

A Novel Approach of Potent Antioxidant and Antimicrobial Agents Containing Coumarin Moiety Accompanied with Cytotoxicity Studies on the Newly Synthesized Derivatives

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ABSTRACT

3-Amino-6-(2-oxo-2H-chromen-3-yl)pyridine-2-carbonitrile (**2**) was synthesized *via* the reaction of 3-(dimethylamino)acryloyl-2H-chromen-2-one (**1**) with malononitrile. The reactivity of the 2-amino-3-cyanopyridine derivative **2** toward a variety of electrophiles as triethylorthoformate followed by nitrogenous nucleophiles as hydrazine hydrate was investigated. Also, enamine **1** was utilized as precursor for synthesis of new 3-heteroarylsubstituted coumarins, in which it reacted with a series of hydrazonoyl halides and *p*-bromobenzenediazonium chloride to yield different 3-substituted coumarins. The structures of the newly synthesized derivatives were confirmed by different spectral tools (IR, ¹H-NMR, mass spectrometry and elemental analysis). All the newly synthesized compounds were screened for their antioxidant, antimicrobial activities and cytotoxicity. The preliminary structure-activity relationship was discussed to illustrate the essential structure requirements.

INTRODUCTION

Coumarins are an extremely important family of heterocyclic compounds owing to their presence in large biologically active substances and are used as pharmaceutical agents. Coumarins are natural extract products; it is the secondary metabolites in plant species, especially in the plants belong to families rutaceae and umbrelliferae (Dean, 1952). Coumarins are classified into simple coumarins, such as pyran and furanocoumarins and other coumarins that can be divided into linear and angular types according to the different substituent group in the ring. Natural and synthetic derivatives of coumarins have wide range of biological and pharmacological

activities such as, as anti-oxidant, anti-HIV, anti-bacterial (Shwu-Chen *et al.*, 2016), antifungals (Ping-Ping *et al.*, 2017), anticancer (Salem *et al.*, 2016), and anti-inflammatory (Witaicenis *et al.*, 2014). Alzheimer's disease, atherosclerosis, osteoarthritis and other diseases as well as many types of cancer diseases (Medzhitov, 2010) have been attributed to inflammatory processes in their pathogenesis. Coumarin derivative exerts its anti-inflammatory action in rat lung edema by inhibiting the TNF- α expression and increasing the VCAM-1 and ICAM-1 expression level (Li *et al.*, 2012). Recent studies have also been found that coumarins possess anti-leishmanial activity (Mislaine *et al.*, 2017). Substituted coumarins have potential pharmaceutical activities as antidiabetic activity (Sung *et al.*, 2017) and they also used as additives to food and cosmetics (Kennedy and Thornes, 1997). These investigations have revealed their potentials as antioxidant, antimicrobial and cytotoxicity. This investigation has been undertaken with the aim of synthesizing novel Coumarin derivatives and evaluating their biological activities.

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EXPERIMENTAL

All melting points were determined on an electrothermal apparatus and are uncorrected. IR spectra were recorded (KBr discs) on Shimadzu FT-IR 8201 PC spectrophotometer. ^1H NMR spectra were recorded in $(\text{CD}_3)_2\text{SO}$ solutions on JNM-LA 400 FT-NMR system spectrometer and chemical shifts are expressed in δ ppm units using TMS as an internal reference. Mass spectra were recorded on a GC-MS QP1000 EX Shimadzu. Elemental analyses were carried out at the microanalytical center of Cairo University. Hydrazonoyl halides (Shawli and Abdelhamid, 1976) were prepared as previously mentioned.

Chemistry

2-amino-6-(2-oxo-2H-chromen-3-yl)pyridine-3-carbonitrile (2).

A mixture of 3-(3-(dimethylamino)acryloyl)-2H-chromen-2-one (**1**) (1.2 gm, 5mmol) and malononitrile (0.33 gm, 5mmol) was heated in acetic acid (10 ml) under reflux in the presence of catalytic amount of ammonium acetate (0.38gm, 5mmol) for 3hrs. on cooling, the separated solid was filtered, washed with water and crystallized from acetic acid to afford **2** as orange crystals in a good yield. Yield: 82%; mp.: 208-210°C; FT-IR (KBr, cm^{-1}): 3409, 3330 $\nu(\text{NH}_2)$, 2208 $\nu(\text{CN})$, 1677 $\nu(\text{C}=\text{O})$, 1606 $\nu(\text{C}=\text{N})$; ^1H NMR (400MHz, $\text{DMSO}-d_6$): δ = 7.3-7.8 (m, 6H, Ar-H), 8.51 (s, 1H, C4-H of coumarin), 6.5 (s, 2H, NH_2); MS: M/z [%] = 264 (M+1, 4%), 263 (M^+ , 17%), 216 (8.08%), 174 (20%), 145 (43.99%), 110 (1.63%), 99 (100%), 77 (75.23%); Anal. calcd. for $\text{C}_{15}\text{H}_9\text{N}_3\text{O}_2$ (263): C, 68.44; H, 3.45; N, 15.96%. Found: C, 68.45; H, 3.42; N, 16.93%.

Ethyl N-(3-cyano-6-(2-oxo-2H-chromen-3-yl)pyridine-2-yl)carboximidate (3)

A mixture of (**2**) (2.3gm, 5mmol) and triethylorthoformate (1.48gm, 10mmol) in acetic anhydride (20ml) was heated under reflux for 6hrs. the reaction mixture was poured onto ice (30gm). The resulting solid was collected and recrystallized from ethanol to give **3** as brown crystals. Yield: 72%; mp.: 172-174°C; FT-IR (KBr, cm^{-1}): 2212 $\nu(\text{CN})$, 1680 $\nu(\text{C}=\text{O})$, 1620 $\nu(\text{C}=\text{N})$, 1604 (C=C); ^1H NMR (400MHz, $\text{DMSO}-d_6$): δ = 1.2-1.37 (t, 3H, CH_2CH_3); 4.1-4.23 (q, 2H, CH_2CH_3); 6.8-8.15 (m, 7H, Ar-H, CH=N); 8.51 (s, 1H, C4-H of coumarin); MS: M/z [%] = 321 (M+2, 2.6%), 319 (M+, 37%), 304 (88%), 290 (26%), 194 (30%) Anal. calcd. for $\text{C}_{18}\text{H}_{13}\text{N}_3\text{O}_3$ (319): C, 67.71; H, 4.10; N, 13.16%. Found: C, 67.69; H, 4.07; N, 13.15%.

3-(4-aminopyrido[2,3-d]pyrimidin-7-yl)-2H-chromen-2-one (4).

A mixture of **2** (2.3gm, 5mmol) and formamide (10 ml) was boiled under reflux for 7hrs. the reaction mixture was poured onto ice (30gm). The resulting solid was collected and recrystallized from acetic acid to give brown crystals. Yield: 62%; mp.: >300°C; FT-IR (KBr, cm^{-1}): 3413 $\nu(\text{NH}_2)$, 1682 $\nu(\text{C}=\text{O})$, 1630 $\nu(\text{C}=\text{N})$, 1604 (C=C); ^1H NMR (400MHz, $\text{DMSO}-d_6$): δ = 6.68 (s, 2H, NH_2); 7.5-8.11 (m, 7H, Ar-H); 8.50 (s, 1H, C4-H of coumarin); MS: M/z [%] = 290 (M^+ , 27%), 274 (8%), 190 (16%), 169 (31 %);

130 (83%); 77 (71%); 51 (65%); Anal. calcd. for $\text{C}_{16}\text{H}_{10}\text{N}_4\text{O}_2$ (**290**): C, 66.20; H, 3.47; N, 19.30%. Found: C, 66.22; H, 3.44; N, 19.28%.

3-(3-amino-4-imino-3H,4H-pyrido[2,3-d]pyrimidin-7-yl)-2H-chromen-2-one (5).

A mixture of **3** (1.6gm, 5mmol) and hydrazine hydrate (1ml, 20mmol) was boiled under reflux for 3hrs. in ethanol (10ml). The solid collected was crystallized from acetic acid to give **5** as white crystals. Yield: 73%; mp.: 242-244°C; FT-IR (KBr, cm^{-1}): 3403 ν (broad, NH , NH_2), 1678 $\nu(\text{C}=\text{O})$, 1620 $\nu(\text{C}=\text{N})$, 1604 (C=C); ^1H NMR (400MHz, $\text{DMSO}-d_6$): δ = 6.8 (s, 2H, NH_2); 7.5-7.91 (m, 6H, Ar-H); 8.50 (s, 1H, C4-H of coumarin); 9.52 (s, 1H, CH-pyrimidine); 11.0 (s, 1H, NH); MS: M/z [%] = 307 (M+2, 1.9%); 305 (M^+ , 17%); 289 (27%); 274 (38%); 195 (26%); 169 (31 %); 130 (13%); 77 (77%); 51 (67%); Anal. calcd. for $\text{C}_{16}\text{H}_{11}\text{N}_5\text{O}_2$ (305): C, 62.95; H, 3.63; N, 22.94%. Found: C, 62.92; H, 3.61; N, 22.93%.

$N^$ -(3-cyano-6-(2-oxo-2H-chromen-3-yl)pyridin-2-yl)- N , N -dimethylformamide (6).

A mixture of **2** (2.6gm, 10mmol) and DMF-DMA (11.9gm, 14ml, 10mmol) in dry xylene (20ml) was refluxed for 4hrs. the reaction mixture was cooled and the collected solid was washed with petroleum ether and recrystallized from Benzene to give **6** as Orange crystals. Yield: 94%; mp.: 180-182°C; FT-IR (KBr, cm^{-1}): 3350 $\nu(\text{CH}$ aliphatic), 2221 $\nu(\text{CN})$; 1682 $\nu(\text{C}=\text{O})$, 1617 $\nu(\text{C}=\text{N})$, 1604 (C=C); ^1H NMR (400MHz, $\text{DMSO}-d_6$): δ = 2.57 (s, 6H, 2 CH_3); 6.5 (s, 1H, CH=N); 7.5-8.12 (m, 6H, Ar-H); 8.51 (s, 1H, C4-H of coumarin); MS: M/z [%] = 320 (M+2, 2.2%); 318 (M^+ , 57%); 303 (2.9%); 288 (18%); 262 (16%); 195 (2%); 77 (70%); 51 (77%); Anal. calcd. for $\text{C}_{18}\text{H}_{14}\text{N}_4\text{O}_2$ (318): C, 67.91; H, 4.43; N, 17.60%. Found: C, 67.94; H, 4.41; N, 17.58%.

7-(2-oxo-2H-chromen-3-yl)pyrido[2,3-d]pyrimidin-4(3H)-one (10)

A mixture of **2** (2.3gm, 5mmol) and formic acid (10 ml) was boiled under reflux for 10hrs. the reaction mixture was poured onto ice (30gm). The resulting solid was collected and recrystallized from ethanol to give yellow crystals. Yield: 82%; mp.: 188-190°C, FT-IR (KBr, cm^{-1}): 3340 $\nu(\text{NH})$, 1675, 1690 $\nu(\text{C}=\text{O})$, 1630 $\nu(\text{C}=\text{N})$, 1614 (C=C); ^1H NMR (400MHz, $\text{DMSO}-d_6$): δ = 7.3-7.52 (m, 6H, Ar-H); 8.52 (s, 1H, C4-H of coumarin); 9.29 (s, 1H, pyrimidine); 11.12 (s, 1H, NH); MS: M/z [%] = 292 (M+1, 17.2%); 291 (M^+ , 18%), 169 (3.1 %); 130 (83%); 77 (51%); 51 (55%); Anal. calcd. for $\text{C}_{16}\text{H}_9\text{N}_3\text{O}_3$ (291): C, 65.98; H, 3.11; N, 14.43%. Found: C, 65.97; H, 3.08; N, 14.40%.

3-(4-chloropyrido[2,3-d]pyrimidin-7-yl)-2H-chromen-2-one (13).

A mixture of **10** (2.9gm, 10mmol) and phosphorus oxychloride (20ml) was heated under reflux for 5hrs. the reaction mixture was cooled and poured onto ice (30gm). The resulting solid was collected and recrystallized from acetic acid to afford **13** as white crystals in a good yield. Yield: 74%; mp.: 244-246°C, FT-IR (KBr, cm^{-1}): 1687 $\nu(\text{C}=\text{O})$, 1620 $\nu(\text{C}=\text{N})$, 1604 (C=C); ^1H NMR (400MHz, $\text{DMSO}-d_6$): δ = 7.32-7.61 (m, 6H, Ar-H);

8.52(s, 1H, C4-H of coumarin); 9.5(s, 1H, pyrimidine); MS: M/z[%]= 311(M+2, 32%); 309(M⁺,29.7%); 274(17%); 169(52%); 130(13%); 77(70%); 51(15%); Anal. calcd. for C₁₆H₈ClN₃O₂ (309): C,62.05; H,2.60; N,13.57%. Found: C,62.07; H, 2.58; N,13.54%.

3-(4-hydrazinylpyrido[2,3-d]pyrimidin-7-yl)-2H-chromen-2-one(14)

A mixture of **13** (1.5gm, 5mmol) and hydrazine hydrate (1gm, 20mmol) in ethanol (10ml) was heated under reflux for 3hrs. the resulting solid was collected and recrystallized from acetic acid to give **14** as yellow crystals. Yield: 67%; mp. : 281-283°C; FT-IR(KBr,cm⁻¹): 3409 ν(NH,NH₂); 1687 ν(C=O), 1630 ν(C=N), 1617(C=C); H¹NMR (400MHz, DMSO-*d*₆): δ = 6.8 (s, 2H, NH₂); 7.15-7.82(m, 6H, Ar-H); 8.50(s, 1H, C4-H of coumarin); 9.51(s, 1H, pyrimidine); 11.02(s, 1H, NH); MS: M/z[%]= 307 (M+2, 17%); 305(M⁺,29.7%); 289(0.17%); 274(7.37); 186 (100%); 174(14%); 146(52%); 97(18%); 77(78%); 50(54%); Anal. calcd. for C₁₆H₁₁N₅O₂ (305): C,62.95; H,3.63; N,22.94%. Found: C,62.93; H, 3.60; N,22.92%.

3-(3,4,5,6,7,8,10-hexaazacyclopenta[*a*]naphthalene-11-yl)-2H-chromen-2-one(15)

Saturated solution of sodium nitrite (10ml) was added while stirring to a cold solution at 0-5°C of **14** (1.7gm, 5mmol) in acetic acid (30ml). The reaction mixture was stirred for 1hr. the resulting solid was collected and recrystallized from acetic acid to give compound **15** as brown crystals. Yield: 85%; mp.:>300°C; FT-IR(KBr,cm⁻¹):1690ν(C=O), 1636ν(C=N), 1607(C=C); H¹NMR (400MHz, DMSO-*d*₆): δ =7.35-7.9(m, 6H, Ar-H); 8.52(s, 1H, C4-H of coumarin); 9.25(s, 1H, pyrimidine); MS: M/z[%]= 318(M+2,1.8%); 316(M⁺,19.5%); 170(57.6%); 149(18%); 131(8.97%); 91(41%); 77(100%); Anal. calcd. for C₁₆H₈N₆O₂ (316): C,60.76; H,2.55; N,26.57%. Found: C,60.74; H, 2.53; N,26.55%.

2-(2-(4-bromophenyl)hydrazine-1-ylidene)-3-oxo-3-(2-oxo-2H-chromen-3-yl)propanal(16)

A solution of *p*-bromobenzenediazonium chloride (5mmol) was added to a mixture of **1** (5mmol), sodium acetate (0.65gm, 5mmol) in ethanol (30ml) at 0-5°C while stirring. The stirring was continued for 3hrs. the resulting solid was collected, washed with water and recrystallized to give **16** as reddish brown crystals from acetic acid. Yield: 87%; mp. : 284-286°C, FT-IR(KBr,cm⁻¹): 3352ν(NH); 3060,2920ν(CH-aliphatic); 2866, 2754ν(CHO); 1715ν(C=O,CHO); 1678ν(broad, C=O), 1636 ν(C=N), 1604(C=C); H¹NMR (400MHz, DMSO-*d*₆): δ =7.47-7.79(m, 8H, Ar-H); 8.52(s, 1H, C4-H of coumarin); 8.81(s, 1H, NH); 13.59(s,1H, CHO); MS: M/z[%]= 398(M+1,17%); 397(M⁺,28%); 328(19.5%); 224(15.6%); 147(26.7%); 85(28.9%); Anal. calcd. for C₁₈H₁₁BrN₂O₄ (397): C,54.16; H,2.78; N,7.02%. Found: C,54.17; H, 2.76; N,6.99%.

3-(3-(2-(4-bromophenyl)hydrazine-1-ylidene)-3H-1,5-benzodiazepin-2-yl)-2H-chromen-2-one(17), 4-(4-(2-(4-bromophenyl)diazen-1-yl)-3-(2-oxo-2H-chromen-3-yl)-1H-pyrazol-1-yl)phenyl azinic acid(18a), 2-(4-(2-(4-bromophenyl)diazen-1-yl)-3-(2-oxo-2H-chromen-3-yl)-1 H-pyrazol-1-yl)-5-(hydroxynitroso)phenyl)azinic acid(18b).

General Method

A mixture of **16** (5mmol) and the appropriate of *o*-phenylenediamine, 4-nitrophenylhydrazine or 2,4-dinitrophenylhydrazine (5mmol) in ethanol(15ml) was refluxed for 2hrs. the resulting solid was collected and recrystallized from the proper solvent to give **17,18a,b** respectively.

3-(3-(2-(4-bromophenyl)hydrazine-1-ylidene)-3H-1,5-benzodiazepin-2-yl)-2H-chromen-2-one(17)

Yield: 92%; mp. : 141-143°C,FT-IR(KBr,cm⁻¹): 3352ν(NH); 1680 ν(C=O), 1616 ν(C=N), 1600(C=C); H¹NMR (400MHz, DMSO-*d*₆): δ =6.85-7.66(m, 13H, Ar-H); 8.52(s, 1H, C4-H of coumarin); 11.23(s, 1H, NH); MS: M/z[%]= 472(M+2,87%); 470(M⁺,20%); 356(50%); 324(16%); 227(16%); 147(29%); 77(90.23%); Anal. calcd. for C₂₄H₁₅BrN₄O₂ (470): C,61.16; H,3.21; N,11.89%. Found: C,61.18; H, 3.19; N,11.87%.

4-(4-(2-(4-bromophenyl)diazen-1-yl)-3-(2-oxo-2H-chromen-3-yl)-1H-pyrazol-1-yl)phenyl)azinic acid(18a).

Yield: 78%; mp.: 155-157°C,FT-IR(KBr,cm⁻¹): 3305ν(CH); 1694ν(C=O), 1616 ν(C=N), 1596(C=C); H¹NMR (400MHz, DMSO-*d*₆): δ =7.18-7.8(m, 13H, Ar-H, CH-pyrazole); 8.50(s, 1H, C4-H of coumarin); MS: M/z[%]= 517(M+4%); 515(M⁺,40%); 470(15%); 340(50%); 314(10%); 227(12%); 145(9%); 77(53%); 50(19%); Anal. calcd. for C₂₄H₁₄BrN₅O₄ (515): C,55.83; H,2.73; N,13.56%. Found: C,55.85; H, 2.71; N,13.53%.

2-(4-(2-(4-bromophenyl)diazen-1-yl)-3-(2-oxo-2H-chromen-3-yl)-1H-pyrazol-1-yl)-5-(hydroxynitroso)phenyl)azinic acid(18b).

Yield: 84%; mp.:164-166°C,FT-IR(KBr,cm⁻¹): 1697 ν(C=O), 1619ν(C=N), 1596(C=C); H¹NMR (400MHz, DMSO-*d*₆): δ =7.18-8.12(m, 12H, Ar-H, CH-pyrazole); 8.50(s, 1H, C4-H of coumarin); MS: M/z[%]= 560(M⁺,5%); 544(45%); 528(9%); 514(6%); 446(50%); 320(12%); 127(8%); 150(19%); 77(90.2%); 50(34%)Anal. calcd. for C₂₄H₁₃BrN₆O₆ (560): C,51.36; H, 2.33; N,14.97%. Found: C,51.34; H, 2.30; N,14.95%.

Ethyl -1-(4-chlorophenyl)-4-(2-oxo-2H-chromen-3-carbonyl)-1H-pyrazole-3-carboxylate(20a), Ethyl -1-(4-nitrophenyl)-4-(2-oxo-2H-chromen-3-carbonyl)-1H-pyrazole-3-carboxylate (20b),3-(3-acetyl-1-(4-chlorophenyl)-1H-pyrazole-4-carbonyl)-2H-chromen-2-one(20c), 3-(3-acetyl-1-(4-nitrophenyl)-1H-pyrazole-4-carbonyl)-2H-chromen-2-one(20d).

General procedures

A mixture of **1** (10mmol), the appropriate of hydrazonoyl halides (**19a-d**)(10mmol) and triethylamine (1.5ml, 10mmol) in dry benzene(20ml)was heated under reflux for 2hrs. the reaction mixture evaporated under reduced pressure and triturated with petroleum ether 40/60. The resulting solid was collected and recrystallized from the proper solvent to give pyrazoles**20a-d** in a good yield.

Ethyl -1-(4-chlorophenyl)-4-(2-oxo-2H-chromen-3-carbonyl)-1H-pyrazole-3-carboxylate (20a).

Beige crystals from ethanol.Yield: 92%; mp. : 180-182°C; FT-IR(KBr,cm⁻¹): 3062ν(CH), 1666,1725ν(C=O), 1630ν(C=N), 1595(C=C); H¹NMR (400MHz, DMSO-*d*6): δ =1.30(t,3H, CH₂CH₃); 4.5(q, 2H, CH₂CH₃), 7.5-8.2 (m, 9H, Ar-H, CH-pyrazole); 8.52(s, 1H, C4-H of coumarin); MS: M/z[%]= 424(M+2,14%); 422(M⁺,10%); 407(8%); 393(15%); 359(54%); 343(15%); 272(21%); 201(17%); 147(29%); 77(90%); 74(54%); Anal. calcd. for C₂₂H₁₅ClN₂O₅ (422): C,62.49; H, 3.58; N,6.63%. **Found:** C,62.47; H, 3.55; N,6.60%.

Ethyl -1-(4-nitrophenyl)-4-(2-oxo-2H-chromen-3-carbonyl)-1H-pyrazole-3-carboxylate (20b).

Beige crystals from ethanol. Yield: 89%; mp. : 149-151°C; FT-IR(KBr,cm⁻¹): 3030ν(CH), 1670,1720 ν(C=O), 1618ν(C=N), 1595(C=C); H¹NMR (400MHz, DMSO-*d*6): δ=1.32(t,3H, CH₂CH₃); 4.23(q, 2H, CH₂CH₃), 7.35-8.1 (m, 8H, Ar-H); 8.23(s, 1H,CH-pyrazole H-5); 8.50(s, 1H, C4-H of coumarin); Anal. calcd. for C₂₂H₁₅N₃O₇ (433): C,60.97; H, 3.49; N,9.70%. **Found:** C,60.96; H, 3.46; N,9.68%.

3-(3-acetyl-1-(4-chlorophenyl)-1H-pyrazole-4-carbonyl)-2H-chromen-2-one(20c).

Yellow crystals from Dioxane. Yield: 91%; mp.:162-164°C; FT-IR(KBr,cm⁻¹): 1666,1680 ν(broad, C=O), 1640ν(C=N), 1600(C=C); H¹NMR (400MHz, DMSO-*d*6): δ=2.5(s,3H, CH₃); 7.11-7.92 (m, 9H, Ar-H, CH-pyrazole H-5); 8.52(s, 1H, C4-H of coumarin); MS: M/z[%]= 393(M+1,60%); 392(M⁺,45%); 377(1.5%); 342(9%); 327(51%); 227(16%); 171(16%); 146(2.9%); 77(91%); Anal. calcd. for C₂₁H₁₃ClN₂O₄ (392): C, 64.21; H,3.34; N,7.13%. **Found:** C,64.23; H, 3.31; N,7.11%.

3-(3-acetyl-1-(4-nitrophenyl)-1H-pyrazole-4-carbonyl)-2H-chromen-2-one(20d).

Yellow crystals from ethanol. Yield: 89%; mp.: 171-173°C; FT-IR(KBr,cm⁻¹): 1666,1680 ν(broad, C=O), 1640ν(C=N), 1604(C=C); H¹NMR (400MHz, DMSO-*d*6): δ=2.53(s,3H, CH₃); 7.11-7.92 (m, 9H, Ar-H, CH-pyrazole H-5); 8.52(s, 1H, C4-H of coumarin); Anal. calcd. for C₂₁H₁₃N₃O₆ (403): C,62.53; H,3.25; N,10.42%. **Found:** C,62.55; H, 3.23; N,10.40%.

3-(2-(4-chlorophenyl)-7-hydroxy-2H-pyrazolo[3,4-*d*]pyridazin-4-yl)-2H-chromen-2-one(21a), 3-(2-(4-nitrophenyl)-7-hydroxy-

2H-pyrazolo[3,4-*d*]pyridazin-4-yl)-2H-chromen-2-one(21b), 3-(2-(4-chlorophenyl)-7-methyl-2H-pyrazolo[3,4-*d*] pyridazin-4-yl)-2H-chromen-2-one(21c), 3-(2-(4-nitrophenyl)-7-methyl-2H-pyrazolo[3,4-*d*]pyridazin-4-yl)-2H-chromen-2-one(21d).

General procedures

A mixture of the appropriate pyrazoles 20a-d (5mmol) and hydrazine hydrate (1gm,1ml,10mmol) in ethanol(10ml) was heated under reflux for 3hrs. the reaction mixture was cooled and the resulting solid was collected and recrystallized from the proper solvent to give pyrazolo[3,4-*d*]pyridazines **21a-d**, respectively.

3-(2-(4-chlorophenyl)-7-hydroxy-2H-pyrazolo[3,4-*d*]pyridazin-4-yl)-2H-chromen-2-one(21a).

White crystals from ethanol. Yield: 69%; mp. :>300°C,FT-IR(KBr,cm⁻¹): 3450ν(OH); 1680 ν(C=O), 1635ν(C=N), 1604(C=C); H¹NMR (400MHz, DMSO-*d*6): δ= 7.32-7.81 (m, 8H, Ar-H); 8.12(s, 1H, CH-pyrazole); 8.52(s, 1H, C4-H of coumarin); 9.42(s, 1H, OH); MS: M/z[%]= 391(M+1, 1%); 390(M⁺,14%); 374(3%); 201(50%); 178(16%); 165(16%); 143(29%); 77(90.23%); 50(18%); Anal. calcd. for C₂₀H₁₁ClN₄O₃(390): C,61.47; H,2.84; N,14.34%. **Found:** C,61.44; H, 2.82; N,14.33%.

3-(2-(4-nitrophenyl)-7-hydroxy-2H-pyrazolo[3,4-*d*]pyridazin-4-yl)-2H-chromen-2-one(21b).

White crystals from ethanol. Yield: 82%; mp. : 291-293°C; FT-IR(KBr,cm⁻¹): 3460ν(OH); 1686 ν(C=O), 1635ν(C=N), 1614(C=C); H¹NMR (400MHz, DMSO-*d*6): δ= 7.32-7.81 (m, 8H, Ar-H); 8.23(s, 1H, CH-pyrazole); 8.50(s, 1H, C4-H of coumarin); 9.51(s, 1H, OH); Anal. calcd. for C₂₀H₁₁N₅O₅ (401): C,59.85; H,2.76; N,17.45%. **Found:** C,59.83; H, 2.75; N,17.42%.

3-(2-(4-chlorophenyl)-7-methyl-2H-pyrazolo[3,4-*d*]pyridazin-4-yl)-2H-chromen-2-one(21c).

Beige crystals from ethanol. Yield: 85%; mp.:>300°C; FT-IR(KBr,cm⁻¹): 1690 ν(C=O), 1635ν(C=N), 1600(C=C); H¹NMR (400MHz, DMSO-*d*6): δ= 2.57(s,3H, CH₃); 7.12-8.14(m, 9H, Ar-H, CH-pyrazole); 8.52(s, 1H, C4-H of coumarin); MS: M/z[%]= 389(M+1,15%); 388(M⁺,27%); 373(10%); 338(12%); 327(16%); 227(16%); 91(2.9%); 77(93%); 50(5%); Anal. calcd. for C₂₁H₁₃ClN₄O₂(388): C,64.87; H,3.37; N,14.41%. **Found:** C,64.88; H, 3.34; N,14.39%.

3-(2-(4-nitrophenyl)-7-methyl-2H-pyrazolo[3,4-*d*]pyridazin-4-yl)-2H-chromen-2-one(21d).

Beige crystals from ethanol. Yield: 81%; mp.:>300°C; FT-IR(KBr,cm⁻¹): 1689 ν(C=O), 1640ν(C=N), 1600(C=C); H¹NMR (400MHz, DMSO-*d*6): δ= 2.57(s,3H, CH₃); 7.32-7.84(m, 8H, Ar-H),8.14(s, 1H, CH-pyrazole); 8.53(s, 1H, C4-H of coumarin); Anal. calcd. for C₂₁H₁₃N₅O₄(399): C,63.16; H,3.28; N,17.54%. **Found:** C,63.13; H, 3.26; N,17.53%.

PHARMACOLOGY

Antimicrobial activity

The samples were prepared by dissolving 2mg in 2ml of DMSO and 100 μ l (containing 100 μ g) was used in this test. The antimicrobial activity of different samples was investigated by the agar cup plate method. Four different test microbes namely: *Staphylococcus aureus* (G+ve), *Pseudomonas aeruginosa* (G-ve), *Candida albicans* (yeast) and *Aspergillus niger* (fungus) were used. Nutrient agar plates were heavily seeded uniformly with 1ml of 10⁵-10⁶ cells/ml in case of bacteria and yeast. A Czapek-Dox agar plate seeded by the fungus was used to evaluate the antifungal activities. Then a hole was made in media by gel cutter (Cork borer no.4) in sterile condition. Then one drop of melted agar was poured into hole and allowed to solidify to make a base layer. After that specific amount of culture filtrate (0.1 ml) was poured into the hole. Then plates were kept at low temperature (4 °C) for 2-4 hours to allow maximum diffusion. The plates were then incubated at 37 °C for 24 hours for bacteria and at 30 °C for 48 hours in upright position to allow maximum growth of the organisms. The antimicrobial activity of the test agent was determined by measuring the diameter of zone of inhibition expressed in millimeter (mm). The experiment was carried out more than once and mean of reading was recorded (Abdel-Aziz *et al.*, 2014; Barry, 1976).

Antioxidant screening assay using DPPH radical scavenging ability

2,2-diphenyl-1-picrylhydrazyl radical (DPPH)
Sigma-Aldrich

DPPH radical scavenging ability

Samples (100 μ l) from stock solution of concentration 1mg/ml in DMF were mixed with 900 μ l of 0.1mM DPPH solution in methanol. The mixture was shaken vigorously and allowed to reach a steady state at temperature 37^oc for 30 min. Decolorization of DPPH was determined by measuring the absorbance at 517 nm, and the DPPH radical scavenging was calculated according to the following equation:

$$\% \text{ scavenging rate} = (A_1 - A_2 / A_1) \times 100$$

Where A₁ was the absorbance of the DPPH solution without test sample and A₂ was the absorbance of DPPH with the test sample. Ascorbic acid was taken as the standard.

Cytotoxicity evaluation

Vero cells obtained from adult African green monkey kidney epithelial cells according to American Type Culture Collection (ATCC) and the European Collection of Animal Cell Cultures (ECACC) repositories No.CCL-81, passage number: 136, kindly obtained from VACSERA, Egypt. Vero cell line was grown in liquid growth medium of Dulbecco's modification of Eagle medium (Silva *et al.*, 2008). Vero cells were stored in liquid nitrogen vapor phase and cells usually take 2-3 passages to reach their regular growth rate after recovery from frozen state (Phelan,

2007). Dulbecco's modification of Eagle medium (MEM) with fetal bovine serum (FBS), Antibiotic (penicillin/streptomycin) fungizone solution (LONZA), 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid (HEPES) buffer 1M pH values ranging from 6.8 - 8.2 (LONZA), Earle's salt and L-glutamine, sterile polystyrene 75 cm² tissue culture flasks with vented caps, sterile serological pipettes and 70% ethanol solution, all to be used in cell propagation. Equipment and solutions coming into contact with cells have been sterile, and proper sterile techniques were used. i.e, 70% ethanol used for decontamination of laminar flow hood and objects brought into the hood, CO₂ Incubator interior, inverted microscope cleaning of objectives and stage daily. Cell culture incubations were in a 37 °C humidified incubator and solutions were warmed to 37 °C just before headway (Suffness *et al.*, 1990; Himmel, 2015).

In a laminar flow hood, Vero cell suspension from the cryovial was transferred into a 75 cm² tissue culture flasks containing 25mL growth medium (GM) of MEM supplemented with heat inactivated 10% FBS, 1% HEPES and 1% antibiotic, A cryovial of Vero cells containing 1x10⁶ cells/1ml was thawed by gently swirling in a 37 °C water bath for 2 minutes and added to incubated flask in 37 °C, Vero cells grown to a 90-95% confluency in a 75 cm² flask, Growth medium from confluent monolayer of Vero cells was removed and cells washed with 1x PBS twice, 5mL of 1X trypsin-EDTA added and cells were incubated at 37°C for 2-3 minutes, until cells detached from the flask, 50ml GM was added. Divide it into 2 flasks each containing 25ml. Flasks incubated at 37°C. Cells checked for confluency using inverted microscope. When cells reached a >90% confluent monolayer, cells were passaged at suitable concentration of cells needed (30.000cells/100ul/well).

Samples were diluted in 100 μ l Maintenance media (MM) as 2 % FBS, 1% HEPES, 1% antibiotic, and 96 % MEM Earle's and the used dilutions were 100 μ g, 50 μ g and 10 μ g per well in triples in 96 well plate. Cell then added in each well and mixed well to be sure of its homogeneity. Plate were covered with sealing and incubated at 37^oc overnight in Co₂ incubator, plate has been checked under inverted microscope and wells were washed with PBS twice to be ready for staining with crystal violet stain (0.5% crystal violet, 5% formalin, 50% ethanol, 0.85% NaCl, H₂O) 10 μ l for 10 minutes followed by washing with distal H₂O three times and let it to dry and measure the plate readings in ELISA plate reader at wave length 630 nm (Himmel, 2015; Ouattara *et al.*, 2011; Baudoin *et al.*, 2007).

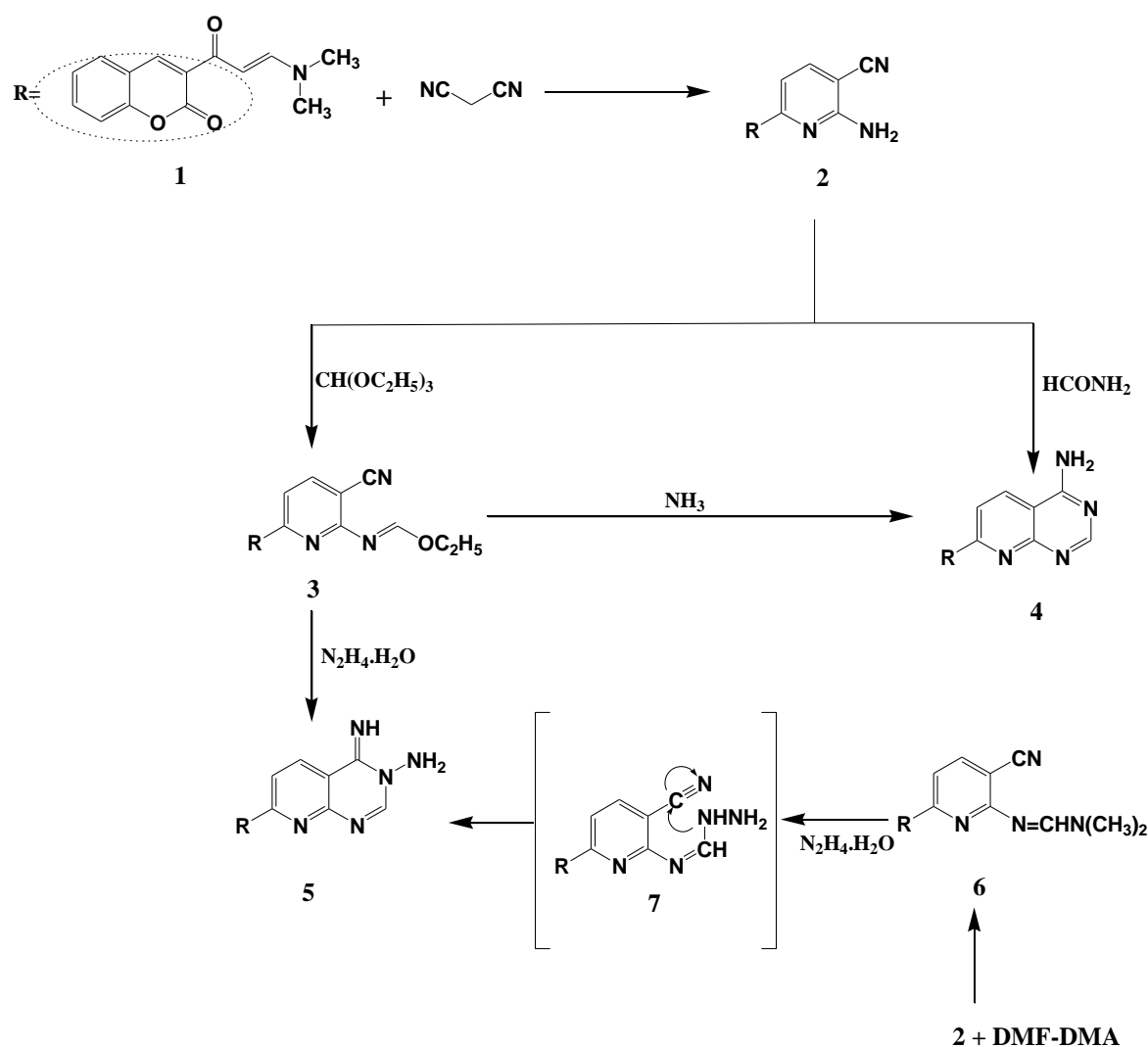
RESULTS AND DISCUSSION

Chemistry

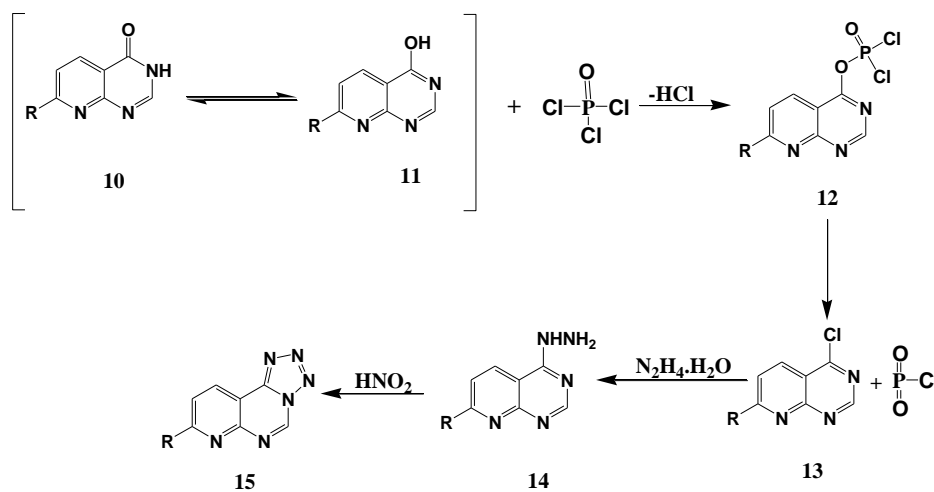
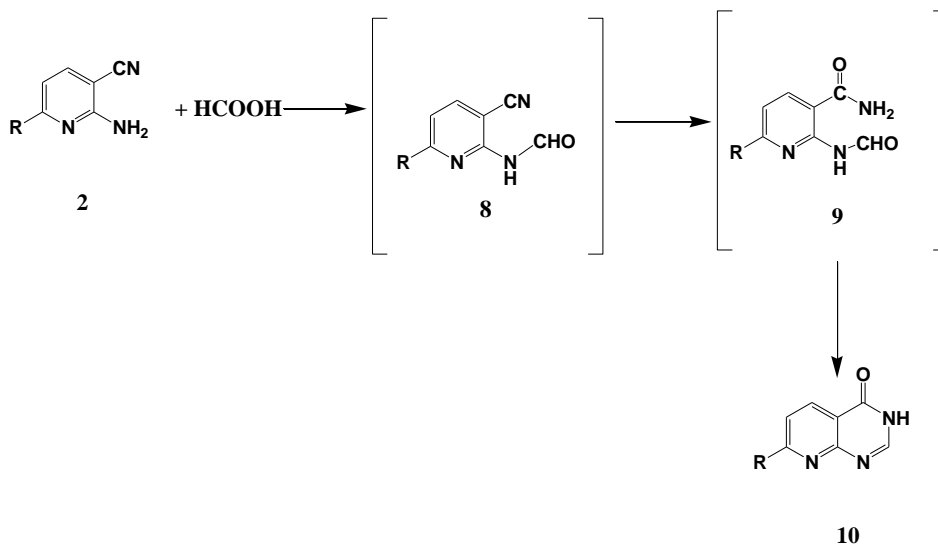
The synthetic procedures adopted to obtain the target compound **1** were previously reported (Abdel-Rahman *et al.*, 2003) to be synthesized through refluxing of 3-acetyl-2H-chromen-2-one with dimethylformamide-dimethyl acetal (DMF-DMA) (Al-Zaydi, 2003). The starting compound **2** was synthesized through the reaction of **1** with malononitrile in acetic acid in the presence of catalytic amount of ammonium acetate. The

newly synthesized aminocyanopyridine derivative **2** has now estimated as key molecule for synthesis of new 3-substituted coumarin heterocyclic compounds through the consideration reaction of compound **2** with triethylorthoformate and formamide to give compounds **3** and **4**, respectively. Structure of **3** was confirmed by spectral data (IR, ^1H NMR, mass spectrometry and elemental analysis). Its IR spectrum devoid any bands for NH_2 group which indicated its participation in the reaction and showed a strong absorption band at 2212 attributed to ν of CN group. Also, the ^1H NMR spectrum of compound **3** showed triplet signal at 1.2-1.37 ppm for three protons of methyl group of CH_2CH_3 group and

showed quartet signal at 4.1-4.23 ppm for two protons of CH_2 group of CH_2CH_3 . In addition, the structure of compound **3** was inferred chemically, in which it reacted with the nucleophile hydrazine hydrate to give the target molecule substituted 3-amino-4-imino-pyridopyrimidine **5**. Compound **5** can be obtained *via* alternative synthetic pathway, in which the reaction of **2** with dimethylformamide-dimethylacetal in dry dioxane give compound **6** which further reacted with hydrazine hydrate to give a product identical in all aspects (m.p. and mixed m.p., IR, ^1H NMR, mass spectrum and elemental analysis) with compound **5** that obtained from the previous method (Scheme 1).



Scheme 1



Cyclocondensation reaction of **2** with formic acid give pyrimidinone fused derivative **10** (Scheme 2). The reaction was carried out thermally in absence of solvent. The reaction proceeded through hydration of nitrile group to carboximide which then submitted to cyclization reaction to yield the compound **10**. The structure of **10** was investigated from its IR, ^1H NMR, mass spectrometry and microanalysis.

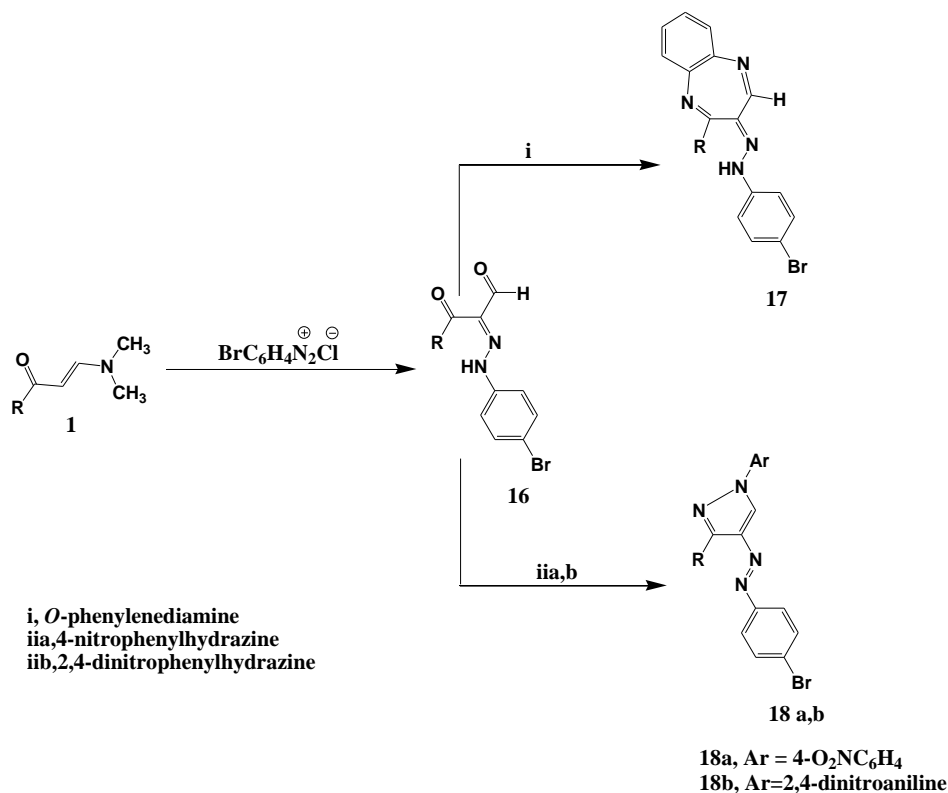
The spectral data revealed the disappearance of CN, NH_2 groups indicating their involvement in the cyclization process. The IR spectra of **10** showed also two strong absorption bands at the regions 1675, 1690 attributed to ν max of two carbonyl groups which indicated the formation of new carbonyl group due to the formation of pyrimidinone fused ring system in addition to the carbonyl group of coumarin ring system.

Formation of compound **10** is thought to take place through expected intermediates which are not isolated. Also, the ^1H NMR spectrum showed singlet signal at 9.29ppm. attributed to the proton of pyrimidinone ring (Scheme 2).

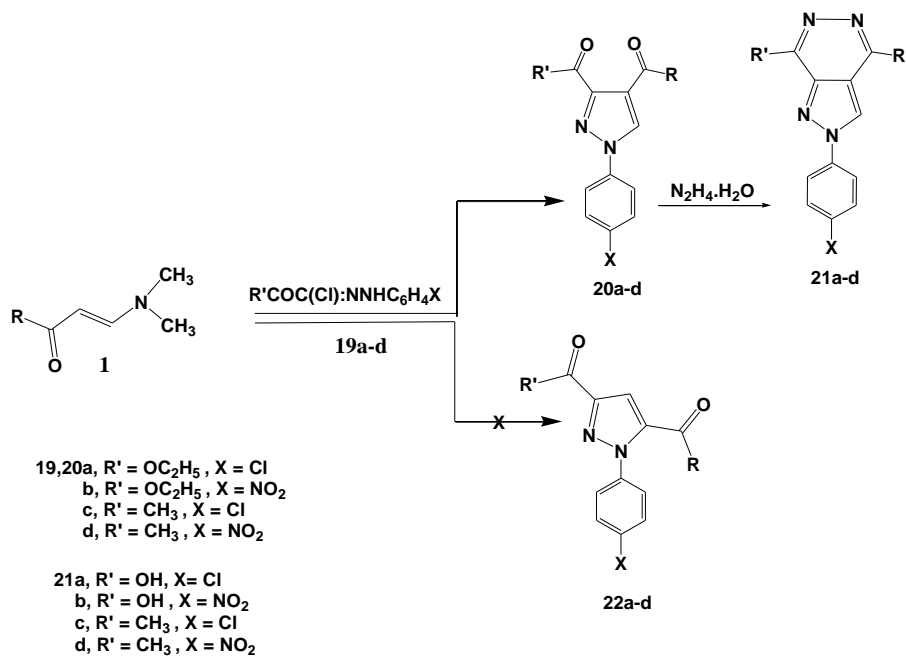
The structure of **10** was verified chemically through its reaction with phosphorus oxychloride to yield the corresponding chloro-derivative **13**, an internal nucleophilic substitution reaction by chloride ion on the carbon ion of lactim form **12** occurred ($\text{S}_{\text{N}}1$ mechanism). The structure of **13** was inferred from its spectral data in which its IR spectrum devoid any bands for $\text{C}=\text{O}$ group. Compound **13** was submitted to react with hydrazine hydrate to give the hydrazide **14**. The latter was converted to **15** via its reaction with nitrous acid at $0-5^\circ\text{C}$ through diazotization of the hydrazide derivative **14** (Scheme 3).

Treatment of enamine **1** with *p*-bromobenzenediazonium chloride in ethanol containing sodium acetate as a buffer solution yielded **16**. The structure of **16** was elucidated chemically *via* its reaction with *o*-phenylenediamine, 4-nitrophenylhydrazine and 2,4-dinitrophenylhydrazine to give the corresponding derivatives **17**, **18a** and **18b** respectively (Scheme 4). When enamine **1** was submitted to react with the appropriate hydrazoneyl halide

19a-d in dry benzene in the presence of a catalytic amount of triethylamine under reflux it gives the corresponding substituted pyrazoles **20a-d**. Structures of compounds **20a-d** were confirmed *via* chemical transformation in which they reacted with hydrazine hydrate to give the corresponding pyrazolo[3,4-*d*]pyridazines **21a-d**, respectively (Scheme 5) which indicated the formation of **20a-d** and not **22a-d**.



Scheme 4



Scheme 5

PHARMACOLOGY

Antimicrobial activity

Resistance to antibiotic is a developing and serious problem, has become an increasingly important and pressing global problem. In communities and hospitals around the world, the number of patients with antibiotic-resistant infections continues to rise. The reasons of antimicrobial resistance are multifactorial. Some of this is due to the overuse of antibiotics in human (Panlilio *et al.*, 1992) and hence microbes intrinsically resist antimicrobials or have acquired resistance mechanism to these substances (Hassan *et al.*, 1999; Lerner, 1998). So, there is an urgent demand for discover new antibiotics.

The newly synthesized 3-substituted coumarin derivatives were evaluated for their *in vitro* antimicrobial potentiality against four microbial strains, namely: *Staphylococcus aureus* (G+ve), *Pseudomonas aeruginosa* (G-ve), *Candida albicans* (yeast) and *Aspergillus niger* (fungus). Neomycin was used as standard antibacterial agent whereas Cycloheximide was used as anti-fungal agent to compare the effect of the tested compounds under the same condition. Results in Table (1) illustrate the antimicrobial activities of different coumarin derivatives. It has been found that compound **20b** have the strongest antibacterial activity against the G+ve bacterium *Staphylococcus aureus* (25mm) as compared to the standard drug (Neomycin) and the other tested compounds. On basis of structure activity relationship (SAR) this strong effect of compound **20b** may be due to the presence of ester and nitro groups attached to the pyrazole ring. Also, compounds **18b**, **20c** and **20d** exhibited high antibacterial activity against *S. aureus* (20, 18 and 17mm, respectively). Whereas, compounds **17**, **10**, **5** and **2** exhibited an equal clear zone values (15mm). The antibacterial activity of the tested compounds toward the G-ve test, *Pseudomonas aeruginosa* has been explored. The highest inhibitory effect was noticed with compounds **2**, **21d**, **10**, **18b**, **3** and **20b** (20, 20, 19, 19, 18 and 18mm, respectively). Moderate antibacterial activity against *P. aeruginosa* has been reported with the compounds **5**, **17**, **20c**, **18a**, **20d**, **14** and **21b** with clear zones of 17, 17, 15, 15, 15, 14 and 16mm respectively. Lower activity was reported with compound **2a** (12mm). Concerning the effect of the synthesized coumarin derivatives on the pathogenic yeast test strain *Candida albicans*, it has been found that compounds **18b**, **2**, **21d**, **3**, **5**, **10**, **17** and **20c** exhibited higher antimicrobial activity against *C. albicans* (20, 19, 19, 18, 18, 18, 18 and 18mm, respectively). Compound **18a** showed moderate activity (14mm), whereas compounds, **20d**, **13** and **14** showed lower activity (12, 13 and 13mm, respectively). The synthesized coumarin derivatives were also tested as antifungal agents using *Aspergillus niger*. Compounds **20b**, **3**, **20d**, **20c** and **18a** exhibited high antifungal activity (32, 25, 25, 25 and 20mm, respectively). Compounds **21c**, **21d** and **18b** exhibited moderate activity (16, 19 and 18mm, respectively). Compounds **10**, **5**, **13** and **17** were considered as the weak antifungal compounds (13, 12, 12 and 12mm, respectively). Compounds **4**, **15**, **16**, **21b** and **21c** did not

exhibit any antimicrobial activity against any tested pathogenic test microbes.

Table 1: Antimicrobial activity of the newly synthesized coumarin derivatives against different test microbes.

Sample	Clear zone (ϕmm)			
	<i>Aspergillus niger</i>	<i>Candida albicans</i>	<i>Pseudomonas aeruginosa</i>	<i>Staphylococcus aureus</i>
Neomycin 100µg	0	25	23	22
Cycloheximide 100 µg	35	0	0	0
2	0	19	20	15
3	25	18	18	13
4	0	0	0	0
5	12	18	17	15
6	0	0	0	0
10	13	18	19	15
13	12	13	0	12
14	0	13	14	13
15	0	0	0	0
16	0	0	0	0
17	12	18	17	15
18a	20	14	15	13
18b	18	20	19	20
20a	0	12	12	14
20b	32	0	18	25
20c	25	17	16	18
20d	25	14	15	17
21a	0	0	14	0
21b	0	0	0	0
21c	16	0	0	0
21d	19	19	20	14

Antioxidant screening assay

Free radicals of different forms are constantly generated for specific metabolic requirement and disposed by an efficient antioxidant network in the body. Free radicals are unstable, so when the generation of these species exceeds the levels of antioxidant mechanism, they can damage cells, proteins and other genetic materials, such as DNA, leads to oxidative damage of tissues and organs, thus leading to many health-related problems. Eventually, their effects lead to chronic diseases, especially degenerative diseases. Many coumarins have a great ability to scavenge reactive oxygen species (ROS)-free radicals, such as superoxide radicals, hydroxyl radicals or hypochlorous acid and to influence processes involving free radical-injury (Gutteridge, 1995).

The scavenging activity for DPPH free radicals was measured according to Zhao *et al.* (2006), with some modifications. Some of the newly synthesized three substituted coumarins showed antioxidant effect as shown in Table 2. Comparing with Ascorbic acid, compounds **3**, **10**, **14**, **20b**, **20c** and **21b** exhibited the strongest antioxidant efficacy in the tested compounds. While compounds **2**, **5**, **13**, **15**, **18a**, **18b**, **20a**, **21a** and **21c** showed a moderate antioxidant activity. The rest of the compounds showed weak antioxidant activity. From the structure activity relationship (SAR) it is noticeable that compound **3** have cyano group attached to the pyridine ring and compound **20b** have

nitro and ester groups attached to the pyrazole ring which may have a special ability to scavenge reactive oxygen species and hence protect the cells from DNA damage.

Table 2: Antioxidant activity assay.

Sample	Absorbance	%Inhibition
2	0.072	88.7
3	0.0365	94.3
4	0.3	53
5	0.17	73.4
6	0.435	31.92
10	0.049	92.33
13	0.16	75
14	0.063	90.14
15	0.114	82.2
16	0.45	29.6
17	0.588	8
18a	0.077	87.9
18b	0.09	85.92
20a	0.08	87.5
20b	0.035	94.52
20c	0.04	93.74
20d	0.33	48.36
21a	0.064	89.9
21b	0.04	93.74
21c	0.109	82.94
21d	0.323	49.5
Ascorbic acid	0.038	96

Cytotoxicity evaluation

Half maximal inhibitory concentration (IC_{50}) in pharmacological research is the factor used to evaluate antagonist drug potency and the effect of dose-response curve of different

concentrations of drug on mammalian cell line growth will postulate the IC_{50} (Cheng and Prusoff, 1973; Lazareno and Birdsall, 1993). According to the National Cancer Institute (NCI), the criteria and the conditions of cytotoxic activity for the crude extract is an IC_{50} values $\leq 20 \mu\text{g/ml}$, is considered to be potentially cytotoxic. The criteria used to categorize the cytotoxicity of isolated fractions against mammalian cell lines, based on U.S. National Cancer Institute (NCI) and Geran protocol, was as follows: $IC_{50} \leq 20 \mu\text{g/ml}$ = highly cytotoxic, IC_{50} 21-100 $\mu\text{g/ml}$ = moderately cytotoxic, IC_{50} 101-200 $\mu\text{g/ml}$ = weakly cytotoxic and $IC_{50} > 501 \mu\text{g/ml}$ = no cytotoxic (Geran, 1972 ; Boik, 2001). The synthesized new 3-substituted coumarin derivatives were evaluated for their *in vitro* cytotoxic potentiality against mammalian cell line. Three concentrations 100 $\mu\text{g/ml}$, 50 $\mu\text{g/ml}$ and 10 $\mu\text{g/ml}$ of newly synthesized coumarin derivatives were used to determine the cytotoxic effect on Vero cell line in compare with the control free of derivatives.

Referring to **Table (3)** it was found that compounds **2, 3, 6, 13, 14, 16, 17, 18b, 20c, 21a, 21b, 21c** showed moderately cytotoxic effect while compounds **4, 15, 18a, 20b** exhibited weakly cytotoxic activity. But compounds **5, 10, 20d** showed moderately to high cytotoxic effect and compounds **20a, 21d** showed highly cytotoxic effect.

From the structure activity relationship (SAR), it is remarkable that compound **20a** has chloro and ester groups attached to the pyrazole ring and compound **21d** has methyl and nitro groups attached to the pyridazine ring which may responsible for their high cytotoxic effect.

Table 3: Cytotoxicity evaluation using concentrations 100, 50, 10 μg respectively of the newly synthesized coumarin derivatives against Vero cell line and calculation the equivalent concentration to IC_{50} for each.

Sample	Mean conc. 100 μg	Mean conc. 50 μg	Mean conc. 10 μg	Equation	Mean control	control IC_{50}	Conc. equivalent to $IC_{50} \mu\text{g}$
2	0.142	0.269	0.429	$y = 313.54x - 34.459$	0.581	0.2905	56.46
3	0.495	0.516	0.683	$y = 410.06x - 101.21$	0.689	0.3445	40.05
4	0.967	0.564	0.412	$y = -149.27x + 150.01$	0.642	0.321	102
5	0.451	0.228	0.645	$y = 112.5x + 3.6851$	0.492	0.246	31.36
6	0.671	0.186	0.685	$y = 14.058x + 46.108$	0.693	0.3465	50.97
10	0.21	0.735	0.464	$y = 73.287x + 18.913$	0.546	0.273	38.9
13	0.51	0.888	0.482	$y = -24.959x + 68.974$	0.473	0.2365	63.1
14	0.66	0.366	0.607	$y = -30.409x + 69.886$	0.453	0.2265	62.99
15	0.677	0.152	0.265	$y = -114.44x + 95.066$	0.547	0.2735	126.2
16	0.203	0.158	0.822	$y = 105.45x + 11.751$	0.682	0.341	47.7
17	0.463	0.52	0.08	$y = -157.74x + 109.23$	0.629	0.3145	59.699
18a	0.44	0.515	0.591	$y = 596.16x + 253.89$	0.544	0.272	416
18b	0.078	0.103	0.591	$y = 142.59x + 16.641$	0.538	0.269	54.95
20a	0.228	0.162	0.062	$y = -538.75x + 134.5$	0.562	0.281	-16.88
20b	2.341	1.641	0.953	$y = -64.819x + 159.96$	0.57	0.285	141.48
20c	0.623	0.591	0.128	$y = -149.82x + 120.35$	0.499	0.2495	82.99
20d	0.187	0.28	0.119	$y = -266.62x + 105.41$	0.477	0.2385	41.82
21a	1.18	1.7	0.897	$y = -45.032x + 110.03$	0.466	0.233	99.53
21b	1.582	0.271	0.354	$y = -49.113x + 89.464$	0.469	0.2345	77.94
21c	0.197	0.289	0.319	$y = 666.45x - 125.5$	0.531	0.2655	51.44
21d	0.085	0.075	0.065	$y = -4500x + 390.83$	0.668	0.334	-1112

(y) Is the concentration equivalent to IC_{50} per μg and (x) is the control IC_{50} Absorbance at wave length 630 nm.

CONCLUSION

The new compounds exhibited promising biological activities and seem to be interesting for further pharmaceutical studies. In addition, the present studies elucidate new methods for synthesis of new substituted coumarin derivatives. Most of the newly synthesized compounds showed strong efficacy against most of the tested microorganisms in case of compound **20b** the activity was stronger than the tested standard drug itself. Also, most of the new derivatives exhibited excellent results when they evaluated for their antioxidant activity like compounds **3**, **10**, **14**, **20b**, **20c** and **21b** which showed strong antioxidant activity when compared to the ascorbic acid. Moreover, compounds **5**, **10**, **20d** showed moderately to high cytotoxic effect and compounds **20a**, **21d** showed highly cytotoxic effect. So, synthesis of new coumarins and evaluation of their biological and pharmaceutical importance still an active area for more investigations.

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Conflict of Interests: There are no conflicts of interest.

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