

# Inhibition of dipeptidyl peptidase IV by fexofenadine: Virtual screening study

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## ABSTRACT

Dipeptidyl peptidase IV (DPP IV) is relatively new anti-diabetic target. DPP IV inhibitors lower fasting and postprandial glucose concentrations by preventing the degradation of the natural hypoglycemic incretin hormones: glucose-dependent insulinotropic peptide and glucagon-like peptide-1. In this work, the high throughput docking software FRED was used as a virtual screening tool against *in house* built drug database to discover new DPP IV inhibitors. One of the highest ranking hits, the antihistamine drug fexofenadine, was found to inhibit recombinant human DPP IV *in vitro* with  $IC_{50} = 4.6 (\pm 1.0) \mu M$ . The anti-diabetic effect of fexofenadine was validated *in vivo* by oral glucose tolerance test. These results could be helpful in the development of novel DPP IV inhibitors based on fexofenadine scaffold for the treatment of type 2 diabetes.

## INTRODUCTION

Dipeptidyl peptidase-IV (DPP IV) is a protease responsible for *in vivo* inactivation of several endogenous peptides including glucose-dependent insulinotropic polypeptide (GIP) and glucagon-like peptide-1 (GLP-1). GLP-1 and GIP are incretins released from the gut in response to intraluminal glucose and play an important role in glucose homeostasis (Campbell and Drucker, 2013; Drucker, 2003). Furthermore, they inhibit apoptosis of  $\beta$ -cells and enhance their regeneration (Brubaker and Drucker, 2004; Yabe and Seino, 2011). However, the effects of these incretins in diabetic patients are significantly reduced and suppression of DPP IV has been shown to improve glycemic control by enhancing the insulinotropic effects of incretins (Karagiannis *et al.*, 2014; Kazafeos, 2011). Therefore, the search for DPP IV inhibitors, as antidiabetic agents, is an active area of research (Almasri *et al.*, 2013; Shu *et al.*, 2014).

Both reversible and irreversible DPP IV inhibitors with *in vivo* efficacy were discovered. The reversible substrate analogs inhibitors take benefit of the high preference of the S1 hydrophobic site of DPP IV for proline binding. These inhibitors are generally amide derivatives of pyrrolidine or thiazolidine. Examples include valine-pyrrolidide and isoleucine-thiazolidide (Fig. 1) (Peters, 2011). The irreversible inhibitors contain, in addition to the five-membered ring, an electrophilic group at position 2 and are even more potent. Nitrile group is the most common electrophilic group used and is capable of forming an enzyme-imidate adduct with Ser630 within the active site and is present in saxagliptin and vildagliptin (Fig. 1) (Kushwaha *et al.*, 2014).

Sitagliptin is an example on reversible non-peptide heterocyclic inhibitors (Fig. 1) (Kim *et al.*, 2005; Liu *et al.*, 2012). It does not contain a pyrrolidine ring to bind within the S1 pocket. As an alternative, sitagliptin has a pyrazole-piperazine fused ring system and binds to the enzyme so that the 2,4,5-trifluorophenyl group binds within the S1 hydrophobic site and the amide group is in the opposite direction to that of substrate analog inhibitors (Kim *et al.*, 2005). Xanthine, aminomethylpyrimidine, and isoquinoline DPP IV inhibitors are another examples on reversible inhibitors in which the amide bond is not necessary for activity (Kushwaha *et al.*, 2014). Several antidiabetic drugs have recently been

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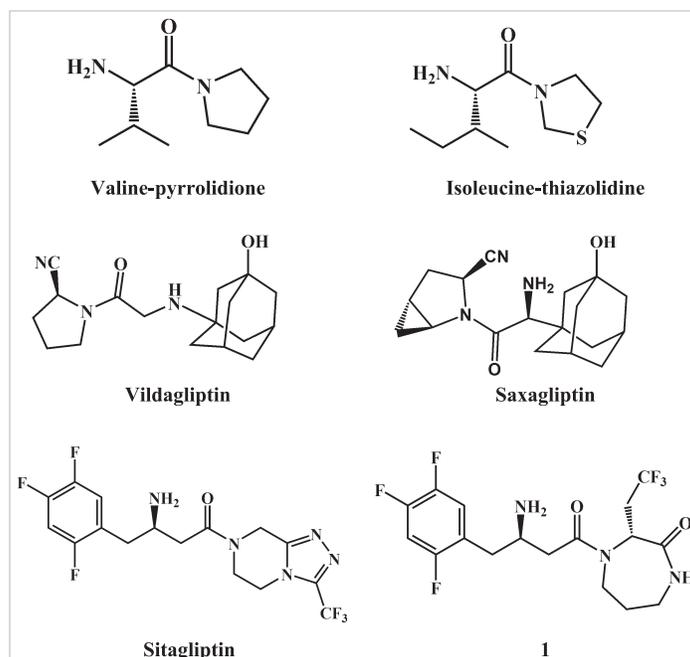


Figure 1. Structures of different classes of DPP IV inhibitors.

explored for their effect on DPP IV. Metformin has been reported to inhibit DPP IV (Kaganda *et al.*, 2015; Lindsay *et al.*, 2005) with  $IC_{50}$  ranging from 29 to 98  $\mu$ M. Thiazolidinediones derivatives, e.g., rosiglitazone and troglitazone; meglitinides, e.g., nateglinide; sulfonyleureas, e.g., glybenclamide and tolbutamide have also been reported to inhibit DPP IV (Duffy *et al.*, 2007). However, to date no work was done to thoroughly investigate the effect of other non-diabetic drugs on DPP IV activity.

Herein, structure-based virtual screening approach was employed to screen *in house* built drug database for hits capable virtually to fit within the DPP IV binding pocket in order to identify drugs with potential anti-DPP IV activities. High throughput FRED docking software was used as a database screening tool.

## METHODOLOGY

### Molecular docking

The 3D coordinates of DPP IV were obtained from the Protein Data Bank (PDB code: 2G63, resolution 2.0Å) (Pei *et al.*, 2007). Hydrogen atoms were added to the protein structure and the co-crystallized ligand was extracted using Discovery studio Visualizer. The protein structure was exploited in subsequent docking experiments in the presence of explicit water and without energy minimization. High throughput docking/scoring technique using FRED docking software (FRED, 2006) was employed to screen *in house* built drug database (1,490 drugs) for hits capable virtually to fit within the DPP IV binding pocket of co-crystallized complex. Although further details about FRED docking are described elsewhere (Almasri *et al.*, 2008; Bustanji *et al.*, 2009), a brief description is provided below. FRED docking program requires a target protein structure, a box defining the active site of the target, a multi-conformer database of the ligand(s) intended for docking and a number of optional parameters as input. The conformational space of the database drugs was explored employing OMEGA program (OMEGA, 2013). The protein structure

and ligands conformers are treated as rigid structures during the docking experiment. Upon completion of docking calculation, poses are scored and ranked. The top-scored pose was chosen using the ScreenScore function. The software parameters that best retrieved the co-crystallized pose of DPP IV complex were employed in the docking experiment. The docking experiment was validated by extracting the crystallographic bound cyanopyrrolidine inhibitor and redocking it to the binding site of DPP IV. This validation resulted in very close model to the crystallographic structure.

### *In vitro* DPP IV enzyme inhibition assay

The assay was conducted using DPP IV Drug Discovery Kit (Biomol, Germany) in which DPP IV inhibition activity was quantified by measuring the release of para-nitroaniline (pNA) from chromogenic substrate (H-Gly-Pro-para-nitroaniline) by DPP IV as previously described (Almasri *et al.*, 2008).

Briefly, fexofenadine was dissolved in DMSO and diluted with Tris buffer solution (pH 7.5). The Recombinant enzyme was diluted in the assay buffer solution to obtain a final concentration of 17.34  $\mu$ Uml<sup>-1</sup>. Afterward, appropriate volumes of fexofenadine stock solution were added to a 15  $\mu$ l aliquot of the enzymatic solution into microplate wells and the volume completed to 50  $\mu$ l with buffer. The obtained mixtures were incubated for 10 minutes at 37°C. Finally, 50  $\mu$ l of 0.20 mM substrate solution was added to each well. The absorbance of the plate was read at 405 nm using microplate reader (BioTek) and the rate of absorbance reduction was monitored over 10 minutes and compared with a negative control. P32/98, a standard DPP IV inhibitor, was used as positive control. The DMSO concentration was kept less than 1.0% in all runs.

The absorbance *versus* time was plotted and the slope and percent inhibition were calculated using the following formula:

Percent inhibition =  $(1 - \text{slope of the compound/slope of the negative inhibitor}) * 100\%$ .

### Oral glucose tolerance test (OGTT) in mice

The animal experimental protocol comply with the guide of laboratory animals use (National Research Council, 2011). Male Balb/c mice 25–30 g obtained from Jordan University animal house were used in the OGTT. Both control and treatment groups were matched for body weight in all experiments. A dose of 34 mg/kg of fexofenadine was given intraperitoneally to Balb/c mice 30 minutes prior to oral glucose load evaluation. Mice were fasted for 6 hours at the end of the light/dark cycle before the test. Glucose solution was administered orally at a dose of 2 g/kg of body weight, subsequently, the blood samples were collected at 0, 15, 30, 60, and 120 minutes and analyzed for glucose level using glucometer (Arkray, Inc., Japan).

### RESULTS AND DISCUSSION

The identification of lead compounds having some activity against a biological target and progressive optimization of the potency and physicochemical properties of these compounds are key steps in the early-stage drug discovery. Virtual screening (VS) is one of the most widely used powerful techniques in lead discovery. The fundamental goal of VS is to reduce the huge virtual chemical space of library of molecules to a manageable number of compounds to screen against a specific target protein (Shoichet, 2004). Two common strategies are usually adopted in database searching: structure-based virtual screening (SBVS) and ligand-based virtual screening (LBVS). In SBVS, we make use of the 3D structural information of the target protein and computational techniques in virtual database screening and to study the essential molecular interactions involved in ligand-protein binding, and thus explain experimental results at the molecular level. Moreover, the use of VS in drug discovery has additional advantages of lowering the cost and speed up the process of delivering new drugs into the market.

In the current work, molecular docking was employed in virtually screening a drug database. Interestingly, one of the highest ranking hits was fexofenadine, a second generation antihistamine drug (Fig. 2). Although fexofenadine structure differs from known DPP IV inhibitors, it binds to DPP IV (Fig. 4A) in a comparable way to known DPP IV inhibitors and interacts with the same essential key amino acids (Almasri *et al.*, 2008). In the present work, docking simulations revealed that fexofenadine binds to the active site of DPP IV with one of the two phenyl moieties laid within the hydrophobic S1 pocket formed by Tyr547, Tyr631, Val656, Trp659, Tyr662, Tyr666, Val711, and His740. A major contribution to binding by this phenyl moiety is achieved by edge-to-face  $\pi$ - $\pi$  interaction with Tyr666 and face-to-face  $\pi$ - $\pi$  interaction with Tyr662. These interactions are comparable to the co-crystallized ligand cyanopyrrolidine ring interactions within the S1 pocket (Fig. 4B) (Pei *et al.*, 2007). The entry of DPP IV substrate within the S1 pocket will be hindered because of the binding of phenyl ring within the same cavity. Furthermore, the key amino acid Tyr547 was engaged in aromatic stacking interaction with terminal *para*-substituted phenyl ring in a similar manner to the co-crystallized ligand aromatic ring (Fig. 4). Moreover, the pyrrolidine nitrogen of the co-crystallized structure

is strongly attracted *via* a hydrogen-bond-reinforced ionic interaction with Glu205 (2.0 Å) and Glu206 (3.8 Å) and similarly the positively ionizable nitrogen of the piperidine ring is within the electrostatic interaction range with the key amino acid Glu206 (6.0 Å). The hydroxyl group attached to the quaternary carbon is hydrogen bonded to the amino group of Asn710 (2.3 Å), which are part of the P2 region, and to Tyr662 (2.5 Å) similar to carbonyl group interaction in the co-crystallized ligand. Additionally, the secondary hydroxyl group is hydrogen-bonded to Tyr547 (3.2 Å) comparable to the cyano group in the co-crystallized structure. The hydroxyl group of Tyr547 was found to play a role in the stabilization of the negatively charged oxyanion of the formed intermediate *via* hydrogen bonding (Thoma *et al.*, 2003). Finally, the hydrocarbon butyl chain of fexofenadine is involved in hydrophobic (van der Waals) interactions with Phe357. The obtained docking results have encouraged us to evaluate the anti-DPP IV activity of fexofenadine *in vitro*. The  $IC_{50}$  of the drug against DPP IV was  $4.6 (\pm 1.0) \mu\text{M}$  (Fig. 3).

To substantiate the anti-DPP IV activity of fexofenadine, it was decided to test its ability to decrease blood glucose level in Balb/c mice using OGTT. Intraperitoneal administration of fexofenadine (34 mg/kg) inhibits glucose excursion, as shown in Figure 5A. Furthermore, it illustrated significant reduction in the AUC of plasma glucose/time curve at three dosing levels (Fig. 5B).

The discovered binding mode of fexofenadine put forward several ways for more improvements in its activity.

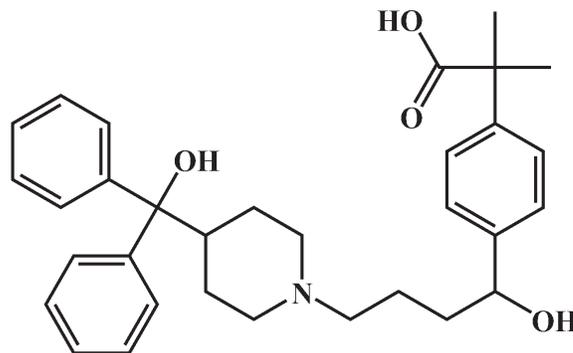


Figure 2. The chemical structure of fexofenadine.

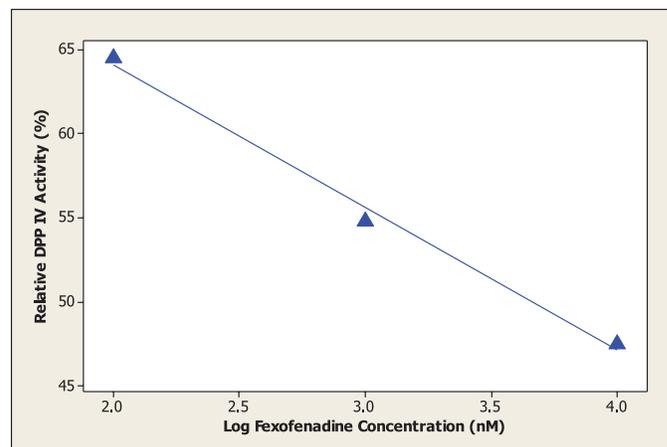
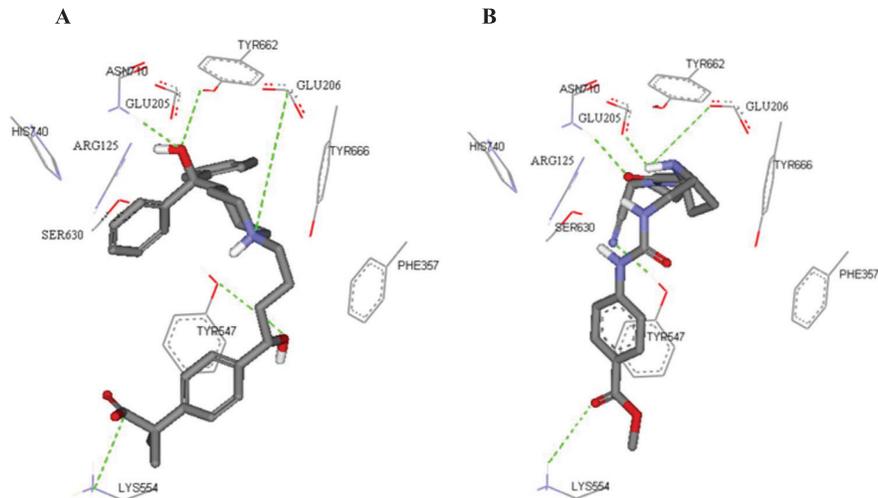
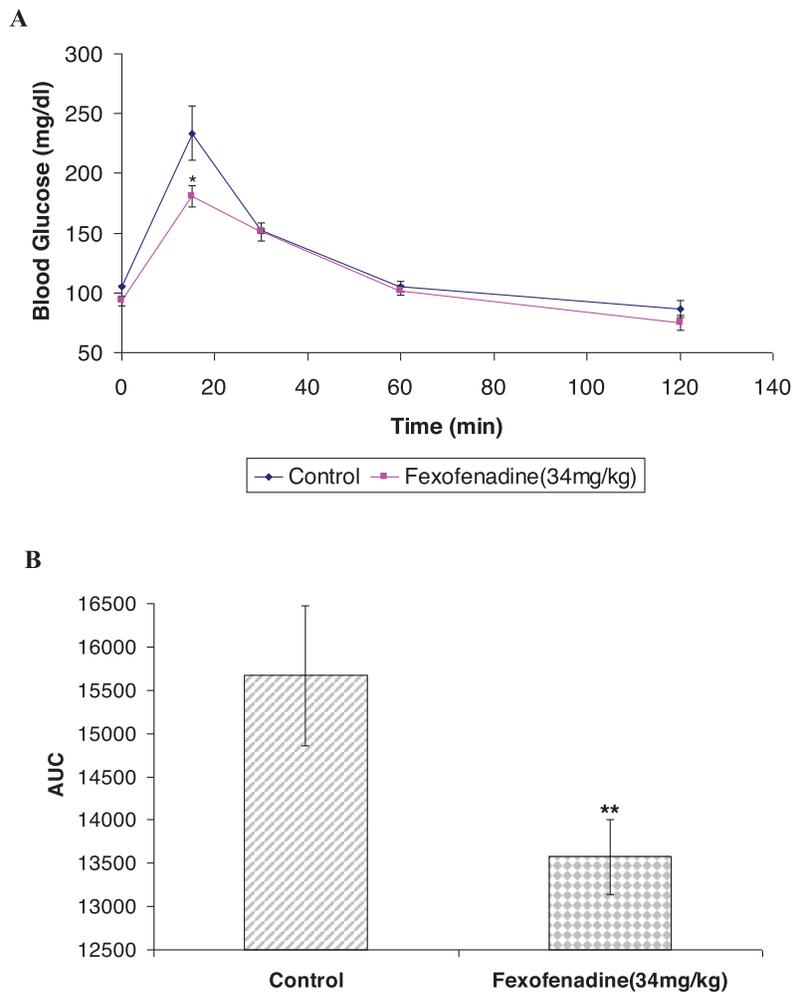


Figure 3. The effect of variable fexofenadine concentrations on the activity of dipeptidyl peptidase IV (DPP IV).



**Figure 4.** The highest ranking FRED predicted pose for fexofenadine (A) compared to the co-crystallized cyanopyrrolidine ligand (B) within the catalytic site of DPP IV (PDB code: 2G63, resolution 2.0). For clarity, only polar hydrogen atoms are shown. Hydrogen bonds are shown as dotted green lines. Catalytic site residues are represented as sticks with the atoms in standard color.



**Figure 5.** Oral glucose tolerance test of fexofenadine using male Balb/c mice. A dose of 34 mg/kg of fexofenadine was administered intraperitoneally to Balb/c mice, and glucose (2 g/kg, p.o.) was given 30 minutes latter (0 minutes). Plasma glucose concentration was measured at the indicated time (A). In A, (♦) Control (DMSO), (■) fexofenadine 34 mg/kg. Data are represented as mean  $\pm$  standard error of the mean. (B). AUC of glucose concentration curve (A) was calculated and displayed. \* $p$  value  $< 0.05$  vs. control, \*\* $p$  value  $< 0.01$  vs. control.

Replacing hydroxyl group attached to quaternary carbon by  $\text{NH}_2$  group could enhance the drug binding via forming a strong hydrogen-bond-reinforced ionic interaction with both Glu205 and Glu206 and hydrogen bond with Tyr662. Moreover, rigidifying or shortening the butyl linker is expected to enhance the activity. It has been shown that the free energy of binding is reduced by approximately 0.7 kcal/mol for each freely rotating bond in a ligand (Andrews *et al.*, 1984). Several success stories are reported in the literature where rigidification of a flexible ligand leads to significant increase in affinity. Moreover, removing the second aromatic ring, which is attached to the quaternary carbon and does not make interactions within the active site, would decrease the molecular weight, and thus increase ligand efficiency (Reynolds *et al.*, 2008).

## CONCLUSION

Docking simulations adopted in this work revealed fexofenadine as potential lead inhibitor for the anti-diabetic target DPP IV *in vitro*. Furthermore, the anti-diabetic effect of fexofenadine was validated by the *in vivo* OGTT. Moreover, the obtained results in this study could be helpful in the discovery of new fexofenadine-like DPP IV inhibitors.

## ETHICAL APPROVAL

All applicable international, national, and/or institutional guidelines for the care and use of animals were followed.

## CONFLICT OF INTEREST

The authors declare that they have no conflict of interest.

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