Saberi *et al.* (2013), and this may be due to the higher surfactant mixture concentration result in higher surfactant molecules at the interface and lower interfacial tension producing smaller droplet size. Tween 40 Hydrophilic-Lipophilic Balance (HLB = 15.6) is considered a good choice for preparation of NE since it is usually produced the smallest droplets size (Hasani *et al.*, 2015). Average droplet size distribution of formulas produced NE ranged from 54.82 to 80.57 nm. Also, the results showed that the viscosity of NE increased without change in pH (5.65) as surfactant mixture concentration increases. Although formula F14 has the smallest droplet size, formula (F10) was chosen as an optimum formula for drug loading because of lowest viscosity and low surfactant mixture percentage gave nanodroplet size NE.

### Evaluation of prepared nanoemulsion-enriched hydrogel of EC

The results in Table 5 of six prepared formulas (G1–G6) of the NE-enriched hydrogel of EC indicate no phase separation, acceptable drug content (99.60–98.70), and acceptable pH range of 5.82–6.52. The viscosity of product showed an increase as the gelling polymer percent increased. Formulas containing HPMC showed higher viscosities than carbopol containing formulas. The spreadability and extrudability of the product showed a decrease as the gelling polymer increased.

### In vitro drug release

The release of drug from the prepared product (Fig. 4) shows an increase in the drug release retardation as carbopol or HPMC percentage increases although HPMC showed higher retardation effect on drug release than carbopol. The results indicated that formula G6 which contains 3.5% HPMC has an extended but complete drug release a period of 8 hours and can be considered as the best formula of NE-enriched hydrogel with good physical properties.



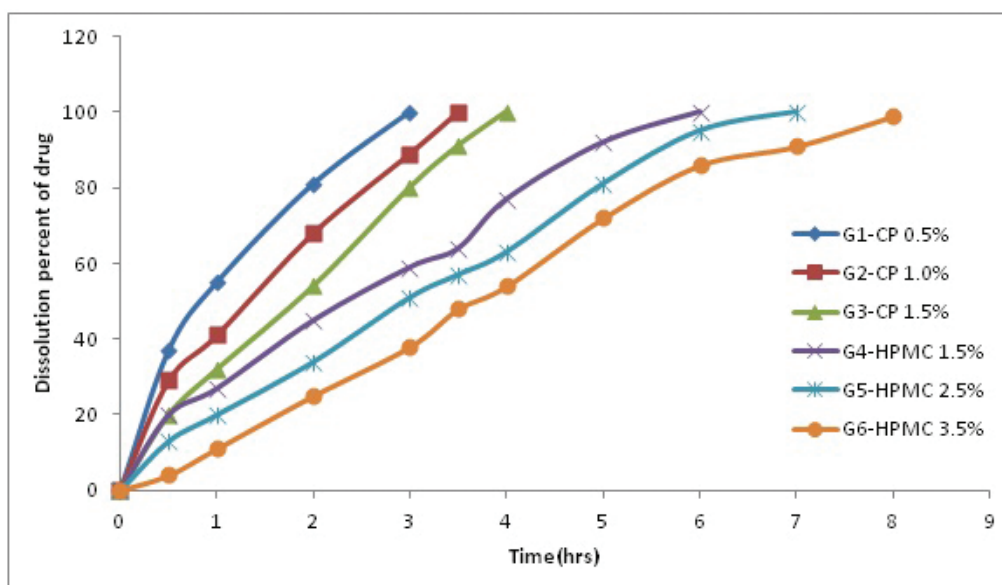
**Figure 3.** Pseudo-ternary phase diagram of peppermint oil: Tween 40/propylene glycol: water exploring the nanoemulsion region as marked points.

**Table 4.** Physical properties of the prepared systems which are composed of peppermint oil: Tween 40/propylene glycol: water.

Formulas code	Appearance	Mean particle size (nm)	Viscosity (CPS)	Polydispersity index (PDI)	pH
F1	Turbid				
F2	Turbid				
F3	Turbid				
F4	Turbid				
F5	Turbid				
F6	Turbid				
F7	Turbid				
F8	Turbid				
F9	Turbid				
F10	Clear	80.57	0.82	0.083	5.65
F11	Clear	75.44	0.85	0.073	5.65
F12	Clear	66.21	0.86	0.072	5.65
F13	Clear	62.73	0.89	0.069	5.65
F14	Clear	54.82	0.99	0.041	5.65

**Table 5.** Evaluation parameters of nanoemulsion-enriched hydrogel of EC.

Formulas code	Phase separation	pH	Viscosity (CPs)	Spreadability (g.cm/sec)	Extrudability (g/sec)	Drug content (%)
G1	No	5.82	1250	35	1.7	99.60
G2	No	5.85	1360	33	1.5	98.40
G3	No	5.91	1730	30	1.3	98.40
G4	No	6.35	5260	20	0.9	99.30
G5	No	6.47	6470	19	0.8	99.73
G6	No	6.52	7200	18	0.7	98.70

**Figure 4.** The dissolution profile of EC from NE-enriched with carbopol (G1–G3) or with HPMC (G4–G6).

## CONCLUSION

Results of this study suggest that formula of nanosized topical product with the complete and extended release of poorly soluble antifungal drug can be prepared with expected higher penetration through the skin due to small particle size.

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### How to cite this article:

Ghareeb MM. Design and in vitro characterization of a topical nanoemulsion-enriched hydrogel of econazole nitrate. *J Appl Pharm Sci*, 2019; 9(01):051–057.