

Production and characterization of polyhydroxybutyrate (PHB) produced by *Bacillus* sp. isolated from Egypt

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ABSTRACT

Synthetic polymers are non-degradable and accumulated in the environment so, the efforts of scientists were forwarded to provide us with alternative environmentally biopolymers. Polyhydroxyalkanoates (PHAs) including polyhydroxybutyrate (PHB) are Group of the interesting biopolymers which have several medical applications such as drug delivery, suture, scaffold and heart valves. PHAs are biological macromolecules, thermoplastics, biodegradable and biocompatible. In this study, new bacterial isolates from Egypt were screened for their ability to produce PHB using Nile red dye. Out of 44 isolates, 19 bacterial isolates were selected according to strong of their fluorescence on mineral salt medium (MSM) agar plates supplemented with Nile red. The most potent strain was identified using biochemical tests as *Bacillus* sp. N-2. Production of PHB was carried out in limitation of nitrogen source using a minimal salt medium (MSM) supplemented with an excess of glucose as sole carbon source. PHB was accumulated in relation to cell dry weight about 20% (PHB/CDW). The obtained biopolymer was purified and analyzed using NMR, FT-IR, TGA and DSC thus; it was highly pure and identified as PHB. Optimization of PHB production from cheap sources appears to be a realistic goal in the future for reducing the costs and obtaining high yield.

INTRODUCTION

The great use of petroleum plastics is the major cause of environmental pollution because the plastics need years for degrading to soluble monomers and many types of toxic gases are drawback during the decomposition process (Anderson and Dawes, 1990; Suriyamongkol *et al.*, 2007; Campos *et al.*, 2014). Consequently, there have been developing open and scientific concerns in regards to the utilization and development of biopolymer as a biologically valuable different option for plastics (Muller *et al.*, 2001; Amara *et al.*, 2011). Among the biodegradable polymers, polyhydroxyalkanoates (PHA) which fit

in with the gathering of polyesters has gotten vigorous consideration due to their features as biodegradable thermoplastics (Albuquerque *et al.*, 2007). PHAs are divided into three groups: short chain length, medium chain length and long chain length PHA (scl-PHAs, mcl-PHAs and lcl-PHAs respectively). These categories were classified according to the number of carbon atoms, where scl-PHA of 3–5 carbon atoms, mcl-PHA of 6–14 carbon atoms and lcl-PHA of more than 15 carbon atoms (Madison and Huisman, 1999). Many microorganisms have been discovered that be able to produce PHAs as insoluble granules intracellularly under stress conditions, in presence of excess carbon source and limited concentration of nitrogen, phosphorus or oxygen (Anderson and Dawes, 1990; Pan *et al.*, 2012). Several microorganisms have been utilized to produce different types of PHAs including Gram positive like *Bacillus megaterium*, *Bacillus subtilis* and *Corynebacterium glutamicum*, Gram negative bacteria like group of *Pseudomonas* and *Alcaligenes eutrophus* and cyanobacteria such as *Nostoc muscorum*, *Synechococcus* sp. and

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Spirulina platensis (Jo *et al.*, 2007; Hassan *et al.*, 2010; López *et al.*, 2012; Sharma and Mallick, 2005; Jau *et al.*, 2005). Based on microorganisms, the structure, physico-chemical properties, monomer composition, mass and size of PHAs differ considerably (Ha and Cho, 2002).

The most surely understood structure of PHAs is poly-3-hydroxybutyrate (PHB), a scl-PHA which can be synthesized by a wide range of microorganisms but the main interest group is *Bacillus* strains due to lack toxicity of the extracted PHB (Gahlawat *et al.*, 2012; Singh *et al.*, 2009). Amara *et al.* (2013) reported the interests of scientists to produce biopolymers and biotechnological products with the safe method to be suitable for different applications.

In addition to the biocompatibility of PHAs, they have properties like that of polypropylene, biodegradability, highly hydrophobic and thermoplastic, with high crystallinity, high melting temperature, and good resistance to organic solvents (Sudesh *et al.*, 2000). PHAs are of biological origin; they could be completely broken down in to water and carbon dioxide by microorganisms found in a wide range of environments (Byrom, 1987). Therefore, they have various medical applications including drug delivery, sutures and embed materials general (Deng *et al.*, 2003). In order to apply the biotechnological products, they should be examined on cell culture and animal model as shown by Abdou and Hassan (2014). The objective of the present study was to isolate a new bacterial isolate has the ability to produce PHB in a safe method for applying in medical applications. The PHB produced by *Bacillus* sp. was identified using ¹H nuclear magnetic resonance (¹HNMR) and Fourier transform infrared (FT-IR). In addition, the physical properties of PHB were investigated through Thermogravimetric analysis (TGA) and Differential scanning calorimetric (DSC) to select the potential applications in the future.

MATERIAL AND METHODS

Isolation of bacterial strains

Soil samples were collected from different sites of Alexandria and Assiut Governorates, Egypt. The samples were collected from depth (7-10 cm) of the soil surface in sterilized falcon tubes and transferred immediately to a microbiology lab.

The samples were suspended in sterilized saline solution (0.85%, w/v) and streaked on nutrient agar medium contained 5.0 g/l peptone, 3.0 g/l beef extract, 5.0 g/l sodium chloride and 15 g/l agar; pH was adjusted at 7.2±0.2). The plates were incubated at 37°C for 48 hrs then; the different colonies were picked and subjected to series steps for obtaining purified strain (Abol Fotouh *et al.*, 2016).

Screening of PHB producing bacteria

Nile red staining

The screening method which was described by Spiekermann *et al.* (1999) was conducted for detecting the ability of bacterial isolates to accumulate PHAs. The isolated strains

were grown on mineral salt medium (MSM) which was prepared according to Schlegel *et al.* (1961) and supplemented with 20 g/l glucose as carbon source (Aramvash *et al.*, 2015). MSM consisted of 9.0 g/l Na₂HPO₄·2H₂O, 1.5 g/l KH₂PO₄, 0.4 g/l NH₄Cl, 0.2 g/l MgSO₄·7H₂O, 0.02 g/l CaCl₂·2H₂O, 1.2 mg/l Fe(III)NH₄- citrate, 0.1 ml/l Trace elements solution 6 and 15 g/l agar. The trace element solution comprised of (per liter of distilled water): 10 mg/l ZnSO₄·7H₂O, 3 mg/l MnCl₂·4H₂O, 30 mg/l H₃BO₃, 20 mg/l CoCl₂·6H₂O, 1 mg/l CuCl₂·2H₂O, 2 mg/l NiCl₂·6H₂O and 3 mg/l Na₂MoO₄·2H₂O and it was sterilized with 0.22 µm sterilized filter system.

The glucose was autoclaved separately, allowed to cooling down to 55°C and it was added in a suitable final concentration to the prepared medium. In addition, MSM was supplemented with a solution of 0.25 mg Nile red in DMSO to give a final concentration of 0.5 µg dye (ml medium)⁻¹. The bacterial isolates were streaked and incubated at 37°C for 5 days. The agar plates were exposed to ultraviolet light (312 nm) using a gel documentation system (Model DI-C1-220, Major science Co., Taiwan) to investigate the accumulation of PHAs in positive isolates.

Strain identification

Morphological and biochemical test

The colony characteristics including the pigment of the potent bacterial isolate were observed. The ability of the bacterial isolate (N-2) to hydrolyze starch, gelatin, casein and cellulose was investigated. Other biochemical tests like methyl red, indole production test, Vogues Proskauer test, motility, catalase, citrate utilization were conducted in the defined medium using specific reagents.

The sugar fermentation was studied by inoculating the bacterial isolate in the defined medium including various sugars (xylose, mannose, sucrose, sorbitol, galactose, rhamnose, glucose, fructose, ribose, lactose and rhamnose). The morphological and physiological characteristics of the bacterial isolates were compared with the data from Bergey's Manual of Determinative Bacteriology (Holt *et al.*, 1994).

Production of PHB by *Bacillus* sp. N-2

Bacillus sp. N-2 was first grown on nutrient broth medium and incubated at 37°C and 150 rpm for 24 hrs to obtain a large amount of cells. Then, 2% (v/v) of the liquid culture was transferred to MSM with 2% glucose as the sole carbon source and incubated at 37°C and 150 rpm for 5 days in shaking incubator (Model Innova™ J-25, New Brunswick scientific, USA).

Extraction and purification of PHB

The purification of PHB was performed according to the method which was optimized by Hassan *et al.* (2010). Briefly, the bacterial cells were collected by centrifugation of culture at 11,000 rpm for 15 min.

The cells were washed twice in saline solution and dried overnight in an oven at 50°C for 24 hrs. The dried cells were

treated with 10% SDS at 100°C for 20 min. After centrifugation, the pellets were washed, dried, and extracted with chloroform at 60°C for 1 hr.

The non-PHB cell matter was removed by filtration using filter paper (what man no. 1) and the dissolved PHB was precipitated from chloroform by addition of 4-6 volume of ice cold methanol. The obtained PHB was left in -20 for 30 min and centrifuged at 11,000 rpm for 15 min to obtain a high yield of biopolymer. The obtained biopolymer was dried overnight at 60°C.

Characterization of PHB

NMR of the purified PHB from *Bacillus sp. N-2*

¹H NMR spectra was carried out by dissolving the purified PHB in deuteriochloroform (CDCl₃) at a concentration of 10 mg/ml and analyzed on a Bruker Avance II 500 spectrometer.

FT-IR analysis of PHB from *Bacillus sp. N-2*

FT-IR analysis of PHB produced by *Bacillus sp. N-2* was investigated by FT-IR (Model 8400 S, Shimadzu, Japan). Sample about 5 mg was mixed well with KBr, and the absorbance of sample was scanned from 400-4000 cm⁻¹ following the method which was described by Tamer *et al.* (2015).

TGA of PHB produced by *Bacillus sp. N-2*

Thermal analysis of Purified PHB sample was performed using TGA instrument (Model 50/50H, Shimadzu, Japan). The analysis method was carried out with 5 mg of biopolymer at a temperature range (27-450°C) under nitrogen atmosphere at a heating rate of 10°C min⁻¹.

DSC analysis of PHB produced by *Bacillus sp. N-2*

The Purified PHB was run in DSC using DSC instrument (Model 60A, Shimadzu, Japan) at a temperature range from -20 to 450°C with the heating rate of 10°C min⁻¹. Melting temperature was compared with standard PHB (Sigma, USA) from previous literature (Corre *et al.*, 2012).

RESULTS AND DISCUSSION

Isolation and screening of a bacterium producing PHA

After incubation of nutrient agar plates medium at 37°C for 48 hrs, a total of 44 different bacterial colonies were picked up as pure colonies and maintained on nutrient agar slants for further experiments. Total 29 isolates were obtained from Alexandria Governorate while 15 isolates were isolated from Assiut governorates.

Different isolates obtained were streaked on plates contained MSM agar medium supplemented with glucose (2%) and Nile red dye (0.5µg/ml). The plates were observed under transilluminator for fluorescent colonies thus, 19 isolates revealed fluorescence under UV light. Amongst 19 bacterial isolates, the potent isolate was obtained from Assiut and named N-2 was

selected based on maximum fluorescence as shown in figure 1. (Shrivastav *et al.*, 2010).

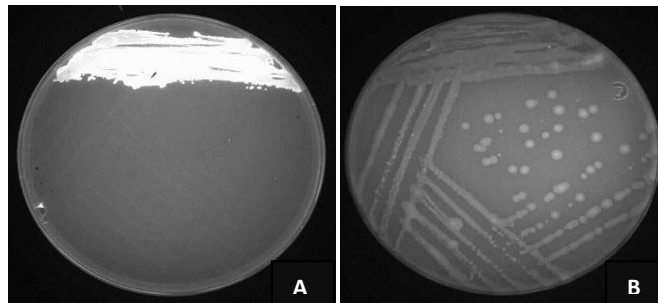


Fig. 1: Screening of bacterial isolates from soil using Nile red viable colony staining method. Figure (1A) indicates to positive isolate which accumulates PHB and shows with fluorescence under UV light; Figure (1B) indicates to negative bacterial isolate.

Strain identification

The morphological characteristics of strain N-2 were investigated. We found that strain N-2 is aerobic, Gram positive, motile, spore-forming, rod-shaped *Bacillus*. In addition, the biochemical features were examined and N-2 was able to ferment different sugars like glucose, galactose, sucrose, mannose, xylose, ribose, fructose, lactose, Sorbitol, Rhamnose and Rhaffinose.

Table 1: Morphological and biochemical features of *Bacillus sp. N-2*.

Morphological and biochemical tests	<i>Bacillus sp. N-2</i>
Morphological:	
Shape	Rods
Gram stain	G+ve
Motility	Motile
spore formation	+ve
Growth at different concentrations of NaCl:	
3%	+ve
5%	+ve
10%	+ve
Biochemical tests:	
Catalase	+ve
Voges- Proskauer	+ve
Inodol production	-ve
Citrate utilization	-ve
H ₂ S production	-ve
Hydrolysis of:	
Casein	+ve
Gelatin	+ve
Cellulose	+ve
Starch	-ve
Carbohydrates fermentation:	
Glucose	+ve
Galactose	+ve
Sucrose	+ve
Mannose	+ve
Xylose	+ve
Ribose	+ve
Fructose	+ve
Sorbitol	+ve
Lactose	-ve
Rhamnose	+ve
Rhaffinose	+ve

The culture exhibited no growth on starch and negative for H₂S production, while N-2 showed the ability to utilize casein, gelatin and cellulose. Moreover, the strain was positive for Voges Proskauer test, methyl red and catalase as well as, negative for

citrate utilization and indol production. The strain showed that it was able to grow on different concentration of NaCl (3%, 5% and 10%).

The morphological and physiological characteristics of N-2 were summarized as shown in table 1. The morphological and biochemical characteristics indicated that the bacterium N-2 belonged to *Bacillus* sp. group according to Bergey's Manual of Determinative Bacteriology (Holt *et al.*, 1994). However, the most results are similar to those which were obtained by Hassan *et al.* (2013) concerning the identification of *Bacillus* sp. strains so; it was named *Bacillus* sp. N-2.

Production and extraction of PHB

PHB was synthesized by *Bacillus* sp. N-2 as described above using accessing of carbon source (2% of glucose) and minimal concentration of nitrogen source (0.4 g/l NH₄Cl). After incubation for 5 days, the cells were harvested and the cell dry weight was 0.85 g/l. The PHB that accumulated in the cells was extracted using SDS and chloroform.

The amount of extracted PHB was recorded (0.17 g/l) and calculated in relation to cell dry weight as 20% CDW. Previous articles reported that the accumulated PHB in the cells from *B. megaterium* and *Massilia* sp. UMI-21 using glucose before optimizing the culture conditions were 21.7% and 2.4% CDW respectively (Dhangdhariya *et al.*, 2015; Han *et al.*, 2014).

Characterization of PHB

NMR of PHB from *Bacillus* sp. N-2

The ¹H NMR of PHB from *Bacillus* sp. N-2 was carried out as shown in figure 2. The resonance of PHB at 1.6 ppm corresponds to the methyl group (CH₃), while the signal at 2.4 ppm is attributed to a methylene group (CH₂). The signal at 5.2 ppm is

indicated to methane group (CH). Consequently, the obtained results are consistent with the previous report (Han *et al.*, 2014).

FT-IR of PHB from *Bacillus* sp. N-2

Figure 3 demonstrates FT-IR spectrum of the purified PHB. The peak at 1721 cm⁻¹ indicates to ester carbonyl group of PHB. Moreover, the peak at 1280 cm⁻¹ is due to the -CH group in the biopolymer (Bayari and Severcan, 2005). The peak at 2935 cm⁻¹ corresponds to the stretching and deformation vibrations of the O-H groups. The presence of the peak at 2983 cm⁻¹ may be due to the C-H...O hydrogen bond. The specific peak at wave numbers 3442 cm⁻¹ attributes to the terminal O-H bonding or water adsorption on the PHB (López *et al.*, 2012). The FT-IR spectrum is very closer to the FT-IR spectra of PHB extracted from *Bacillus shackletonii* K5, *B. megaterium* and commercial PHB (Liu *et al.*, 2014; Dhangdhariya *et al.*, 2015).

TGA analysis of PHB from *Bacillus* sp. N-2

TGA of purified biopolymer from *Bacillus* sp. N-2 strain was performed to detect the thermal stability of PHB. Figure 4 shows that degradation of PHB was done in three stages and the melting point of the polymer was started at 134°C, but only 0.8% of the total mass was lost. At 237°C, The polymer was lost about 1.6% of total mass in second stage. It should be noted that PHB was almost degraded at 275°C, with maximized degradation occurring at 320°C and in this stage about 96% of polymer was lost. Dhangdhariya *et al.*, (2015) reported that the PHA was degraded in two stages and completely degrade at 300°C, while our results revealed that the PHB was decomposed in three stages and resist till 320°C. On the other hand, produced PHB from *Bacillus shackletonii* K5 was decomposed completely in two stages and reach to 280°C as maximum temperature (Liu *et al.*, 2014).

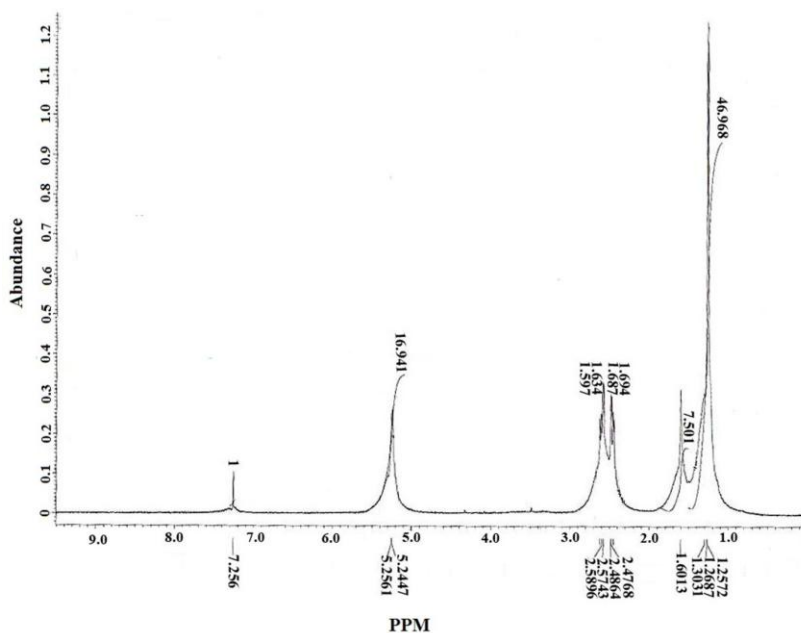


Fig. 2: ¹H NMR signals of purified PHB produced by *Bacillus* sp. N-2 using glucose as carbon source.

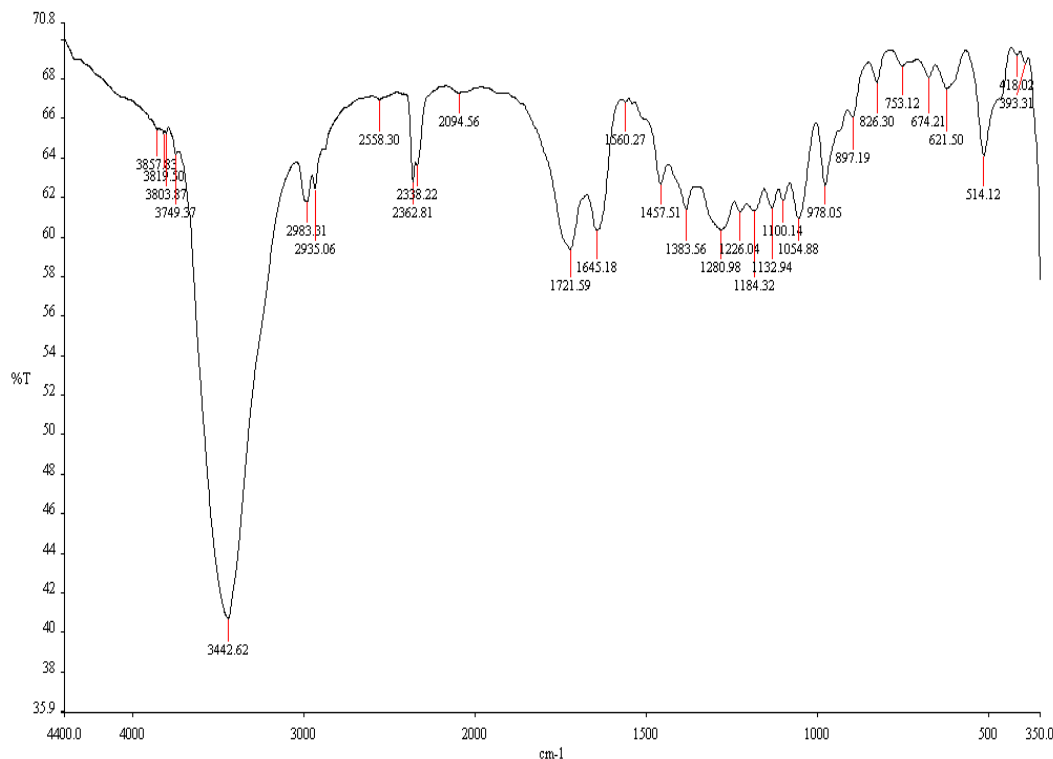


Fig. 3: FT-IR spectrum of purified PHB produced by *Bacillus* sp. N-2 using glucose as carbon source.

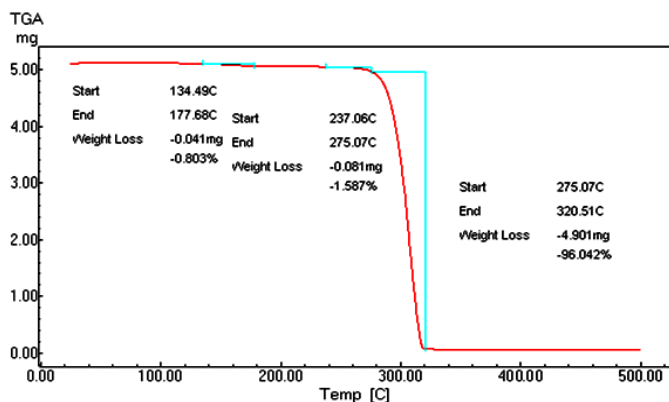


Fig. 4: TGA analysis of purified PHB produced by *Bacillus* sp. N-2 using glucose as carbon source.

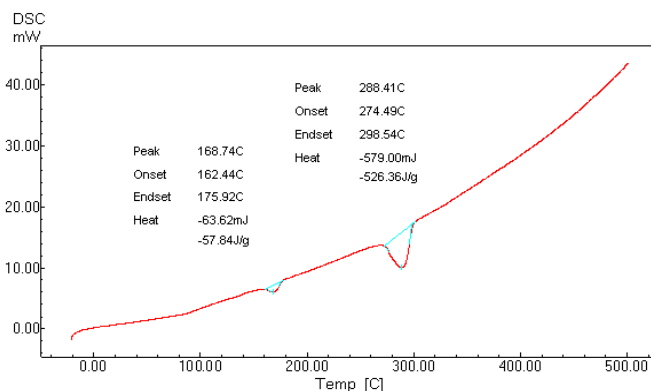


Fig. 5: DSC analysis of purified PHB produced by *Bacillus* sp. N-2 using glucose as carbon source.

DSC analysis of PHB from *Bacillus* sp. N-2

DSC analysis was conducted to investigate the melting temperature, glass transition temperature and heat associated in melting of PHB. The T_m of PHB was 175.9°C and the heat associated with melting was 57.84 J/g for the first peak. The second peak showed at 298°C in correlated with heat about 526.3 J/g (figure 5). These results are proved by the previous literature (Vizcaino-Caston *et al.*, 2015).

CONCLUSION

In the current study, different bacterial isolates were obtained from Egypt and screened for PHB production using glucose as carbon source and Nile red as detector. A new bacterial

isolate (*Bacillus* sp. N-2) was selected, and it was successfully characterized and identified using morphological and biochemical characters. However, it had the ability to produce PHB with amount about 20% of cell dry weight. The produced PHB was characterized using ^1H NMR, FTIR, TGA and DSC. In addition, the results demonstrated that the biopolymer is PHB and highly purified. The future approaches will be concerned with optimization of culture conditions and utilization of houses and industry wastes as a carbon source for PHB production using *Bacillus* sp. N-2.

CONFLICT OF INTEREST

The authors have no conflict of interest to declare.

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