

# Virtual screening of natural and synthetic inhibitors of cyclooxygenase COX-2 enzyme using docking-scoring functions

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## ABSTRACT

In this paper, theoretical elucidation of cyclooxygenase interaction with synthetic and natural bioactive molecules using molecular docking is studied with molecular docking implicating solvation parameters. Obtained results show that synthetics and natural inhibitors of thym interact differently with cyclooxygenase inflammation enzyme after including solvation parameter and confirm primary studies concerning the anti-inflammatory effect. We conclude that the solvation parameter must be taken into account in all molecular docking studies because of different results which permits a better comprehension of the inhibition process and more clear ideas to develop new drugs. Results allow us to propose chlorogenic as a novel molecule to be developed into a new novel drug.

## INTRODUCTION

Molecular modeling methods present useful tools in medicinal and biological research. Indeed, molecular modeling is very important and indispensable to understand the interaction between disease's enzymes and inhibitors for the conception of new drugs; it permits to save time and financial spending. According to different research studies, natural molecules from thyme essential oil and flavonoids (Apigenine, Luteoline, Thymol, Carvacrol, Naringenine, and Chlorogenic) are extremely recommended to treat inflammation by inhibition responsible enzyme. Many inhibitors are used for Cyclooxygenase-2 (COX-2) inhibition but synthetic ones are the most used namely the non-steroidal anti-inflammatory drugs (NSAID) (Etoricoxib, Celecoxib, Ibuprofen, and Rofecoxib). Inflammation is a part of the complex biological response of body tissues to harmful stimuli, such as pathogens, damaged

cells, or irritants. The function of inflammation is to eliminate the initial cause of cell injury, clear out necrotic cells and tissues damaged from the original insult and the inflammatory process, and to initiate tissue repair (Miliani, 2007). Thyme, known as a powerful antiseptic, is also an antibacterial, antiviral, anti-fungal, and major antiparasitic. As such, it is used against different infections, both of the otorhinolaryngology sphere and of the respiratory, genitourinary, and digestive systems. Thyme is also a major antiviral effective against herpes simplex. Several studies indicated that thyme can be useful for people suffering from inflammatory diseases (ID) (Kuate, 2017). Indeed, it has been proved that carvacrol, a component of thyme oil, activate PPAR $\alpha$  and  $\gamma$  and suppresses COX-2 expression (Hotta *et al.*, 2010). These results may be important in understanding the anti-inflammatory and anti-lifestyle-related disease properties of carvacrol. Also, it has been indicating that combined treatment with appropriate concentrations of thyme and oregano essential oils can reduce the production of pro-inflammatory cytokines, and thereby attenuate 2,4,6-trinitrobenzene sulfonic acid induced colitis in mice (Bukovska *et al.*, 2007). In a study from Japan's Nara Women's University, researchers found that one of thyme oil's constituents, carvacrol, actually inhibits the COX-2 enzyme part of the body's inflammatory process that produces pain

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(Katsukawa *et al.*, 2010). This strategy of inhibiting COX-2 has been utilized by pharmaceutical medications including NSAID. Until now, some of these NSAID and COX-inhibitor drugs come with side effects such as cardiovascular and digestive problems, which docile herbs like thyme don't seem to come with (Salmalian *et al.*, 2014). Actually, the comparisons between different ligands inhibition of the same enzyme can be done by means of molecular modeling; in fact, this technique is used widely in drug design. In this work, we aim to carry out a comparative study of the COX-2 inhibition between synthetic inhibitors namely (NSAID) and the natural inhibitors (thyme essential oil derivatives). In order to rationalize the properties of the inhibitors and to determine the reaction processes involving these compounds, we studied the interaction and binding of the complex formed with inhibitors of COX-2 (natural and synthetic), with better complementarities (better activity). Molecular modeling study is performed using molecular operating environment (MOE) software to advise among molecules contained in thyme (natural inhibitor) which is better for treatment of the Inflammation, and also what is the best synthetic inhibitor (NSAID) including solvation parameter.

## MATERIAL AND METHODS

### Cyclooxygenase-2 enzyme

COX-2 is highly inducible in response to cellular activation by hormones, pro-inflammatory cytokines, growth factors, and tumor promoters. COX-2 activates procarcinogens, promotes angiogenesis, and indirectly increases free radical production (Picot and Loll, 1994).

### Cyclooxygenase-2 synthetic and natural inhibitors

Some COX-2 inhibitors are used in a single dose to treat pain after surgery. COX-2 inhibitors have been found to be effective in suppressing inflammatory neurodegenerative pathways in mental illness, with beneficial results in trials for the major depressive disorder as well as schizophrenia (Hemler *et al.*,

1976). Synthetic and natural inhibitors are reported in Tables 1 and 2.

### Preparation and optimization of both enzyme and inhibitors

Download of COX-2 was done from PROTEIN DATA BANK (code 4PH9) with the three-dimensional structure obtained by X-ray diffraction (resolution 1.81 Å). We note that the COX-2 crystallizes as a monomer (Fig. 1) with residues and atoms. Compounds of inhibitors were downloaded from the PubChem database. Structures and CID code are reported in Tables 3 and 4. Using MOE software (MOE, 2013), we select the active site in the enzyme and we minimize the energy of both enzyme and molecules (a, b, and c). Energy minimizing was done under the following conditions: temperature = 300°K, pH = 7, the geometry was performed using the field strengths in the MMFF94x implanted in MOE and Hamiltonian AM1. Figure 2 shows the active site of the enzyme with a molecule of co-crystallization. Minimized energy of ligands and their toxicity are given in Table 5. Natural ligands present a very important biological activity in accordance with the Lipinski rule of 5 (Powers *et al.*, 2006).

### Docking and building complexes

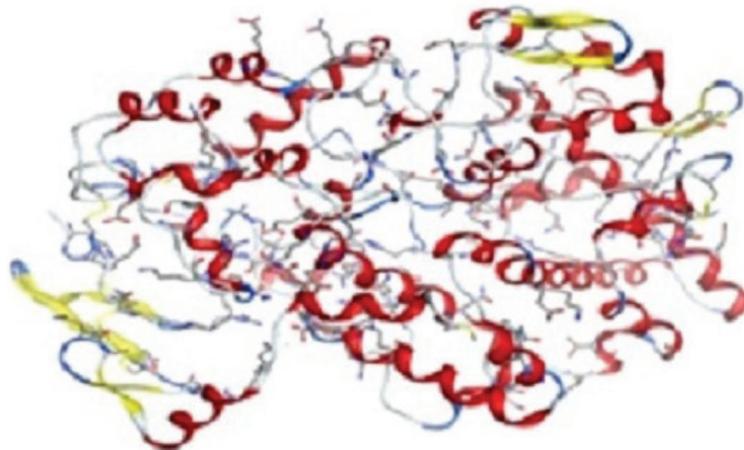
After ligand building, we proceed to positioning it in the active site of COX-2. For this, we used the molecular docking module using MOE software. Once the ligand-receptor complex is formed, it will adopt the most stable conformation, i.e., the lowest energy level. The purpose of the dock application is looking at favorable conformational binding between medium-size ligands and a not so soft macromolecular target, which is usually a protein (Goto *et al.*, 2008; Manikrao *et al.*, 2011). For each compound, a number of conformations called poses were generated to identify favorable binding modes. The search for binding modes is generally constrained to a small specific region of the receptor called the active site. First docking is without the solvation parameter (without H<sub>2</sub>O molecules), the second docking is done taking into account the presence of H<sub>2</sub>O molecules.

**Table 1.** Physico-chemical properties of synthetic inhibitors for cyclooxygenase.

No.	Name	IUPAC name	PubChem CID	Molar mass	Formula
1	Etoricoxib (Arcoxia)	5-Chloro-6'-methyl-3-[4-(methylsulfonyl)phenyl]-2,3'-bipyridine	123619	358.84	C <sub>18</sub> H <sub>15</sub> ClN <sub>2</sub> O <sub>2</sub> S
2	Ibuprofen	2-(4-(2-Methylpropyl)phenyl)propanoic acid	3672	206.29	C <sub>13</sub> H <sub>18</sub> O <sub>2</sub>
3	Celecoxib	4-[5-(4-Methylphenyl)-3-(trifluoromethyl)pyrazol-1-yl]benzenesulfonamide	2662	381.3730	C <sub>17</sub> H <sub>14</sub> F <sub>3</sub> N <sub>3</sub> O <sub>2</sub> S
4	Rofecoxib	3-(4-methylsulfonylphenyl)-4-phenyl-2H-furan-5-one	5090	314.355	C <sub>17</sub> H <sub>14</sub> O <sub>4</sub> S
5	Valdecoxib	4-(5-methyl-3-phenyl-1,2-oxazol-4-yl)benzenesulfonamide	119607	314.359	C <sub>16</sub> H <sub>14</sub> N <sub>2</sub> O <sub>3</sub> S

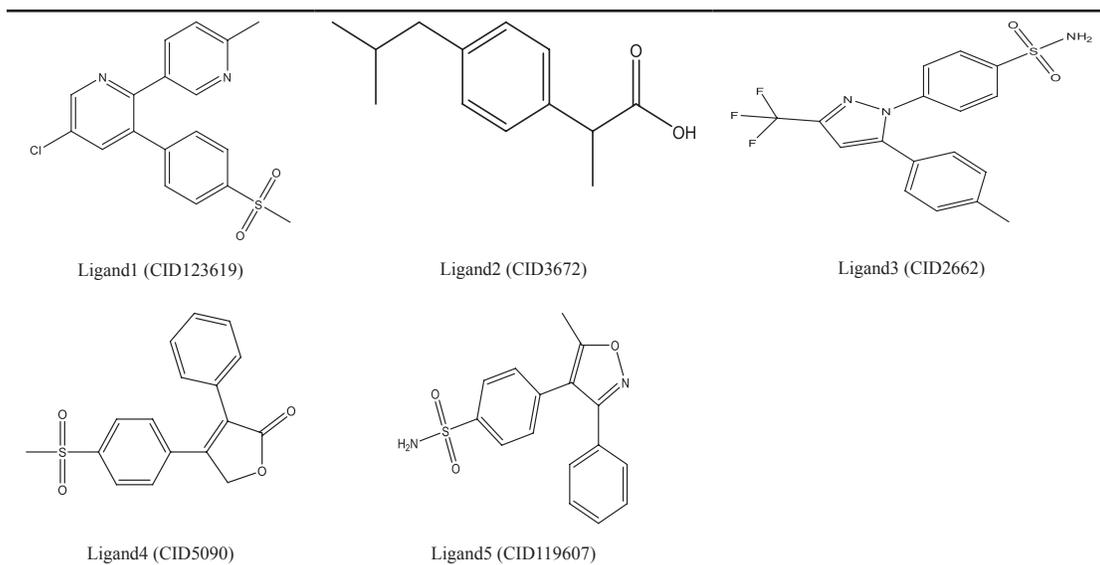
**Table 2.** Physico-chemical properties of cyclooxygenase natural inhibitors.

No.	Name	IUPAC name	PubChem CID	Molar mass	Formula
6	Apigenine	5,7-dihydroxy-2-(4-hydroxyphenyl)-4H-1-benzopyran-4-one	5280443	270,2369	C <sub>15</sub> H <sub>10</sub> O <sub>5</sub>
7	Luteoline	5,7-dihydroxy-2-(3,4-dihydroxyphenyl)-chromen-4-one	5280445	286,2363	C <sub>15</sub> H <sub>10</sub> O <sub>6</sub>
8	Naringenine	5,7-dihydroxy-2-(4-hydroxyphenyl)-2,3-dihydrochromen-4-one	932	272.256	C <sub>15</sub> H <sub>12</sub> O <sub>5</sub>
9	Chlorogenique	(1S,3R,4R,5R)-3-[(E)-3-(3,4-dihydroxyphenyl)prop-2-enoyl]oxy-1,4,5-trihydroxycyclohexane-1-carboxylic acid	9476	354,3087	C <sub>16</sub> H <sub>18</sub> O <sub>9</sub>
10	Thymol	5-methyl-2-propan-2-ylphenol	6989	150.2210	C <sub>10</sub> H <sub>14</sub> O
11	Carvacrol	2-methyl-5-propan-2-ylphenol	10364	150.2210	C <sub>10</sub> H <sub>14</sub> O



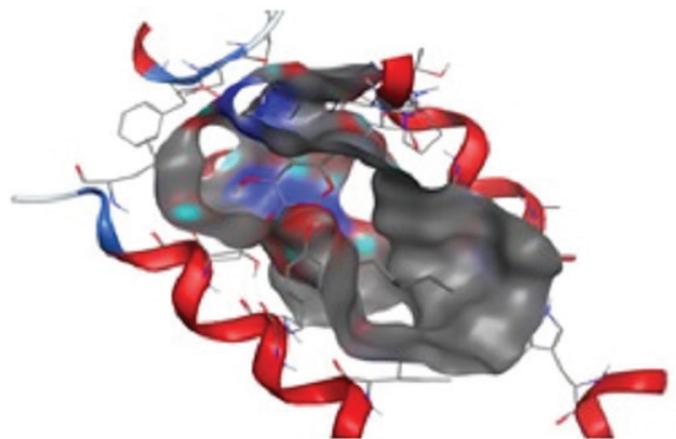
**Figure 1.** Simplified model of COX-2 enzyme.

**Table 3.** COX-2 synthetic inhibitors.

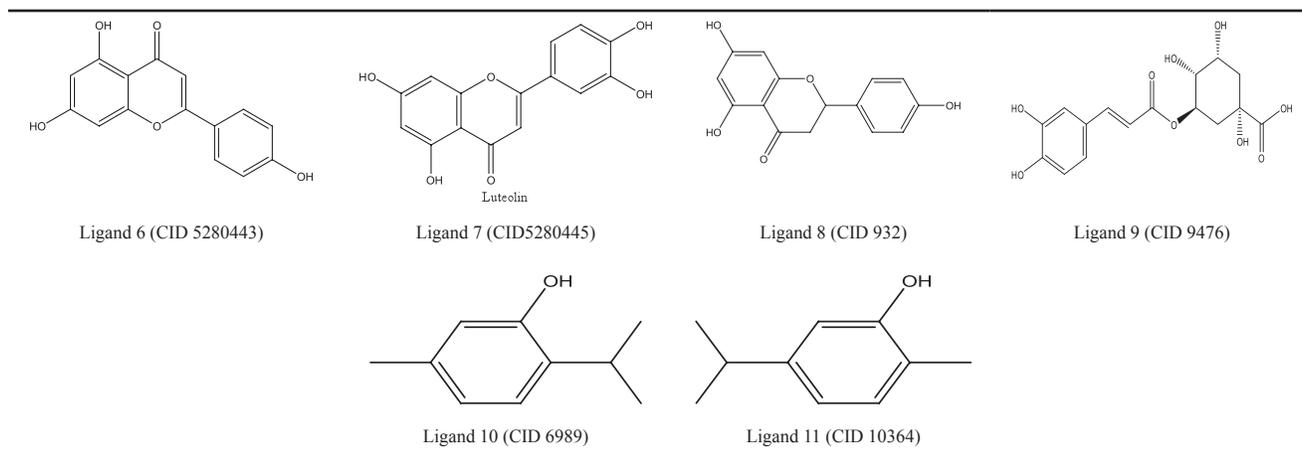


## RESULTS

The obtained results are given in Tables 6–9 which showed that the orientation of the ligands plays a significant role in positioning the ligands in the active site of the enzyme; one can conclude that the introduction of bulky groups causes a rearrangement of conformation inside the cavity of the active site, which will be probably the complementarity and consequently the activity. Two-dimensional molecular method of the screen has been attributed to the MOE software, which is designed to visualize the active sites of the complex (protein-ligand). The ligand is prepared and made with an improved 2D depiction layout algorithm, and protein residues version is arranged around it to indicate links spatial proximity (Labute *et al.*, 2001). Residues are marked with their amino acid code of three letters and job classification (Clark *et al.*, 2006; 2008). If there are multiple channels in the system, the positions are prefixed by the letters of the alphabet. Interactions between 2.5 Å and 3.1 Å are considered high and those between



**Figure 2.** Isolated active site of COX-2 enzyme.

**Table 4.** COX-2 natural inhibitors.**Table 5.** Energy minimization of synthetic and natural molecules (Kcal/mol).

Ligand	Molecules	Energies(Kcal/mol)	LogP	LogS	Toxicity
<i>A. Non-steroidal anti-inflammatory drug (NSAID)</i>					
1	Etoricoxib	7.94964e + 001	4.18	-4.88	No
2	Ibuprofen	4.16073e + 001	1.74	-3.90	No
3	Celecoxib	9.63550e + 001	4.92	-5.87	No
4	Rofecoxib	5.84172e + 001	2.56	-4.35	No
5	Valdecoxib	5.73013e + 001	4.06	-5.42	No
<i>B. Natural molecules from thym</i>					
6	Apigenine	3.87013e + 001	2.42	-3.46	No
7	Luteoline	3.71596e + 001	2.13	-3.10	No
8	Naringenine	4.96049e + 001	2.61	-2.45	No
9	Chlorogenique	6.05794e + 001	-1.98	-1.75	No
10	Thymol	2.46880e + 001	2.82	-2.69	No
11	Carvacrol	2.31478e + 001	2.82	-2.69	No

**Table 6.** Energy balance of five synthetic complexes without water (Kcal/mol).

Mol	Score	RMSD-refine	Econf	E-place	E-score1	E-refine	E-score2
Ligref	-5.98953867	3.33348608	90.9748306	-55.971561	-11.535297	-14.653989	-5.98953867
Complexe-1	-5.5511508	0.874776959	101.846687	-72.530326	-11.707312	-1.5834083	-5.5511508
Complexe-2	-5.04773426	1.19075108	-21.584247	-45.121669	-10.714869	-8.9626646	-5.04773426
Complexe-3	<b>-5.81052923</b>	<b>1.10454786</b>	<b>97.3103104</b>	<b>-111.58769</b>	<b>-11.930303</b>	<b>-8.5994358</b>	<b>-5.81052923</b>
Complexe-4	-5.65758514	2.09524655	61.886795	-45.174076	-10.906614	-11.889950	-5.65758514
Complexe-5	<b>-5.99548626</b>	<b>1.29100323</b>	<b>8.29981232</b>	<b>-62.341873</b>	<b>-11.399541</b>	<b>-10.729586</b>	<b>-5.99548626</b>

S = the final score is the score of the last step, RMSD\_refine = the mean square deviation between the laying before refinement and after refinement pose, E\_conf = energy conformer, E\_place = score of the placement phase, E\_score1 = score the first step of notation, E\_refine = score refinement step and number of conformations generated by ligand E\_score2 = score the first step notation, number of poses = Number of conformations.

**Table 7.** Energy balance of five synthetic complexes in water (Kcal/mol).

Mol	Score	RMSD-refine	E-Conf	E-place	E-score1	E-refine	E-score2
Ligref	-5.98953867	3.33348608	90.9748306	-55.971561	-11.535297	-14.653989	-5.98953867
Complexe-1	-4.12505865	1.17556381	137.318176	-37.738773	-12.849839	33.137359	-4.12505865
Complexe-2	<b>-6.95738792</b>	<b>1.47064769</b>	<b>-24.571804</b>	<b>-43.209449</b>	<b>-11.920853</b>	<b>-7.4806613</b>	<b>-6.95738792</b>
Complexe-3	-5.458056062	1.63348651	114.992928	-12.394986	-14.971972	-13.645959	5.458056062
Complexe-4	<b>-6.59314966</b>	<b>2.07637978</b>	<b>60.4710274</b>	<b>-11.949843</b>	<b>-13.657680</b>	<b>9.8269319</b>	<b>-6.59314966</b>
Complexe-5	-6.21619844	1.12231815	11.4346561	-51.159500	-14.687335	16.068758	-6.21619844

**Table 8.** Energy balance of six natural complexes without water (Kcal/mol).

Mol	Score	RMSD-refine	E-conf	E-place	E-score1	E-refine	E-score2
Ligref	-5.98953867	3.33348608	90.9748306	-55.971561	-11.535297	-14.653989	-5.98953867
Complexe-6	-5.59241152	1.16817784	11.374465	-55.811657	-11.775262	-12.479655	-5.59241152
Complexe-7	<b>-6.05466223</b>	<b>1.10594547</b>	<b>11.7080164</b>	<b>-55.302166</b>	<b>-12.968064</b>	<b>-12.566381</b>	<b>-6.05466223</b>
Complexe-8	<b>-6.11540222</b>	<b>1.52258539</b>	<b>13.5257673</b>	<b>-55.108078</b>	<b>-12.103406</b>	<b>-14.377618</b>	<b>-6.11540222</b>
Complexe-9	-5.64023209	1.36583459	4.23327494	-101.06633	-12.995637	-7.0157551	-5.64023209
Complexe-10	-4.5781517	1.24809861	14.5425158	-47.112014	-9.0278539	-9.9694833	-4.5781517
Complexe-11	-4.74362326	1.62836754	10.1525116	-43.908340	-8.6032047	-10.292403	-4.74362326

**Table 9.** Energy balance of six natural complexes in water (Kcal/mol).

Mol	Score	RMSD-refine	E-conf	E-place	E-score1	E-refine	E-score2
Ligref	-5.98953867	3.33348608	90.9748306	-55.971561	-11.535297	-14.653989	-5.98953867
Complexe-6	-5.95740128	0.723993957	13.9371176	-60.723114	-18.410959	3.6698224	-5.95740128
Complexe-7	-5.95684624	0.464796275	17.0319653	-63.233600	-20.522974	9.0500469	-5.95684624
Complexe-8	<b>-6.10765457</b>	<b>1.54488122</b>	<b>24.1423931</b>	<b>-61.313488</b>	<b>-18.576202</b>	<b>6.8101348</b>	<b>-6.10765457</b>
Complexe-9	<b>-9.11451435</b>	<b>2.35423303</b>	<b>12.7112684</b>	<b>-55.682323</b>	<b>-23.526517</b>	<b>-12.988913</b>	<b>-9.11451435</b>
Complexe-10	-5.717237	0.347254157	11.2530117	-43.133323	-10.690600	-6.7005376	-5.717237
Complexe-11	-5.67867804	3.40616655	10.1097002	-44.592021	-11.159195	-14.990188	-5.67867804

3.1Å and 3.55Å are average. Greater than 3.55Å interactions are weak (Ritchie and Kemp, 2000).

#### Docking interpretation of Synthetic inhibitors without water

Results given in Table 6 (Fig. 3a,b) show that the complex-5 has the lowest energy (-5.99548626 Kcal/mol) and is more active than complex-3 (-5.81052923 Kcal/mol).

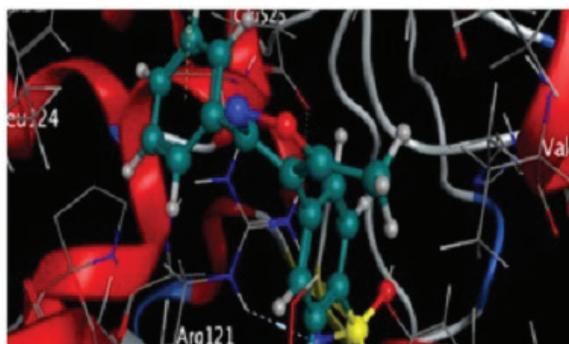
For complex 5: Valdecoxib interacts with the amino acids [ARG 121 (A) H-acceptor N6 (NE; NH<sub>2</sub>); ARG 121 (A) ionic [N6 (NE; NH<sub>2</sub>), (O3, NH<sub>2</sub>), and LYS 83 (A) pi-cation] at a distance of 3.13 Å, 3.01 Å, 3.58 Å, and 4.87 Å, respectively (for the 1st, 2nd strong interaction, 3th and 4th weak interaction), with the existence of electric force PRO 84 this suggests that Valdecoxib can inhibit COX-2 and interfere with [ARG 121 (A) H-acceptor N6 (NE; NH<sub>2</sub>); ARG 121 (A) ionic [N6 (NE; NH<sub>2</sub>), (O3, NH<sub>2</sub>) and LYS 83 (A) pi-cation] (Yamaguchi *et al.*, 2014).

For complex 3: Celecoxib interacts with the amino acids [ARG 121 (A) H-acceptor O5 (NE, NH<sub>2</sub>) ARG 121 (A) ionic O5

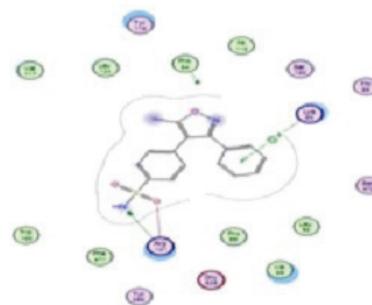
(NE, NH<sub>2</sub>), (O6, NH<sub>2</sub>); LYS 83 (A) pi-cation; TYR 116 (A) pi-H] at a distance of 2.87 Å, 2.79 Å, 3.89 Å, and 3.40 Å, respectively (for the 1st, 2nd strong interaction, 4th average interaction, and 3rd weak interaction), with the existence of three electric force PRO 84 wich suggesting that Celecoxib can inhibit COX-2 and interfere with [ARG 121 (A) H-acceptor O5 (NE, NH<sub>2</sub>) ARG 121 (A) ionic O5 (NE, NH<sub>2</sub>), (O6, NH<sub>2</sub>); LYS 83 (A) pi-cation; TYR 116 (A) pi-H] (Yamaguchi *et al.*, 2014).

#### Docking interpretation of Synthetic inhibitors with water

Our results given in Table 7 and Figures 4a and 4b show that the complex-2 presents the best score (-6.95738792 Kcal/mol) succeeded by complex-4 (-6.59314966 Kcal/mol). For this complex, Ibuprofen interacts with the amino acids [LYS 83 (A) (H-acceptor 83, (A) ionic)] at a distance of 2.84 Å (for the 1st, 2nd strong interaction) with the existence of electric force PRO 84 wich suggesting that Ibuprofen can inhibit COX-2 and interfere with LYS 83 (A) (H-acceptor 83, (A) ionic)] (Yamaguchi *et al.*, 2014).



**Figure 3(a).** Diagram interaction of complex-5 (COX-2 + Valdecoxib).



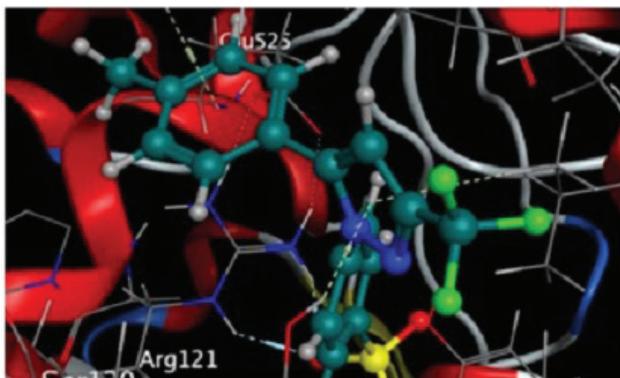
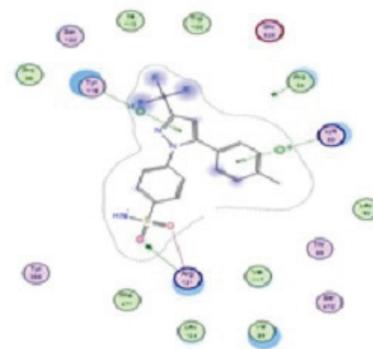


Figure 3(b). Diagram interaction of complex-3 (COX-2 + Celecoxib).



For complex 4: Rofecoxib interacts with the amino acid [LYS 83 (A) H-acceptor] at a distance of 2.68 Å (for the 1st) with the existence of electric force PRO 84 with suggesting that Rofecoxib can inhibit COX-2 and interfere with [LYS 83 (A) H-acceptor] (Yamaguchi *et al.*, 2014).

#### Docking interpretation of natural inhibitors without water

Obtained results (Table 8 and Fig. 5a,b) show that the complex-8 has the lowest energy (-6.11540222 Kcal/mol) and is more active than complex-7 (-6.05466223 Kcal/mol).

The energy of the reference ligand is important in comparison with that obtained by the Naringenine natural ligand. Therefore, we can validate Naringenine as a reference inhibitor.

Indeed the corresponding complex energies are successively (ref: -5.98953867 Kcal/mol and Naringenine: -6.11540222 Kcal/mol).

In interaction between enzyme COX-2 and Naringenine, we did not find any bonding, only possible forces are electric (PRO 84 and LYS 83) with the existence of the Van der Walls forces, but the total energy of complex is very low comparing to other ligands in interactions.

For the interaction of Luteoline with COX-2, we get one bonde between PRO 86 (A) H-donor (O4, O) with the length of 3.08 Å; it is a strong interaction but two electric forces with (PRO 84 and LYS 83) with the existence of the Van der Walls forces.

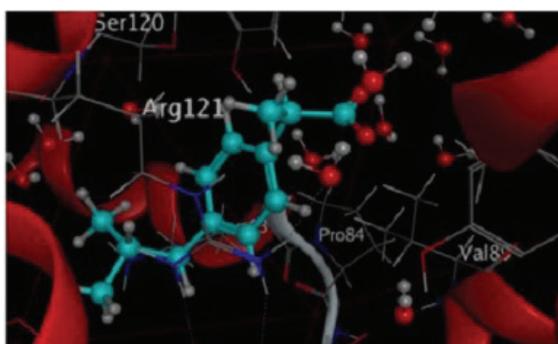


Figure 4(a). Diagram interaction of complex-2 (COX-2 + Ibuprofen).

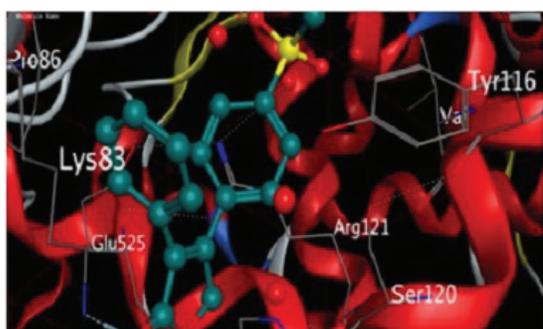
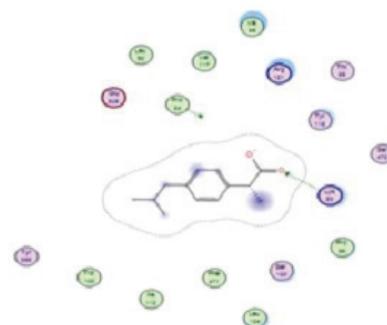
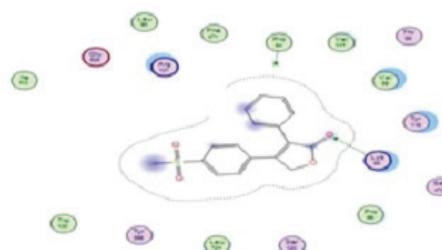


Figure 4(b). Diagram interaction of complex-4 (COX-2 + Rofecoxib).



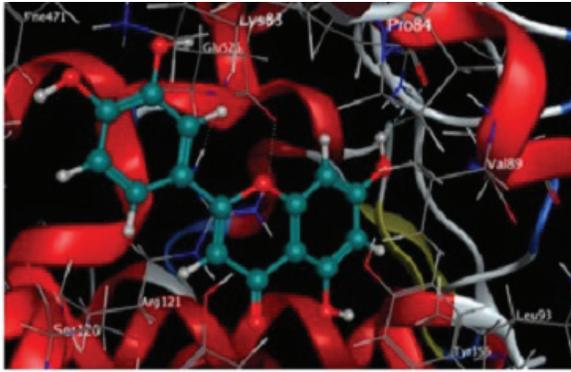


Figure 5(a). Diagram interaction of complex-8 (COX-2 + Naringenine).

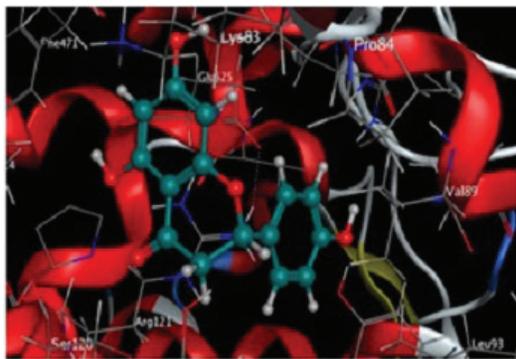
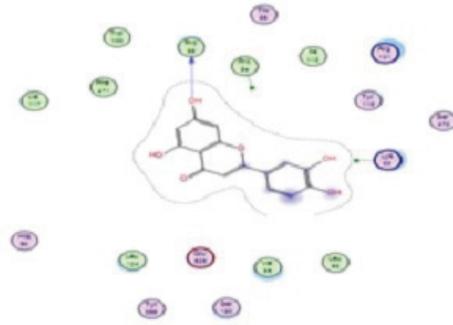
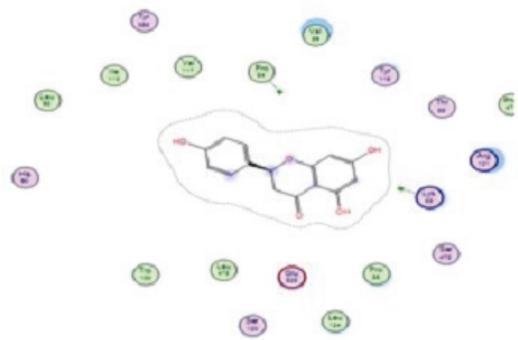


Figure 5(b). Diagram interaction of complex-7 (COX-2 + Luteoline).



### Docking interpretation of natural inhibitors with water

Table 9 and Figures 6a,b show that the complex-9 has the lowest energy ( $-9.11451435$  Kcal/mol) and is more active than complex-8 ( $-6.10765457$  Kcal/mol).

On the other hand, the reference ligand complex energy is greater comparing with that obtained for the natural ligand Naringenine. Therefore, we can validate Chlorogenique as a reference inhibitor. Complex energies (ref:  $-5.98953867$  Kcal/mol and Chlorogenique:  $-9.11451435$  Kcal/mol).

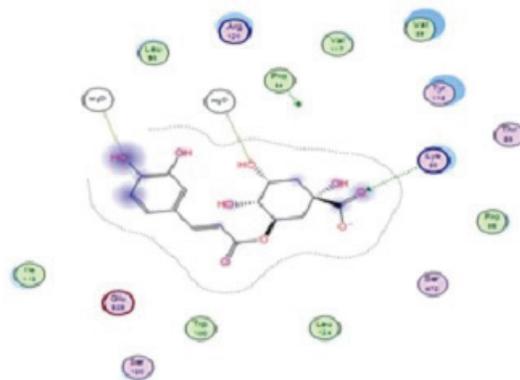
For complex 9: Chlorogenique interacts with the amino acids [LYS 83 (A) (H-acceptor 83, (A) ionic)] at a distance of

$3.04 \text{ \AA}$  (for the 1st strong interaction) and interaction with water HOH 0 (N4; N5) H-donor at a distance of  $3.01, 2.71 \text{ \AA}$  (for the 1st and 2nd strong interaction) with the existence of electric force PRO 84 wich suggesting that Chlorogenique can inhibit COX-2 and interfere with LYS 83 (A) (H-acceptor 83, (A) ionic)] (Yamaguchi *et al.*, 2014).

For complex 8: Naringenine interacts with the amino acid [PRO 86 (A) H-acceptor] at a distance of  $3.30 \text{ \AA}$  (for the 1st average interaction with the existence of electric forces PRO 84 and PRO 86 wich suggesting that Naringenine can inhibit cyclooxygenase-2 and interfere with [PRO 86 (A) H-acceptor] (Yamaguchi *et al.*, 2014). We can conclude from our obtained results



Figure 6(a). Diagram interaction of complex-9 (COX-2 + Chlorogenique).



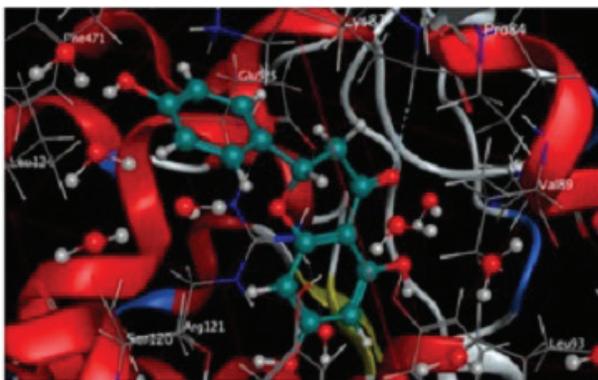
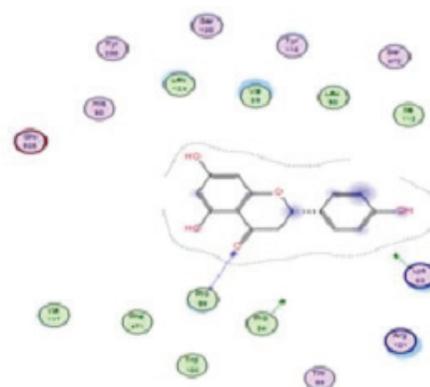


Figure 6(b). Diagram interaction of complex-8 (COX-2 + Naringenine).



that Ibuprofen and Chlorogenic acid would be the best to slow down the evolution of the treatment ID. This is confirmed by comparing their energies: Energy [Ibuprofen ( $-6.95738792$  Kcal/mol) < Rofecoxib ( $-6.59314966$  Kcal/mol). Energy (Chlorogenic acid ( $-9.11451435$  Kcal/mol) < Naringenine ( $-6.10765457$  Kcal/mol)].

## CONCLUSION

In this work, we studied the interaction of cyclooxygenase-2 (inflammation enzyme) by molecular docking taking into account solvation parameter (presence of water molecules). Obtained results allow us to conclude that the synthetic NSAID (Ibuprofen) and also the natural flavonoid inhibitor (chlorogenic acid) present a more optimized interaction for better inhibition study of COX-2 in purpose to treat ID. Obtained results allow us to propose a natural and reliable treatment with natural products containing Chlorogenic acid during the first stage of the inflammatory disease. We also propose further studies to develop chlorogenic acid into a new drug.

## CONFLICT OF INTEREST

The authors declare no conflict of interest.

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